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## PhD Thesis in

Rational design, synthesis and in vitro evaluation of cell-active chemical probes for bromodomain-containing proteins

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#### Abstract

Bromodomains (BRDs) are epigenetic readers able to selectively recognize the acetylated lysine residues on histone and non-histone proteins. Through their activity, bromodomaincontaining proteins (BRDs) are involved in a wide range of cellular events, such as chromatin remodeling and transcriptional activation. One of the most studied and druggable family of bromodomain-containing proteins is the Bromo and Extra Terminal domain (BET) family, whose members (BRD2, BRD3, BRD4, and BRDT) contain two highly homologous bromodomains: BD1 and BD2. Despite several ligands have been discovered, there is still need to identify novel classes of compounds with high potency, selectivity and in vivo activity.

This Ph. D. project is focused on the design, synthesis, biochemical and biophysical evaluation of new chemical probes for BET proteins. To this purpose, three different medicinal approaches were applied to obtain different classes of compounds. Exploiting the bisubstrate approach, bivalent ligands were designed and synthesized. Biochemical and biophysical assays allowed the identification of compound $\mathbf{3}$ (EML896), a promising bivalent compound able to bind both BD1 and BD2 bromodomain of BET proteins. Applying a frozen analogue approach, the diazobenzene core of reported diazobenzene-ligands was rigidized, yielding a benzimidazole scaffold. A small library of benzimidazole-based ligands has been designed and synthesized and compound 15 (EML765) was identified as promising BD1 selective ligand. Finally, at the University of Dundee, the attention was focused on compounds able to induce protein degradation. Specifically, proteolysis targeting chimera compounds (PROTACs) containing BD 1 and BD 2 selective warhead were designed, synthesized and biological evaluated.


## CHAPTER I

INTRODUCTION

### 1.1 Epigenetic

Post-translational modifications (PTMs) on histones are key mechanisms to regulate the expression of specific genes inside the cells. Despite the identical genetic material, cells retain distinct characteristics and biological functions in the tissues and organs in which they are expressed. Epigenetics is the study of reversible and heritable changes which modulate genes activity without altering the underlying DNA sequences. ${ }^{1}$

To minimize the space occupied in cells, DNA is organized in a multi-protein complex called chromatin. The basic element of chromatin is the nucleosome, a structure of DNA wrapped around an octamer of histone proteins composed of a tetramer, containing two copies of both H3 and H4, combined with two H2A/H2B dimers. Histones are highly conserved basic proteins, enrich of lysine and arginine both of which have a positive charge, which helps them to bind the negatively charged phosphate backbone of DNA. The differential condensation state of chromatin regulates the accessibility of transcriptional machinery to genes for their initially transcription into mRNA that finally gets translated into proteins. ${ }^{2}$

There are two major states of chromatin. Heterochromatin is densely packed and transcriptionally silent. On the other hand, euchromatin is less condensed, more accessible, and therefore transcriptionally active.

The conformation of chromatin is dynamically modified through epigenetic modifications such as methylation, acetylation, phosphorylation, sumoylation and ubiquitination. These posttranslational modifications (PTMs) on histone and non-histone proteins act as epigenetic marks that regulate the expression of specific genes inside the cells and are, consequently, able to regulate different processes such as differentiation, development, proliferation and genome integrity.

Over the past years, different proteins have been identified and characterized as epigenetic actors (Figure 1.1). Writers are enzymes able to catalyze the insertion of chemical modifications
into either histone tails or the DNA. Epigenetic erasers are enzymes able to remove the added modifications. Epigenetic readers are domains able to selectively recognize the inserted epigenetic marks. ${ }^{3}$


Figure 1.1 Representation of the proteins which insert (writers), remove (erasers) and read (readers) the epigenetic modifications. ${ }^{3}$

### 1.2 Lysine acetylation: an important epigenetic mark

One of the main epigenetic mechanisms in the regulation of biological processes is the protein acetylation, which occur on lysine residues of histone proteins. Three classes of epigenetic proteins act in this process: 'writers' or lysine acetyltransferases (KATs), which insert acetyl groups to proteins; 'erasers' or lysine deacetylases (KDACs), which remove acetyl groups; and 'readers' or acetyllysine binders, which selectively recognize the acetylated proteins.

### 1.2.1 Lysine Acetyltransferases (KATs)

Acetylation is a dynamic process, mediated by acetyltransferases (KATs), that involves the transfer of acetyl group to $\varepsilon$-amino group of lysine residues from acetyl-coenzyme A (Ac-CoA), which acts as a cofactor. ${ }^{4}$ Traditionally, KATs are classified into two major classes: type A and type B. To the type B class belong the cytoplasmic acetyltransferases, which add the acetyl mark on newly synthesized histone H 4 at K 5 and K 12 (as well as certain sites within H 3 ), driving their translocation into the nucleus and their correct deposition into chromatin. On the other hand, the type A KATs are a heterogeneous family of enzymes, located in the nucleus, which act as activators to enhance transcription. ${ }^{5-6}$ In fact, the acetylation neutralizes the positive charge of lysine residues of histone proteins, decreasing their binding to the negatively charged phosphate of DNA. In this way, the chromatin adopts a more open and less condensed conformation, accessible to the transcriptional machinery. Moreover, the acetyl group acts as a mark, specifically recognized by acetyllysine readers, forming a new surface for protein association and cellular signaling transduction (Figure 1.2). ${ }^{7}$


Figure 1.2 Schematic representation of acetylation process and the impact on different biological function of different proteins. ${ }^{7}$

Based on structure similarities in the binding pocket, the nuclear KAT are properly divided into five major families: the MOZ, Ybf2, SAS2, and Tip60 (MYST) family; the general control
non-repressible 5 (GCN5)-related N -acetyltransferase (GNAT) family; the general transcription factor HATs containing the TAF250 domain; the CREB-binding protein (CBP) and the E1A-associated protein of 300 kDa (p300) family; and the steroid/nuclear receptor coactivators (SRC/NCoA) family. ${ }^{6}$

Changes in the activity and/or expression levels of KATs, are related to different diseases states. Several studies suggest that CBP has a critical role in the regular development of the hematopoietic system whereas p300 has an important tumor-suppressor role. Different type of cancers such as lung, colon, breast and ovarian carcinomas are related to mutations and/or deletions of p300 and/or CBP. Chromosomal translocation occurring on p300/CBP genes are associated with leukemia and lymphomas. ${ }^{7-9}$ Also the GNAT and MYST family members are closely linked to the hallmarks of cancer. Outside oncology indications, several data support a role for histone acetylation in diverse neurological disorders. It has been extensively reported that the acetylation balance is important in memory storage and neuronal plasticity. Some KAT or KDAC are highly expressed in brain areas involved in learning and memory, such as the hippocampus and prefrontal cortex. Several reports shown that neurological disorders such as Parkinson's, Huntington's and Alzheimer's diseases are strictly connected to down-regulation or loss of function of p300/CBP. ${ }^{10-11}$

Interestingly, knockout or chemical inhibition of a GNAT family member, the p300/CBPassociated factor (PCAF) improve cognitive and behavioral deficits in model of Alzheimer's disease. ${ }^{12}$ Mutations of CBP gene result in Rubinstein-Taybi syndrome (RTS), characterized by a short stature, intellectual disability, distinctive facial features, and broad thumbs. ${ }^{8}$ High p300 levels induce hypertrophy in cardiomyocytes and lead to a greater accumulation of fatty acids, insulin resistance and inflammation of the liver, with an important role in cardiovascular diseases and diabetes. ${ }^{13}$

### 1.2.2 Lysine deacetylases (KDACs)

The acetyl group is specifically removed from lysine residues in histones proteins by lysine deacetylases (KDACs) or histone deacetylases (HDACs), epigenetic erasers. Deacetylation restores the positive charge on lysines that is related with the condensed and transcriptionally inactive state of chromatin.

HDACs are divided into five classes based on sequence similarities (Figure 1.3). Class I includes HDAC1, HDAC2, HDAC3 and HDAC8; class IIa consists of HDAC4, HDAC5, HDAC7 and HDAC9; class IIb comprises HDAC6 and HDAC10; class III comprises the sirtuins from SIRT1 to SIRT7; HDAC11 is the only member of class IV. A zinc ion is required for catalysis of enzymes from classes I, II and IV. On the other hand, sirtuins required NAD+ as cofactor for enzyme activity and own deacetylase and ADPribosylase activity (Figure 1.3). ${ }^{14}$


Figure 1.3 Schematic representation of deacetylases based on: number of aminoacids (on the right of each proteins); cellular localizations (Nuc, nuclear; cyt, cytoplasmic; Mito, mitochrondial); deacetylation mechanism (on the lower-right). ${ }^{14}$

As well as acetyltransferases, HDACs control the equilibrium of acetylation state and their aberrant activity has been associated with several pathological conditions such as cancer,
neurological diseases, metabolic disorders, inflammatory diseases, cardiac diseases, and pulmonary diseases. ${ }^{15}$ It is important to mention four HDAC inhibitors approved by FDA for different oncological indications. In 2006, Vorinostat was approved for the treatment of cutaneous manifestations in patients with cutaneous T cell lymphoma. Three years later, Romidepsin was firstly approved for cutaneous T-cell lymphoma and then for peripheral T-cell lymphomas. For the same lymphomas, Belinostat was approved in 2014. One years later, Panobinostat was approved for the treatment of multiple myeloma. ${ }^{16}$

### 1.2.3 Acetyllysine readers: bromodomain

The acetyl group on lysine residues of histone proteins is selectively recognized by epigenetic readers, called bromodomain (BRD). First identified in the brahma gene in Drosophila melanogaster, hence the name, bromodomains are protein-protein interactions modules of approximately 110 aminoacids.

The human proteome encodes 61 bromodomains, which are present in 46 human nuclear and cytoplasmic proteins. These include histone methyltransferases such as ASH1L and mixed lineage leukemia protein (MLL), bromodomain-containing protein (BRD9), HATs (p300/CBP associated-factor and GCN5L2), transcriptional co-activators (TRIM), ATP-dependent chromatin remodeling complexes (bromodomain adjacent to zinc finger domain protein 1B (BAZ1B)), nuclear-scaffolding proteins (polybromo 1 (PBRM1/PB1), helicases (SWI/SNFrelated matrix-associated actin-dependent regulators of chromatin subfamily A (SMARCAs)) and the bromo and extra-terminal domain (BET) family. Based on sequence similarity, the BRD are divided into eight structural classes (Figure 1.4). ${ }^{17-18}$


Figure 1.4 Sequence similarity-based phylogenetic tree of the 61 human bromodomains, divided into eight structural classes. ${ }^{18}$

Despite the unique architecture of each protein, bromodomains have a conserved binding pocket that comprises four $\alpha$-helices (named $\alpha \mathrm{Z}, \alpha \mathrm{A}, \alpha \mathrm{B}$ and $\alpha \mathrm{C}$ ) linked to form a large loop (the ZA loop) and a shorter loop (the BC loop). Different composition of the ZA and the BC loop allow the binding of different BRDs to distinct lysine acetylation sites. ${ }^{19}$

The ability of human BRDs to selectively recognize acetylated peptides was shown by Filippakopoulos et al in 2012. According to other studies, the binding of BRDs to acetylated histone peptides in vitro is relatively weak, suggesting that the high affinity binding in vivo is associated to additional interaction domains. ${ }^{20-21}$

The orientation of the bound peptide can dramatically change with different domains adjacent to the BRD module. Despite that, the acetyl group is recognized in a conserved manner, involving two hydrogen bond with the carbonyl oxygen of acetyl-lysine (KAc): a direct one, with the side-chain amino group of a conserved asparagine residue in the BC loop and a watermediated hydrogen bond with a tyrosine residue in the ZA loop. ${ }^{22}$

BRD-containing proteins are involved in the regulation of gene expression through the modulation of transcription and have a broad range of roles in cellular homeostasis. ${ }^{23}$

The implication of aberrant functions of BRD-containing proteins in several human diseases stimulated the development of BRD inhibitors and, over the past few years, the identification of several chemical probes allowed to elucidate the biological functions of these proteins.

The most important BRD-containing proteins and their ligands are summarized in the next Sections.

### 1.3 Bromodomain-containing proteins

### 1.3.1 HATs containing a bromodomain: p300/CBP and PCAF

The acetyltransferases CBP and p300 (Section 1.2.1) contain a bromodomain flanking the HAT catalytic domain that is important in binding of $\mathrm{CBP} / \mathrm{p} 300$ to chromatin and in induction of histone acetylation at specific sites. ${ }^{24,25}$ Recently, different ligands specific for the bromodomains of CBP and p300 have been reported (Figure 1.5). SGC-CBP30, the first compound reported to bind CBP/p300 bromodomain with high potency ( $\mathrm{K}_{\mathrm{D}}$ values of 21 and 32 nM for CBP and p300 BRDs, respectively) and good selectivity over other bromodomain proteins, was discovered in 2014 by Structure Genomic Consortium (SGC). CBP30 reduce immune cell production of several pro-inflammatory cytokines with further therapeutic applications not only in an oncology context but also in the treatment of autoimmune disorders. ${ }^{26-27}$


SGC-CBP30


I-CBP112


GNE-272


GNE-781


CCS1477

Figure 1.5 Selected ligands of p300/CBP bromodomain.

The same group developed a benzoxazepine-based compound (I-CBP112) less potent than CBP-30 ( $\mathrm{K}_{\mathrm{D}}$ values of 0.142 and $0.625 \mu \mathrm{M}$ for CBP and p 300 , respectively) but with a better selectivity toward BET family members ( $\mathrm{IC}_{50}$ of $15 \mu \mathrm{M}$ for BRD 4 (BD1) compared to $3.2 \mu \mathrm{M}$ for BRD4 (BD1) of CBP-30). It has been shown that a synergistic combination of I-CBP112 and A-485, a CBP/p300 acetyltransferase inhibitor, can decrease the transcription of androgendependent and pro-oncogenic prostate genes, reducing cell proliferation in prostate cancer. ${ }^{28}$

In 2017, GNE-272 was identified as promising pyrazole-based compound and further SAR studies allowed the identification of GNE-781 with high potency ( $\mathrm{IC}_{50}$ of 6.2 nM ), selectivity over other BRD-containing proteins and anti-tumor activity in AML tumor model. ${ }^{29-30}$ CellCentric has initiated a phase I/IIa clinical trial (NCT03568656) for the CCS1477 compound in patients with metastatic prostate cancer and other solid tumors. This compound is a potent ( $\mathrm{K}_{\mathrm{D}}$ values of 1.7 and 1.3 nM for CBP and p300 BRDs respectively), selective (approximately 170 fold over BET members) and orally bioavailable inhibitor of the bromodomain of p300 and CBP. ${ }^{31}$

PCAF is another multi-domain protein containing a histone acetyltransferase unit and a single bromodomain.

During transcription, this protein associates CBP and p300 and several type of diseases such as cancer, neuroinflammation and HIV infection are related to its misregulation. ${ }^{32-33}$

### 1.3.2 Bromodomain-containing protein 9 (BRD9)

Bromodomain-containing protein 9 and 7 (BRD9 and BRD7) are high homology components of the chromatin remodelling SWI/SNF (BAF) complex. Also these proteins are disease related (e.g. pediatric malignant rhabdoid tumors) and small molecules inhibitors are available (Figure 1.6). ${ }^{34}$ The first identified BRD7/BRD9 ligand was LP99, which helped to better understand the role of these proteins in the regulation of inflammatory cytokines. ${ }^{35}$

In 2016, GlaxoSmithKline scientists identified the first selective chemical probes for BRD9 (I-BRD9) with excellent selectivity over other BRD-containing proteins and a surprisingly selectivity over BRD7 (approximately 200 fold), despite the $85 \%$ sequence homology. Treatment with this ligand lead to the down-regulation of sensitive genes implicated in cancer and immunology, suggesting a role of BRD9 in this pathways. ${ }^{36-37}$ The SGC group identified two potent, selective and cell-permeable chemical probes, BI-7273 ( $\mathrm{IC}_{50}$ of 19 nM and 117 nM for BRD9 and BRD7 respectively) and BI-9564 ( $\mathrm{IC}_{50}$ of 75 nM and $3.4 \mu \mathrm{M}$ for BRD9 and BRD7, respectively). ${ }^{38}$





Figure 1.6 Selected ligands of BRD9/BRD7 bromodomain.

### 1.3.3 BAZ2A/2B, TAF1 and PB1 bromodomain

Other bromodomain-containing proteins are the bromo adjacent to zinc finger 2 A (BAZ2A) and 2B (BAZ2B), component of the nucleolar remodeling complex (NoRC) which play a role in the regulation of noncoding RNAs. Through their bromodomain, these proteins interact with the acetylated histone tails. High expression levels of BAZ2A have been reported in prostate cancer whereas poor outcome of pediatric B cell acute lymphoblastic leukemia is associated with high expression levels of BAZ2B. One representative BAZ ligand is GSK2801 depicted in Figure 1.7. This compound has a nanomolar potency ( $\mathrm{K}_{\mathrm{D}}$ of 136 and 257 nM for BAZ2B and BAZ2A, respectively), selectivity over other BRDs and a good cell-permeability, showing a synergistic activity with BET inhibitors to induces apoptosis in triple-negative breast cancer. ${ }^{39}$


GSK2801
Figure 1.7 BAZ2A/BAZ2B bromodomain ligand.

TFIID, a transcription complex, contain a core subunit TAF1, which contains a tandem BRD module that recognizes multiply acetylated H 4 peptides.

A subunit of the PBAF chromatin remodeling complex, PB1 interact with different acetylated lysine through its six tandem bromodomain and its mutations are often found in clear cell renal cell carcinoma. ${ }^{19}$

### 1.4 BET proteins: structure and physio-pathological roles

One of the most important and druggable family of BRD-containing proteins is the BET family (Bromodomain and Extra-Terminal Domain) composed by four proteins: bromodomaincontaining protein 2 (BRD2), BRD3, BRD4 and the bromodomain testis-specific protein
(BRDT). With the only exception of BRDT, a tissue-specific isoform expressed in male germ cells, these proteins are ubiquitously expressed in the cell nucleus.

All the BET members share a common domain architecture featuring two evolutionarily highly conserved N -terminal bromodomains (BD1 and BD2) and an extra-terminal domain (ET). The active acetyl-lysine binding pocket has the classical architecture of bromodomain modules. Characteristic of all the BET family members is the "WPF shelf" (W81, P82, F83), a hydrophobic region of the BC loop that includes conserved Trp/Pro/Phe motif (Figure 1.8). ${ }^{40}$


Figure 1.8 General structure of BET proteins considering BRD4 as representative of the family.
Highlighted the ZA channel, the acetyl-lysine binding pocket and the WPF shelf. ${ }^{41}$

BET proteins play essential roles in different biological processes such as epigenetic memory, regulation of cell-cycle progression through gene transcription, and maintenance of chromatin architecture.

Over the past few years, the development of BET chemical probes allowed to better characterize their biological functions and their potential as therapeutic targets. However, the functional differences among the four BET members remain not fully elucidated. ${ }^{42}$

Bromodomain-containing protein 4 (BRD4) is probably the most studied member of BET family. Through its two BRDs, BRD4 interacts with di-acetylated histone H3 and/or H4. There are three isoforms expressed in humans: one long isoform of 1362 residues (BRD4 A) which
also possess a C-terminal domain (CTD), and two shorter isoforms (BRD4 B and C) that differ by a unique 72-residue at the C terminus (Figure 1.9). ${ }^{43}$


Figure 1.9 Schematic representation of the long isoform (BRD4 A) and the two short isoforms (BRDA B and C) of BRD4. ET (Extra-terminal domain); CTD (C-terminal domain). ${ }^{44}$

Important is the role of BRD4 in cell cycle control, affecting cell proliferation, apoptosis, transcription and other cellular processes. The C-terminal domain (CTD) and the second bromodomain of BRD4 interacts with cyclin T and CDK9, components of the positive transcription elongation factor complex (P-TEFb). Through its phosphorylation activity, this complex promotes the activation of RNA polymerase II and, consequently, transcriptional elongation. The second bromodomain (BD2) recognizes an acetylated region of cyclin T , preventing the association of $\mathrm{P}-\mathrm{TEFb}$ with 7SK/HEXIM, a ribonucleoprotein complex that maintain $\mathrm{P}-\mathrm{TEFb}$ in a kinase-inactive state. ${ }^{45}$ BRD4 marks the start sites of many M/G1 genes, promotes cell-cycle progression to S phase and seems important for the G 2 transition to M phase. ${ }^{45-46}$

Bromodomain-containing protein 2 (BRD2) recognize mainly the acetylated lysine 12 (K12Ac) of H 4 and activates transcription (Figure 1.10). Important is the association of BRD2 with E2F, a transcription factors which play a key role during the G1/S transition, regulating the expression of several genes. ${ }^{45,47}$


Figure 1.10 Schematic representation of BRD2, BRD3 and BRDT domain. ${ }^{44}$

One of the main function of bromodomain-containing protein 3 (BRD3) is the binding with an acetylated lysine adjacent the zinc finger domain of GATA1, a transcription factor with an essential role in hematopoiesis (Figure 1.10). Mutation in the first bromodomain of BRD3 or interference in its interaction resulted in a lower ability of GATA1 to associate chromatin, inhibiting erythroid maturation. ${ }^{48}$

BRDT binds to di-acetylated H 4 and mediates the chromatin remodeling during spermatogenesis (Figure 1.10). It has been shown that this bromodomain testis-specific protein is essential for normal spermatogenesis, and mutation in its first bromodomain results in abnormal spermatids and sterility in mice. ${ }^{45}$

The roles of BET proteins in transcriptional regulation connect these proteins to several disease states. Mutation or overexpression of BET proteins are often found in different type of cancers. Fusion of NUT (nuclear protein in testis) with BRD4 or BRD3 are found in the NUT midline carcinoma (NMC), an aggressive and poorly differentiated carcinoma that originates mainly from midline sites such as the head, neck or mediastinum. BET proteins can directly regulate the expression of MYC, a well-characterized oncogene often mutated or overexpressed in several type of malignancies, with important implications in hematologic cancer models, such as MLL-fusion leukemia, acute myeloid leukemia (AML), Burkitt's lymphoma, multiple myeloma, and B-cell acute lymphoblastic (BLL) leukemia. Different brain tumor MYC dependent or independent such as medulloblastoma, diffuse intrinsic pontine glioma (DIPG) and glioblastoma multiform (GBM) are sensitive to BET inhibition. ${ }^{46,49}$

Important is the role of BET proteins in inflammation, metabolic disorders and adipogenesis. The implication of BET inhibition in inflammatory and autoimmune disorders is due to their ability to interfere with pro-inflammatory transcription factor such as NF-kB, STAT3, STAT5 and RORC. ${ }^{50}$ The reduced secretion of cytokines (TNF $\alpha$, IL-1b, IL-6 and IL-8) and matrix metalloproteinase (MMP-1, MMP-3 and MMP-13) can be helpful in patients with rheumatoid arthritis, osteoarthritis and psoriasis. ${ }^{17}$

BET proteins can drive the expression of $\operatorname{PPAR} \gamma$, a transcription factor important in the adipocyte differentiation and their inhibition may have important implications in metabolic diseases related to adipose tissue dysfunction and obesity. ${ }^{51}$

BET inhibition is also investigated in the context of cardiovascular disease and atherosclerosis. At cardiac level, BRD4 expression is induced during cardiac hypertrophy. Moreover, the correlation with NF-kB and GATA4, known to direct heart failure progress, reinforce the implications of these proteins to cardiac disorders. ${ }^{17}$

### 1.5 BET ligands

Over the past few years, the high therapeutic implications of BET proteins prompted a growing interest in developing potent, selective and cell-active BET ligands. Typically, a BET ligand is a small molecule able to insert within the hydrophobic pocket between the ZA and BC loops displacing an acetylated peptide. From the structural point of view, the ligands feature an acetyl-lysine mimetic group, a portion able to form a hydrogen bond to the conserved asparagine and tyrosine residues, thus mimicking the interactions of the lysine-acetylated peptide. ${ }^{18,41}$

Different chemotypes have been identified, such as methyltriazolodiazepines and triazepines, 3,5-dimethylisoxazoles, dihydroquinazolinones, tetrahydroquinolines and 2thiazolidinones.

### 1.5.1 Diazepine-based and triazepine-based compounds as BET ligands

One of the first reported ligands of BET proteins with high-affinity and selectivity share the triazolodiazepine chemical scaffold. The finding that clinically approved benzodiazepines, such as alprazolam and midazolam, bind BET bromodomains with low affinity prompted the researchers to use this versatile scaffold as template for the development of BET selective
ligands. At the same time, a patent published in 2009 by Mitsubishi Pharmaceuticals details the binding of triazolothienodiazepine-based compounds to BET proteins. These evidence drive the design of several triazolodiazepine-based compounds as BET ligands, which feature a thienodiazepine or a benzodiazepine core and a triazole as KAc mimetic group (Figure 1.11). The interactions of these compounds within the binding pocket are very similar. The two adjacent nitrogen atoms of the 1,2,4-triazole form hydrogen bonds with the conserved asparagine and tyrosine residues, whereas the methyl group binds to a small hydrophobic pocket, mimicking the interaction of the native acetyl-lysine ligand. The fused aromatic ring of the triazolodiazepine bind within a lipophilic pocket, the ZA channel, while the 5-phenyl group interacts with the WPF shelf, a critical region for good binding, affinity and selectivity for BET proteins. ${ }^{47}$

(+)-JQ1


CPI-203



RO6870810


MS417


I-BET762

Figure 1.11 Triazolothienodiazepine and benzodiazepine BET ligands. In red, the KAc mimetic group.

One of the first published triazolothienodiazepine-based BET ligand was (+)-JQ1, developed by Filippakopoulos group. ${ }^{52}$ This compound shown a highly selectivity toward BET family over non-BET bromodomain-containing proteins with affinity for each BET proteins ranged
from 49 nM to 190 nM . Only the S enantiomer, (+)-JQ1 was active, whereas the R enantiomer (-)-JQ1 showed no significant interaction ( $\mathrm{IC}_{50}$ of 10.000 nM ). JQ1 displayed antiproliferative effects in preclinical models of several tumor types including multiple myeloma, lymphoma, acute lymphoblastic leukemia, medulloblastoma, NMC (NUT midline carcinoma), mostly due to modulation of MYC expression. ${ }^{46}$ In a murine model of human interstitial fibrosis, it has been shown that the myocardial expression of classical hypertrophic marker genes is reduced with the treatment of JQ1, preventing the development of left ventricular hypertrophy, and systolic dysfunction. ${ }^{17}$

This compound was intensively used for deepen the biological function of BET proteins but is not enter in clinical trials for its short half-life. ${ }^{53}$

A structurally JQ1-related compound with a longer half-life is RO6870810. This compound enters in Phase I, multicenter, open-label study for patients with advanced solid tumors. Subcutaneous dosing is generally well tolerated, with good results in NUT-midline carcinoma (NMC) treatment associated with reversible irritation of the injection site, increases of bilirubin and anorexia. ${ }^{54}$

The replacement of tert-butyl ester of JQ1 with a methyl ester, led the formation of MS417, a triazolothienodiazepine-based compound, able to decrease the transcriptional activity of NF$\kappa$ B in HIV-associated nephropathy. Other triazolothienodiazepine compounds related to JQ1 are CPI-203, with superior oral and intraperitoneal bioavailability and OTX-015 with high efficacy in different cancer cell lines. ${ }^{18}$ The latter has completed a Phase I trial in patients with acute myeloid leukemia (AML) and solid tumors. This compound is orally available but has different problems related to dose-limiting effects such as thrombocytopenia and gastrointestinal events in patients with or without leukemia. ${ }^{55}$ The effects of OTX-015 were also investigated in brain tumors in a nonrandomized, multicenter Phase IIa trial but the trial was terminated after one year. ${ }^{53}$

An important triazolobenzodiazepine is I-BET762 with high potency (IC so $_{0}$ of $32.5-42.5$ $\mathrm{nM})$ and selectivity for BET proteins. ${ }^{56-57}$ In vivo studies suggested that I-BET762 can reduce MYC expression in different type of cancers. Moreover, this compound is able to suppress proinflammatory gene expression in LPS-stimulated macrophages such as pro-inflammatory cytokines (e.g., IFNB1, IL1B, IL6 and IL12A) and chemokines (e.g., CXCL9 and CCL12). In 2012, the compound was entered in Phase I of clinical trials for the treatment of NUT midline carcinoma, and other refractory hematologic malignancies. ${ }^{54}$

The bioisosteric replacement of the chiral carbon in the diazepine scaffold with a nitrogen atom furnished triazolobenzotriazepine-based compounds. The most promising compound (BzT-7) has a good BRD4 binding affinity ( $\mathrm{K}_{\mathrm{D}}$ values of $0.64 \mu \mathrm{M}$ ) and a similar binding mode of benzodiazepine-related compounds (Figure 1.12). ${ }^{58}$ The chlorinated analogue (Cl-BzT-7) showed a similar potency of BzT-7 and a slightly better activity toward the first bromodomain of BET (Panagis Filippakopoulos, unpublished data).


BzT-7


Cl-BzT-7

Figure 1.12 Triazolobenzotriazepine BET ligands. In red, the KAc mimetic group.

### 1.5.2 3,5-dimethylisoxazole-based compounds

3,5-dimethylisoxazole moiety has been shown to be an excellent KAc mimetic group. In fact, the isoxazole oxygen atom is able to mimic a hydrogen bond with the amide of asparagine residue, while the water-mediated hydrogen bond with tyrosine is formed by isoxazole nitrogen. Starting from this fragment, GSK scientist identified I-BET151, a 3,5-dimethylisoxazole-based compound (Figure 1.13). ${ }^{59}$


Figure 1.13 Development of 3,5-dimethylisoxazole BET ligand (I-BET151) starting from 3,5dimethylisoxazole fragment. In red, the KAc mimetic group.

This compound revealed high selectivity for BET proteins and high potency $\left(B R D 4 K_{D}=\right.$ 100 nM ). This ligand is able to modulate transcriptional programs responsible for cell-cycle progression and apoptosis in MLL-fusion-driven leukemia cells. ${ }^{59}$


ISOX I R = -cyclopropyl
ISOX II R = -cyclopentyl


OXF-BD02


OXF-BD03


OXF-BD04

Figure 1.14 Exemplifies other 3,5-dimethylisoxazole-based BET ligands. In red, the KAc mimetic group.

From the same fragment, other 3,5-dimethylisoxazole-based BET ligands were developed (Figure 1.14). All the compounds share an aromatic ring in position 4 of isoxazole ring. On this, N-cyclopropyl or N-cyclopentyl sulfonamide is inserted to occupy the WPF shelf and improve BET selectivity (ISOX I and ISOX II). ${ }^{60-61}$ On the other hand, the substitution of the aromatic ring (a phenolic or phenyl acetate group) with a phenyl methanol group afforded two different compounds, OXFBD02 and OXFBD03 with a good potency ( $\mathrm{IC}_{50}$ values of 384 nM and 371 nM for BRD4 (BD1), respectively) but with a short metabolic half-life (approximately 40 min for OXFBD02) in human liver microsomes. The replacement of the phenyl methanol group
with a 3-pyridinemethanol furnished the related compound OXFBD04 with higher affinity and an increased metabolic half-life (approximately 6.5 h ) in human liver microsomes. ${ }^{62-63}$

### 1.5.3 Dihydroquinazolinones and tetrahydroquinolines

Two different fragments were identified as KAc mimetic group by different researchers in a fragment-screening campaign: the 3,4-dihydro-3-methyl-2(1H)-quinazolinone (MQZ) fragment and the N -acetyl-2-methyltetrahydroquinoline (THQ) fragment (Figure 1.15).




THQ-fragment


I-BET726

Figure 1.15 Development of BET ligands starting from MQZ and THQ fragments. In red, the KAc mimetic group.

Optimization of MQZ fragment with the insertion of different substituents able to insert in the WPF shelf, resulted in PF-1, a potent compound ( $\mathrm{IC}_{50}$ value of 220 nM for BRD4) with high selectivity for the BET bromodomains. Different studies shown that the treatment of sensitive cell lines with PFI-1 caused cell-cycle block and induction of apoptosis. ${ }^{64}$

Decoration of THQ fragment lead to I-BET726, a selective and potent BET ligand ( $\mathrm{K}_{\mathrm{D}}$ of 4 nM ), active in the treatment of neuroblastoma tumors in vivo and with an excellent antiproliferative activity in some solid tumor cell lines. ${ }^{65}$

### 1.5.4 2-Thiazolidinones

Using a fragment-based drug discovery approach followed by docking and X-ray crystallography, Zhao et al. were able to screen a library of fragment, identifying 2-thiazolidinone-based ligands as strong BET binders. Structure-based optimization allowed the identification of some compounds with submicromolar IC $_{50}$ values, with a good potential for further optimization (Figure 1.16). ${ }^{66}$


Figure 1.16 Develop of 2-thiazolidinones ligands starting from an identified fragment. In red, the KAc mimetic group.

### 1.6 Selectivity within BET family

One of the most challenging aspects in the BET field is the design of selective ligands for each BET proteins and also for their corresponding first and second bromodomains. Despite the excellent progress made, all the ligands mentioned above are pan-BET ligands, i.e. compounds with no selectivity within BET family members. ${ }^{67}$

Nevertheless, the first and the second bromodomain of each BET proteins seem to have different functions. For instance, the first bromodomain of BRD4 binds di-acetylated $\mathrm{H} 4 \mathrm{~K} 5 \mathrm{ac} / \mathrm{K} 8 \mathrm{ac}$, regulating transcriptional activation thorough the attachment of this domain and its associated proteins to target gene promoter. On the other hand, the second bromodomain recruits mostly non-histone proteins such as the pTEFb complex to target genes. In BRD3, only
the first bromodomain binds to the transcription factor GATA1, suggesting a separate role of the second bromodomain of the protein. ${ }^{67-68}$

These evidences suggested that a selective targeting of singular BET and the discrimination between the first and the second bromodomain could results in distinct transcriptional outcomes, mitigating unwanted effects such as acquired-resistance of tumor cells treated with BET inhibitor, in addition to dose-limiting toxicities and limited long-term effectiveness. ${ }^{53,69}$

Unfortunately, the very high sequence identity at the KAc binding pocket (95\%) of BET family members has limited the design of selective ligands and only few selective compounds are known. Three crucial residue positions that differ between the first and the second bromodomain of BRD4 close the KAc peptide-binding site are shown by sequence comparison (Figure 1.17). In particular:

- glutamine (Q85) in BD1 correspond a lysine residue (K378) in BD2;
- aspartic acid in the BC loop of BD1 (D144) is a histidine residue in BD2 (H437);
- isoleucine in BD1 (I146) is a valine residue in BD2 (V439). ${ }^{67}$


Figure 1.17 Differences between BD1 and BD2 showed by superposition of the x-ray structure of BRD4-BD1 (in yellow, PDB 4QB3) and BRD2-BD2 (in blue, PDB 4J1P). ${ }^{67}$

Exploiting these small differences, some selective ligands have been developed (Figure 1.18).
BD1-selective ligands


MS436


MS611



LT052
BD2-selective ligands


RVX-208


ABBV-744


TC-AC 28


GSK340

Figure 1.18 Selected ligands BD1 and BD2 selective. In red, the KAc mimetic group.

From a structural point of view, all these compounds share different KAc mimetic groups and different substituents which clash the different amino acids in $\mathrm{BD} 1 / \mathrm{BD} 2$ domain, driving the selectivity of compounds. For example, the sulfonamide oxygen of MS436 forms a watermediated hydrogen bond with the side-chain amine of a lysine residue (K91) of BRD4 (BD1) mutated in alanine residue (A384) in BRD4 (BD2). These interactions favor a high potency ( $\mathrm{K}_{\mathrm{i}}$ of $30-50 \mathrm{nM}$ for BRD4 (BD1) and a 10-fold selectivity for the first bromodomain. ${ }^{70}$ A related diazobenzene-based compound, MS611 showed 100-fold selectivity for the first bromodomain of BRD4 (Ki of 0.41 nM for BD1 of BRD4 and $41.3 \mu \mathrm{M}$ for BD2 of BRD4). ${ }^{68}$

The X-ray crystal structure of Olinone in complex with BD1 of BRD4 highlighted interactions with the aspartate residue (D144) mutated in histidine (H435) in BD2 of BRD4 which would clash with the indole moiety of compound. Olinone exhibited 100 -fold higher selectivity toward $\mathrm{BD} 1\left(\mathrm{~K}_{\mathrm{D}}\right.$ of $3.4 \mu \mathrm{M}$ ) than BD 2 (no detectable binding) for all of the BET bromodomains. ${ }^{68}$

On the other hand, interactions with the typical histidine residue in BD 2 allowed to gain selectivity toward the second bromodomain.

RVX-208 is packed against the histidine residue (H433), establishing $\pi-\pi$ stacking interactions which limit the di-methyl-phenyl moiety in one fixed conformation that is not conserved in BD1, considering that the histidine is replaced in an aspartate residue. The compound displays high potency (IC $\mathrm{I}_{50}$ values of $87 \mu \mathrm{M}$ and $0.51 \mu \mathrm{M}$ for BD1 and BD2 of BRD3, respectively) and about 170-fold selectivity for the second bromodomain. ${ }^{71-72}$

Exploiting a 'bump and hole' approach, TC-AC-28 was developed. The indole moiety establishes $\pi$ - $\pi$ stacking interactions with the histidine residue in the second bromodomain of BRD2 substituted by an aspartate residue in BD1 which cannot create such interactions. The compound has high affinity and approximately 20 -fold selectivity for BD2 over BD1 ( $\mathrm{K}_{\mathrm{D}}$ of 800 and 40 nM against BD 1 and BD 2 of BRD2, respectively). ${ }^{73}$

Optimization of the tetrahydroquinoline I-BET726 (Section 1.5.3) furnished GSK340 with nanomolar potency and approximately 50 -fold selectivity over BD2 of BRD4 ( $\mathrm{pIC}_{50}$ of 5.5 and 7.2 against BD 1 and BD 2 of BRD4, respectively). ${ }^{74}$

These compounds allowed to better understand different pathways in which the two domains are involved. For instance, the BD1 selective ligand Olinone stimulated the progression of primary oligodendrocyte progenitors, while simultaneously inhibition of both BD1 and BD2 domain prevented it and retained the cells at a progenitor state, with interesting implications in disorders characterized by myelin loss such as aging and neurodegeneration stage. ${ }^{68}$ On the other hand, the BD2 selective ligand, RVX-208 affect the transcription of a lower number of genes toward the pan-BET inhibitor JQ1. ${ }^{71}$ These evidence suggested that the discrimination between BD1 and BD2 could achieve more selective transcriptional effects.

Currently, only BD2-selective ligands reach the clinical phase. RVX-208 is in phase III trial for patients with high-risk cardiovascular disease with type 2 diabetes mellitus and low high-
density lipoprotein (HDL). Surprisingly, this compound is the only BET ligand to reach phase III trial and one of the few BET ligands for non-oncology indications. ${ }^{75}$

ABBV-744, currently the most potent ( $2-5 \mathrm{nM}$ ) and selective BD2 ligand (approximately 250-640-fold) is in phase I of clinical trials with indications for prostate cancer and AML. ${ }^{76-77}$

In 2019, a BD1 selective ligand was developed (LT052) starting from a benzo[cd]indol-2(1H)-one scaffold. This compound showed nanomolar BRD4 BD1 potency and 138-fold selectivity over BRD4 BD2 with good cellular results, exploring new therapeutic approach for the treatment of acute gout arthritis. ${ }^{78}$

### 1.7 Bivalent BET ligands

Beyond the classical medicinal chemistry approaches, alternative strategies can be used to develop more potent and selective chemical probes. The presence in BET proteins of two bromodomains offer the possibility to engage both domains using a bivalent ligand, designed linking two 'warheads' that engage distinct binding sites within proteins.

Bivalent chemical probes have the potential to exhibit a better affinity and selectivity compared to a monovalent probe, which may bind not only a structural domain of a protein of interest, but also many structurally related domains.




Figure 1.19 Bivalent ligands of bromodomain-containing proteins.

To date, few bivalent ligands have been developed (Figure 1.19). Bradner et al. designed bivalent BET ligands connecting two molecules of JQ1 with different linker lengths. ${ }^{79}$ The most promising compound, MT1 has high potency ( $\mathrm{IC}_{50}$ of 3.09 nM toward BRD4 (BD1)) compared to the monovalent ligand JQ1 (IC 50 of 20.9 nM toward BRD4 (BD1)) and has a 400fold improvement in activity in AML than the corresponding monovalent ligand JQ1. The better activity is probably due to the capability of MT1 to induce dimerization of individual BET bromodomains. Interestingly, the high-resolution structure shown that one molecule of MT1, simultaneously recognizing two bromodomains of BRD4(BD2), creates a new hydrophobic pocket between the two bromodomain monomers. ${ }^{79}$ Evaluation of the nature and rigidity of the linker, afforded MS645 a bivalent compound related to MT1, with a shorter and lipophilic linker. Notably, MS645 has an important effect in solid tumor, inhibiting BRD4 interactions with transcription enhancer/mediator proteins MED1 and YY1, required for accelerated proliferation of solid tumor TNBC cells. ${ }^{80}$

Another class of bivalent ligands that induce dimerization are the biBET ligands. The X-ray crystal structure of a biBET ligand in complex with the first bromodomain of BRD4 revealed the formation of a dimer, in which both the triazolopyridazine and N -methylpiperazinone act as KAc mimic group. These compounds cause cell death in BRD4-dependent cell lines three orders of magnitude higher than JQ1. Remarkably, both JQ1 and biBETs inhibit the acute lymphoblastic leukemia cell line RS4-11 but only bivalent ligand reach near-complete cell killing, highlighting possibly beneficial from bivalent inhibition. ${ }^{81,82}$

### 1.8 PROTAC approach

Over the past few years, an emerging strategy to inhibit proteins activity is their selective induced-degradation using PROTACs compounds.

Proteolysis Targeting $\underline{\text { Chimeras (PROTACs) are heterobifunctional molecules able to hijack }}$ the target protein to ubiquitin-proteasome machinery (UPS) for their selective degradation.

Structurally, a PROTAC compound includes a ligand for the protein of interest (POI) and a ligand for the E3 ubiquitin ligase connected by a linker. In this way, these heterobifunctional molecules are able to bridge the target protein and an E3 ubiquitin ligase. The latter binds to an E2 enzyme, promoting the polyubiquitination of the protein. The polyubiquitinated target is recognized as a substrate by the proteasome and is consequently degraded. As the PROTAC remains unmodified, it can initiate a new degradation event (Figure 1.20). ${ }^{83}$


Figure 1.20 Schematic mechanism of PROTACs.

In order to better understand this process, in the following Sections will be detailed the main elements involved in this mechanism.

### 1.8.1 The ubiquitin/proteasome system (UPS)

One of the major mechanism for protein degradation involve the ubiquitin/proteasome system (UPS) which maintain a good balance of intracellular protein levels eliminating damaged, misfolded and mutant proteins. Proteins are covalently conjugated with ubiquitin, a highly conserved 76 -residue protein. In this way, proteins are labelled and targeted into proteasome for their proteolysis into small peptides of 3-24 amino acids.

Protein ubiquitination is an ATP-dependent enzymatic reaction mediated by three enzymes: an ubiquitin-activating enzyme (E1), an ubiquitin-conjugating enzyme (E2), and an ubiquitin ligase (E3) (Figure 1.21)..$^{84-85}$

E1 ubiquitin-activating enzymes mediates the formation of a high-energy thioester bond between the C-terminus of ubiquitin and cysteine residues in E1, in an ATP-dependent manner. Activated ubiquitin is transferred to E2 ubiquitin-conjugating enzymes via transthioesterification reaction. The transfer of ubiquitin from E2 enzyme to Lys residue or the N terminus of the substrate is mediated by an E3 ubiquitin ligase, which selectively recognize the protein substrate. Commonly, the lysine ubiquitinated are the K48- or K11-residues. ${ }^{84-85}$


Figure 1.21 Schematic representation of the ubiquitination process. ${ }^{84}$

### 1.8.2 E3 ligases

The specific recognition of target proteins for their degradation is mediated by E3 ligases, key mediator of UPS system. The high number of these ligases in mammals, approximately $500-1000$, suggested a high specificity and versatility of this process. ${ }^{86}$

Commonly, E3 ligases are divided into four major families:

- the Really Interesting New Gene (RING)-finger-type;
- the HECT (homologous to the E6-AP carboxyl terminus) type, which transfer ubiquitin to the target substrate protein after the transfer to their own cysteine residues;
- the U-box-type, which have a RING-like domain but lacking the cysteine and histidine zinc co-ordination sites;
- PHD-finger type.

The largest family of E3 ligases is the RING-finger-type, which use a zinc finger domain to interact with E2 enzyme for direct ubiquitin transfer to the substrate. One of the most important subfamilies is the cullin-RING ligase complexes (CRLs) built on Cullin proteins (Figure 1.22). The substrate is specifically recognized through a substrate-recognition protein anchor to cullin by an adaptor protein. The binding with the ubiquitin-conjugating enzyme (E2) occurs at the RING component. ${ }^{87}$


Figure 1.22 General representation of the cullin-RING ligase complexes (CRLs). ${ }^{87}$

Considering that alterations in the correct levels of proteins or their quality are related to different type of pathological states, proteasome and E3 ligases have become interesting therapeutic target. The identification of E3 ligase ligands is important also in PROTAC compounds development. Herein, we focus the attention on the most used E3 ligase ligands for BET PROTACs: the cereblon (CRBN) and the Von Hippel-Lindau (VHL) ligand.

### 1.8.2.1 Cereblon (CRBN) as E3 ligase

Cereblon (CRBN) is the substrate-recognition protein of the E3 ubiquitin ligase complex containing Cullin 4 (CUL4) protein. Other member of this complex are the DNA damagebinding protein 1 (DDB1) which connect Cereblon to the cullin protein (CUL4), and Roc1
(regulator of cullin 1) which interact with E2 ubiquitin-conjugating enzyme through its RING finger domain. ${ }^{88}$

In 2010, Handa and co-workers described for the first time the E3 ligase Cereblon as the primary target of thalidomide. Since then, thalidomide analogues, called immunomodulatory drugs (IMiDs), such as lenalidomide, pomalidomide, 4-OH-thalidomide and its chemically and biologically more manageable analogues (the 5-OH-thalidomide and the 5-piperazinethalidomide 105) were developed as CRBN ligands (Figure 1.23). ${ }^{88-90}$

thalidomide

pomalidomide

lenalidomide



5-piperazine-thal (105)

Figure 1.23 Structure of immunomodulatory drugs (IMiDs) as Cereblon binders.

The crystal structure of DDB1-CRBN complex bound to IMiDs allowed to clarify their interaction within the complex. The glutarimide ring of IMiDs is inserted within a hydrophobic tryptophan pocket, termed the thalidomide-binding domain (TBD). The phthalimide ring is exposed on the surface of the CRBN protein, modifying its ability to interact with new substrates. ${ }^{88}$ The C 4 phthalimide aniline position of thalidomide (different from lenalidomide and pomalidomide) is solvent exposed. At the same manner, the phthalimide C5 and C6 positions are fully solvent exposed, allowing modifications in this position to obtain more manageable thalidomide analogues (5-OH-thal and 5-piperazine-thal (159)). The binding of the (S)-enantiomer is preferred over the ( R )-enantiomer but under physiological conditions thalidomide undergoes rapid racemization. ${ }^{91}$

The mechanism of IMiDs is complex and not entirely elucidated but some studies suggested two different effects. On one hand, these compounds compete with endogenous substrates, decreasing their degradation. On the other hand, IMiDs can redirect the ligase to degrade new proteins, changing CRBN's E3 ligase substrate preference. For example, these compounds are able to promote the ubiquitination and degradation of transcription factors as Ikaros (IKZF1) and Aiolos (IKZF3); lenalidomide but not thalidomide and pomalidomide can stimulate the degradation of casein kinase $\mathrm{I} \alpha(\mathrm{CKI} \alpha)$.

Despite the progress made, the knowledge about the target proteins degraded by IMiDs are limited and further studies are needed to better elucidate their mode of action. ${ }^{88}$

### 1.8.2.2 Von Hippel-Lindau (VHL) E3 ligase

Von Hippel-Lindau tumor suppressor protein (VHL) is the substrate receptor of the VHL E3 ubiquitin ligase. Other members of this complex are a RING finger protein (Rbx1), which interact with E2 enzyme and adaptor proteins (Elongin B [EloB] and Elongin C [EloC]) able to connect VHL to a cullin protein (Cul2). The most important substrate of VHL is the hypoxia inducible factor $1 \alpha$ (HIF-1 $\alpha$ ), a transcription factor constitutively expressed, which activate the transcription of several genes in response to low oxygen conditions. To balance its intracellular levels, HIF-1 $\alpha$ is selectively hydroxylated by prolyl hydroxylase domain (PHD) enzymes at two specific proline residues within the HIF-1 $\alpha$ oxygen-dependent degradation domain (ODD). Only the hydroxylated form is recognized by VHL E3 ligase complex, ubiquitinated and degraded by UPS system. ${ }^{92}$

Considering that VHL interact with HIF-1 $\alpha$ by hydroxyproline residue, this scaffold was used as starting point for the design of selective VHL ligands (Figure 1.24).


II


VH-032


VH-298

Figure 1.24 Structure of VHL-ligands.

The first VHL ligand (II) has a low micromolar affinity ( $\mathrm{K}_{\mathrm{D}}$ of $5.4 \mu \mathrm{M}$ ) and the same binding site of HIF-1 $\alpha .^{93}$

To optimize binding affinity and lipophilicity, Ciulli group developed VHL ligands inserting different substituents on the hydroxyproline central scaffold to fill the left-hand side (LHS) and right-hand side (RHS) of the protein-protein interaction surface. Maintaining the aryl group of ligand II, the RHS was filled with a 4-methylthiazole, which does not alter the conformation of the protein. On the other hand, the LHS sub pocket was occupied firstly by a $t$-butyl group, then by an acetamido group, with important interactions of the carbonyl group within the pocket. In this way, VH032 was designed as VHL ligand with improved binding affinity $\left(K_{D}\right.$ of 0.185 $\mu \mathrm{M})$ and lipophilicity. ${ }^{94}$ The replacement of the terminal methyl group of VH032 with a cyanocyclopropyl group, which better fill the LHS pocket, furnished VH298 ligand. This compound has affinity in the nanomolar range ( $\mathrm{K}_{\mathrm{D}}$ of 90 nM ), higher passive cell permeability and good cellular activity. ${ }^{95}$

Importantly, loss of binding specificity occurs changing the stereochemistry at the carbon atom bearing the hydroxyl group of the hydroxyproline ring. ${ }^{95}$

### 1.8.3 BET degraders

The possibility of drive the degradation of a specific target was demonstrated nearly 20 years ago with the development of the first peptide-based degrader. The use of synthetic compounds
instead of a peptidic ligands prompted the development of several PROTACs as potential drugs. ${ }^{96}$ In the BET field, several degraders have been developed (Figure 1.25).



CRBN-BASED BET PROTACs


Figure $\mathbf{1 . 2 5} \mathrm{BET}$ degraders.

In 2015, the first examples of CRBN-based BET PROTACs were reported: dBET1 and ARV-825. These molecules share CRBN ligand as E3 recruiter and (+)-JQ1 as BET warhead, differing for the nature and length of linkers. ${ }^{97-98}$ These compounds are efficient degraders of all BET proteins, resulting in anti-tumor efficacy and cell-proliferation inhibition more effective than small-molecule BRD4 ligands. In 2018, two BET degraders were developed using as E3 recruiter CRBN ligands. dBET260 has an azacarbazole-based scaffold as BET binder, showing BET degradation at concentrations as low as 30 pM in the RS4; 11 leukemia
cell line. ${ }^{99}$ QCA570 contain a $[1,4]$ oxazepines-BET ligands and is able to induce degradation at low picomolar concentrations. ${ }^{100}$

On the other hand, VHL-based PROTACs have been developed. Ciulli and co-workers designed MZ1 as a potent BET degrader, connecting JQ1 to a VHL-recruiting ligand, using a PEG3 as linker. Surprisingly, MZ1 showed a preferential degradation for BRD4 over BRD2 and BRD3 despite JQ1 is a pan-BET inhibitor. ${ }^{101}$ The crystal structure of MZ1 in complex with VHL and the second bromodomain of BRD4 clarify that the formation of a ternary complex induce new protein-protein and protein-ligand contacts that contributing to the high stability and cooperativity of ternary complex. ${ }^{102}$ Higher selectivity for the second bromodomains over the first bromodomains of BETs is maintained by the first macrocyclic PROTAC developed by the same research group starting from MZ1. ${ }^{103}$ Similar to MZ1 but with a linker of different nature and length, ARV-771 is an effective BET degrader, offering a good therapeutic strategy in a castration-resistant prostate cancer (CRPC). ${ }^{104}$

Using tetrahydroquinoline as BET ligand and VHL as E3 recruiter, MZP-61 was developed. Despite the BET warhead has high potency, the resulting PROTAC is less potent then the related JQ1-based PROTACs. This is due a negative cooperativities of ternary complex formation, which can affect the activity of the degraders. ${ }^{105}$

### 1.8.4 Advantages and limits of PROTAC approach

The good results obtained in different cellular studies suggested several potential benefits of induced-selective degradation. First of all, the chosen ligand can bind anywhere on the target protein, also if does not have the ability to block specific functions of the protein, making druggable proteins for which the design of selective small-molecule is challenging. Through targeted protein degradation, the selectivity of a ligand can be modulated, taking advantages from the new protein-protein interactions within the ternary complex. ${ }^{106}$

The catalytic mode of action of PROTACs is associated with low level of target protein occupancy at concentrations much lower than classical small-molecules, avoiding off-target effects due to high occupancy of targets or high doses. Once that PROTACs restore proteins to basal levels, its action is limited to a smaller pool of the novo resynthesized proteins, involving low tissue concentrations of degraders. In some cell, the time to recover the sufficient pool of proteins to reintroduce physiological signaling may be long, increasing the duration of action of PROTACs. The low drug exposures, low doses, and long dosing interval can overcome the possibly low oral absorption with the development of formulations for controlled release. ${ }^{106}$

On the other hand, PROTACs have several limits due to their structure, physicochemical properties and their mode of action.

Most of the compounds have a high molecular weight and a complex structure, which complicate their synthesis and purifications. The nature and the length of the linkers should be choosing carefully, considering the high oxidative metabolism especially at this level and possibly secondary interactions. ${ }^{107}$ Physicochemical properties such as solubility, chemical stability and non-specific binding to protein in media should be considered also for the quality and the interpretation of data.

Moreover, the complex mode of action of PROTACs should be examined. First of all, the designed degraders must access the proper intracellular compartment in which both the targeted protein and the E3 ligase are located. Often, cell permeability is a big limitation, considering the high molecular weight and total polar surface area of compounds. Once inside, the degraders must bind at the same time both with the target protein and the E3 ligase, forming a stable ternary complex (POI-PROTACs-E3 LIGASE). ${ }^{106}$ High concentrations of PROTAC can promote the binary complex instead the ternary system. This aspect, called 'hook effect', could inhibit substrate degradation and make dosing of PROTACs in patients a tricky challenge. ${ }^{106}$

Once formed, the ternary complex must allow the two partners to adopt an appropriate conformation suitable for the transfer of ubiquitin(s) to the right acceptor site and with sufficient
efficiency, faster than the lifetime of the ternary complex and the action of deubiquitinase enzymes. The ubiquitin transferred should be accessible and detectable by proteasome to start the degradation process. Protein ubiquitination could be not sufficient for protein degradation: modifications in protein conformation can alter the recognition by the proteasome, avoiding the degradation process. Moreover, the rate of protein degradation should be faster than the de novo resynthesis, an important aspect to be considered. ${ }^{106}$

## CHAPTER II

## AIM OF WORK

### 2.1 Aim of work

Despite the excellent progress made in the BET field, there is still a need to clarify all the pathways in which BET proteins are involved and their connections with pathological states in different cell types.

Although the promising results, the outcomes obtained from BET ligands in clinical trials reveal critical points, suggesting different responses in clinical efficacy from preclinical models. Efficacy, dose optimization, metabolism, toxic effects, mechanisms of resistance and limited long-term efficiency are key aspects to be optimized. ${ }^{22,53}$

In this context, the identification of compounds with improved selectivity toward BET family members and/or their singular bromodomain (BD1 and BD2) offers a valuable chemical tool to ascertain their role during various physiological and physio-pathological conditions.

Aim of this PhD project is the design, synthesis, biochemical and biophysical evaluation of new chemical probes for BET proteins, in order to obtain potent compounds with improved selectivity. To this purpose, three different medicinal approaches were applied to obtain different classes of compounds (Figure 2.1):
the bisubstrate approach;
the frozen analogue approach;
the PROTAC approach.


Figure 2.1 Aim of work: different medicinal chemistry approaches to obtain selective BET ligands.

### 2.2 Bisubstrate approach

The presence of two bromodomains in BET proteins makes these proteins suitable for the development of bivalent ligands.

Compounds presented here have been designed with the aim to bind simultaneously the first and the second bromodomain of the same BET protein. Such derivatives could allow, for the first time, the co-crystallization of both bromodomains within the same BET protein and therefore obtain important structural information for the design of selective ligands. The design strategy, illustrated in Figure 2.2, involved the connection of a BD1 and a BD2 selective ligand with proper linkers.


Figure 2.2 Design of bivalent ligands bridging BD1/BD2 selective ligands with proper linkers.

As depicted in Figure 2.2, Cl-BzT-7 and RVX-208 were selected as BD1 and BD2 selective ligands, respectively.

To define the putative proper connection position for the attachment of the linkers, the crystal structures of BzT-7, the deschlorinated derivative of Cl-BzT-7, and of RVX-208 were considered.


Figure 2.3 Crystal structure of the first bromodomain of human BRD4 in complex with BzT7 (PDB 3U5L).

As the N-methyl group of BzT-7 points out the binding pocket, this position can be exploited to anchor the linker (Figure 2.3). Moreover, structure-activity relationships of related analogues (Section 1.5.1) indicate that this position well tolerate the presence of a carbonylic function. Therefore, a coupling reaction between the carboxylic acid analogue of Cl-BzT-7 and an amine could be exploited to insert the linker.


Figure 2.4 Crystal structure of the second bromodomain of human BRD2 in complex with RVX-208 (PDB 4MR6).

For RVX-208, the crystal structure reveals that the hydroxy-ethylether group in 4'of RVX208 points out of the binding pocket and makes only a few contacts with the bromodomain surface (Figure 2.4). Therefore, this position was selected as point for the attachment of the linker.

Evaluated the proper attachment position, different linkers are chosen. Considering that the role of the linker is critical and cannot be clearly anticipated, we selected a broad range with different shape, polarity and length. The combination of these three units yielded the collection of compounds depicted in Table 2.1.

It has to be noted that, in compound $\mathbf{1}$ the BD 1 and BD 2 ligands were directly connected. Compound 14, that features a different point of linker attachment on RVX-208, was designed in order to confirm the accuracy of the attachment point previously selected.

Table 2.1 Bivalent ligands (compounds 1-14).
$\mathbf{4}$

### 2.3 Frozen analogue-approach

Among the selective ligands of the first bromodomain of BET proteins, SARs of diazobenzene-based compounds, MS436 and MS611 have been extensively investigated (Section 1.6). Despite their good potency and selectivity for BD1 over BD2 of BRD4, the presence of diazo-group makes these compound metabolic unstable and prevents their further development. Exploiting the frozen analog approach, the diazo-group of this class of compound was constrained into an imidazole yielding benzimidazole derivatives. The benzimidazole scaffold is a privileged chemotype, synthetic accessible and metabolically stable. Moreover, the conformational restriction of a flexible ligand improves drug-likeness and could positively affect potency and selectivity. ${ }^{108-110}$

Taking in account also the SARs of diazobenzene-based compounds, a small library of benzimidazole-based ligands was designed (Table 2.2).

Table 2.2 Benzimidazole-based ligands (compounds $\mathbf{1 5}$ - 32)


| 19 (EML796) | $-\mathrm{CH}_{3}$ | -H | $-\mathrm{CH}_{3}$ | -OH | $-\mathrm{CH}_{3}$ | -H |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 20 (EML801) |  | -H | $-\mathrm{CH}_{3}$ | -OH | $-\mathrm{CH}_{3}$ | -H |
| 21 (EML 797) |  | -H | $-\mathrm{CH}_{3}$ | -OH | $-\mathrm{CH}_{3}$ | -H |
| 22 (EML802) |  | -H | $-\mathrm{CH}_{3}$ | -OH | $-\mathrm{CH}_{3}$ | -H |
| 23 (EML 760) |  | -H | -H | -OH | -H | -H |
| 24 (EML761) |  | -H | -OH | -H | -H | -H |
| 25 (EML762) |  | -H | -H | -Cl | -H | -H |
| 26 (EML763) |  | -OH | -H | -OH | -H | -H |
| 27 (EML 764) |  | -H | -OH | -OH | -H | -H |
| 28 (EML766) |  | $-\mathrm{CH}_{3}$ | -H | -OH | -H | $-\mathrm{CH}_{3}$ |
| 29 (EML806) |  | -H | $-\mathrm{CH}_{3}$ | -OH | -H | -H |
| 30 (EML799) |  | $\approx$ | $-\mathrm{CH}_{3}$ | -OH | $-\mathrm{CH}_{3}$ | -H |
| 31 (EML804) |  |  | $-\mathrm{CH}_{3}$ | -OH | $-\mathrm{CH}_{3}$ | -H |
| 32 (EML805) | S |  | $-\mathrm{CH}_{3}$ | -OH | $-\mathrm{CH}_{3}$ | -H |

First, as proof of concept of the frozen analog approach strategy, compound 15, the corresponding benzimidazole analog of MS436, was synthesized.

Then, a SAR study was undertaken on the substituent on the sulfonamide nitrogen (compounds 16 - 22). Detected the best substituent in this position, structural modifications of the phenyl ring in position 2 of the benzimidazole core were explored (compounds 23 - 32). Bearing in mind SARs of diazobenzene-based compounds, hydroxyl and methyl groups were preferentially selected. In fact, in MS436, the hydroxyl group on the phenyl ring forms a hydrogen bond with the conserved asparagine residue in the KAc binding pocket and, together with the ortho methyl group, acts as KAc mimetic.

### 2.4 PROTAC approach

Despite the remarkable results obtained with BET PROTACs developed so far, all of them contain a pan-BET ligand as BET warhead. It is simple to speculate that using selective BET ligands the effectiveness of this type of molecules could be improved.

Therefore, using as BET warhead BD1 and BD2 selective ligands, a collection of BET PROTACs was designed (Figure 2.5). This part of the PhD project was developed at the School of Life Sciences (University of Dundee) under the supervision of Prof. Alessio Ciulli.


Figure 2.5 Design of PROTACs with selective BET ligands

EML765, the benzimidazole-based compound identified during this PhD project (Section 2.3), was selected as BD1-selective ligand.

To define the proper connection position for the attachment of the linker on the benzimidazole scaffold, the crystal structure of MS436, the diazobenzene analogue of EML765 was exploited (Figure 2.6).



15, EML765

Figure 2.6 crystal Structure of the first bromodomain of human BRD4 in complex with MS436 (PDB 4NUD).

Assuming similar binding mode for the two compounds, position 5 of the pyridine ring seems to be not involved in important interactions with the protein. Therefore, this position was identified as linker-attachment point and functionalized with an alkynyl group for proper linker connection (15A, Figure 2.7).

RVX-208 was again picked as BD2 selective ligand. In analogy to the design of bivalent ligands (Section 2.2), the hydroxy-ethylether group was chosen as appropriate position for linker attachment.

A derivative of thalidomide and a proline-based compound were exploited as Cereblon (CRBN) and von Hippel-Lindau (VHL) ligands, respectively. Finally, four different linkers were used.

The combination of these components yielded 16 compounds depicted in Figure 2.7.


Figure 2.7 Combination of components for PROTACs design.

Biophysical, biochemical and cellular assays were exploited to evaluate the activity and the selectivity of designed compounds (Section 3). The synthetic strategies used to obtain the target compounds are detailed in Section 4.

## CHAPTER III

## RESULTS AND DISCUSSION

### 3.1 Bivalent ligands

The activity of the bivalent ligands $\mathbf{1}$ - $\mathbf{1 4}$ (Table 2.1) was evaluated by research group of Prof. Panagis Filippakopoulos by means of two different assays. Differential Scanning Fluorimetry (DSF) was performed for the preliminary evaluation of the binding properties. Sedimentation velocity assay was performed to investigate the ability of compounds to engage simultaneously the first and the second bromodomain.

### 3.1.1 Preliminary biophysical evaluation: DSF assay

DSF assay is a rapid and inexpensive screening technology helpful to identify ligands that bind the target protein. Briefly, the increase of temperature promotes the transition of a protein from the native to the unfolded state and the thermal stability of a protein is quantified as the midpoint of thermal denaturation or melting point (Tm), that is the temperature at which both native and unfolded states are equimolar. The presence of ligand that binds the protein stabilize the folded protein and, consequently, shift its Tm in concentration and potency-dependent manner. In DSF, a compound with a low fluorescence signal in a polar environment and high fluorescence in a non-polar environment is added to a protein solution and, during heating, the fluorescence of the solution is monitored. During protein unfolding, the hydrophobic core becomes exposed and the fluorescence increase until the protein is completely denatured and, thus, the Tm can be determined. The difference of Tm in the presence or in the absence of a specific ligand $(\Delta \mathrm{Tm})$ could be related to its affinity for the protein. ${ }^{111}$

Compounds were screened in a panel of proteins containing the first (BD1) and the second (BD2) bromodomain of the four BET proteins and four selected bromodomain-containing proteins outside the BET family, in order to evaluate off-target effects (Figure 3.1).

Compound 2 was not evaluated in DSF assay because it is intensely colored and, consequently, could interfere in the assay. The results obtained for compound 4 were not considered because the resultant curve gave multiple transitions.


Figure 3.1 DSF data of compounds 1-14 and of RVX-208 and JQ1 used as reference compounds. Compounds have been initially screened at $50 \mu \mathrm{M}$ concentration. Temperature shift data are color coded as indicated in the figure. ${ }^{\text {a }}$ not tested: bright yellow compound which interfere in the assay; ${ }^{\mathrm{b}}$ the DSF curve gave multiple transitions.

First of all, the inactivity of compounds $\mathbf{1}$ and $\mathbf{1 4}$, which features no separation between the BD 1 and BD 2 warheads and a different point of linker attachment on BD 2 ligand, respectively, confirms that a linker at 4' position of the pendant phenyl ring of RVX-208, is essential for activity. Moreover, all the tested compounds have no off target effects, being inactive on the selected bromodomain-containing proteins outside the BET family.

Then, we investigated the role of the linker.
Compounds 3 and 5-7, featuring only PEGs linker, are able to bind both BD1 and BD2 domain with a preference for BD2. Generally, increasing the length of the chain decrease the affinity and compound $\mathbf{3}$ proved to be the more potent.

The presence of phenyl ring bridging the triazobenzotriazepine warhead to the PEG linker drastically decrease the affinity of compounds. In fact, compounds $\mathbf{8}$ and $\mathbf{9}$ showed only modest activity in all the proteins tested and, therefore, this type of linker was not further investigated. It is to note that his outcome was unpredicted as an aromatic amide is present in some diazepinebased compounds (i.e. OTX-015, Section 1.5.1)

To evaluate the role of the linker flexibility, a triazole ring was introduced as a spacer between the two BET ligands, connecting the triazolobenzotriazepine warhead with an alkyl chain (compounds 10 and 11) or a PEG chain (compounds 12 and 13).

Results represented in Figure 3.1 indicate that the presence of the triazole ring is generally well tolerated and, consistently to the results obtained with compounds $\mathbf{3 - 7}$, increasing the distance between the two warheads decreases the affinity. Again, the shorter derivative (compound 10) is the more potent of the series.

Thus, compounds $\mathbf{3}$ and $\mathbf{1 0}$ were selected for further evaluation.

### 3.1.2 Sedimentation velocity assay

Sedimentation velocity assay was performed to investigate the ability of compound $\mathbf{3}$ (EML896) and compound 10 (EML823) to engage simultaneously the first and the second bromodomain.

This technique is an analytical method of ultracentrifugation that measures the movement of proteins under a high centrifugal force to define the size, shape and interactions of proteins. The high speed shifts the proteins from the center to the external of the rotor until all the proteins form a layer outside. The rate of this process is function of the sedimentation coefficient of the protein, influenced by the molecular weight and by molecular shape. Linear proteins have more hydrodynamic friction and, consequently, a smaller sedimentation coefficient than a globular protein of the same molecular weight. ${ }^{112}$

Upon binding with a bivalent ligand which engage both BD1/BD2 domain, the protein adopts a more globular conformation, changing its sedimentation velocity. Moreover, changes in the molecular weight of proteins can prove if there is an intramolecular or intermolecular binding.

The sedimentation velocity assay performed with the compound 10 (EML823) and the tandem BRD4 (the constructs that contain BD1, BD2 and the long 15kDa linker), indicated that the compound engages only one side of the protein. In fact, as illustrated in Figure 3.2, EML823 does not distort the protein and does not change the sedimentation velocity of the tandem BRD4 (dotted black).



Figure 3.2 Sedimentation velocity of tandem BRD4 (1:2) with no ligand (dotted black), with EML823 (in blue) and with a reference compound (Bvb-7, in red).

On the other hand, compound 3 (EML896) gave positive results. As illustrated in Figure 3.3, the compound changes the size and shape of sedimentation velocity of all four BET proteins (the tandem constructs of BRD2, BRD3, BRD4 and BRDT). The shape change is more evident in the case of BRD2 and BRD4 (Figure 3.3, panel band d). Moreover, the molecular weight of proteins is unchanged, suggesting that the both domain within the same proteins are engaged. However, it seems that the compound binds preferentially one protein site. In fact, as depicted in Figure 3.3, panel e, the distributions are not homogenous: the compound distorts all the protein binding to both sites, but, at the same time, binds only to one site keeping the protein
linear. Nevertheless, these results are promising for the development of bivalent compounds and compound EML896, together with the information on the role of the linker obtained in the present work, could be considered a valuable lead.







Figure 3.3 (a) sedimentation velocity of tandem BRDT (1:2) with no ligand (dotted black), with 10 (EML896) (in green); (b) sedimentation velocity of tandem BRD2 (1:2) with no ligand (dotted black), with 10 (EML896) (in red); (c) sedimentation velocity of tandem BRD3 (1:2) with no ligand (dotted black), with 10 (EML896) (in blu); (d) sedimentation velocity of tandem BRD4 (1:2) with no ligand (dotted black), with $\mathbf{1 0}$ (EML896) (in light blue); (e) 3D distribution. Higher f/fo represents the linear species and the lower one is the compacted BD1/BD2 engaged complex.

### 3.2 Benzimidazole-based ligands

The affinity of all benzimidazole-based ligands (compounds $\mathbf{1 5}$ - 32, Table 2.2) was preliminary evaluated by DSF assay (Prof. Panagis Filippakopoulos). Subsequently, the biological profile of 15 (EML 765), the most potent and selective derivative was further explored by means of isothermal titration calorimetry (Prof. Panagis Filippakopoulos) and timeresolved fluorescence resonance energy transfer (Prof. Alessio Ciulli) assays.

### 3.2.1 Preliminary biophysical evaluation: DSF assay

The ability of benzimidazole-based ligands to bind a wide panel of bromodomain-containing proteins was preliminarily evaluated using the DSF assay (Figure 3.4).


Figure 3.4 DSF data measured on compounds $\mathbf{1 5 - 3 2}$ and on JQ1 used as reference compound. The compounds have been initially screened at $50 \mu \mathrm{M}$ concentration. Temperature shift data are color coded as indicated in the figure.

The binding properties of $\mathbf{1 5}$, the benzimidazole analog of MS436, validate the efficacy of our frozen analogue approach. Indeed, even if with a lower $\Delta \mathrm{Tm}$ than the unselective JQ1, compound $\mathbf{1 5}$ increase the melting temperature of all the BET proteins but the effect is more marked for BRD4 (1), the first bromodomain of BRD4. The investigation of the substitutions on the sulfonamide nitrogen revealed that the pyridine ring is essential for activity and selectivity. In fact, the primary sulfonamide (18) and the methyl sulfonamide (19) have no affinity toward all the BET domains considered. Similar results were observed when the pyridine ring is detached from the sulphonamide function (16 - 17). Surprisingly, the replacement of pyridine ring with a cyclopropyl group (20) results in a completely lost of activity toward BET members and a strong off-target binding on bromodomain of CBP.

Instead, the insertion of a cyclopentyl or cyclohexyl group (21 and 22, respectively) results in compounds with similar activity to $\mathbf{1 5}$ toward the first bromodomain of BRD4 but less selectivity, with activity on other non-BET bromodomain containing proteins.

Based on these data, the pyridine sulfonamide portion was maintained and a structure-activity relationship (SAR) study was undertaken on the phenyl ring in position 2 of the benzimidazole core. The main result is that the 3,5-dimethyl-4-phenolic group is fundamental for the activity. In fact, the shift of both methyl group (28) or the removal of one methyl (29) results in compounds completely inactive. The removal of both methyl group (23) results in compound with a weaker affinity compared to compound $\mathbf{1 5}$. The replacement of the 4 -hydroxyl group with a chlorine (25), its shift in position 3 (compound 24) or the 3,4-dihydroxyphenyl derivative (compound 27) results in completely inactive molecules. Instead the 2,4-dihydroxyphenyl derivative (compound 26) acts as a promiscuous bromodomain binder. These results corroborate the hypothesis that the 3,5-dimethyl-4-phenolic group acts as KAc mimetic group and was not changed. To evaluate the impact on the activity and on selectivity of substituents with high steric hindrance, a phenyl ring was introduced in position 2 (compounds $\mathbf{3 0}-\mathbf{3 2}$ ). As represented in Figure 3.4, all these modifications are detrimental for the activity and the parent compound $\mathbf{1 5}$ remains the derivative with highest affinity.

### 3.2.2 ITC assay

As the DSF assay gave only qualitative information, to quantify the binding of $\mathbf{1 5}$ (hereafter referred to as EML765), Isothermal Titration Calorimetry (ITC) was performed.

This physical assay measures the thermal changes occurred when a ligand bind the target protein contained in the sample cell. The ITC instrument has two identical cells: the sample cell, containing the target protein and a reference cell, kept at the same temperature. When an interaction occurs, the instrument regulates the temperature of the reference cell to maintain the
thermal equilibrium between the two cells. In this way, the binding affinity, the enthalpy changes and the binding stoichiometry can be determined. ${ }^{113}$

We were pleased to find that EML765 has a strong affinity ( $\mathrm{K}_{\mathrm{D}}$ of 70 nM ) toward BRD4 (BD1) comparable to $\mathbf{J Q 1}\left(\mathrm{K}_{\mathrm{D}}\right.$ of 50 nM ), used as reference compound (Figure 3.5).


Figure 3.5 ITC data related to the binding of compound EML765 to the first bromodomain of BRD4.

### 3.2.3 Selectivity profile of EML765

The preliminary results of EML765 obtained with the DSF screening suggested that the compound is able to discriminate BD1 and BD2 domain of BET proteins. Thus, the selectivity of the compound was investigated.

With this aim, during the time spending at the University of Dundee, under the supervision of Prof. Alessio Ciulli, I performed a time-resolved fluorescence resonance energy transfer (TRFRET) binding assay. This method involves an energy transfer from a donor to an acceptor chromophore. Upon excitation, the donor chromophore can transfer the acquired energy to an acceptor donor if there is a small distance between them. According to literature, in our assay $6 \times$ His-tag BD1 or BD2 domain of each BET protein interacts with europium chelate-labeled anti-6His antibody, a donor chromophore. At the same time, the Alexa647 dye conjugated to
the BET ligand $\mathbf{J Q} \mathbf{- 1}$ is added: the binding of $\mathbf{J Q 1}$ to the protein bring the acceptor (Alexa647) in close proximity to the donor chromophore. Upon excitation, there is an energy transfer from donor to acceptor resulting in the emission of a photon, showing FRET fluorescence.

In presence of a ligand able to bind the protein, there is competition with $\mathbf{J Q} \mathbf{- 1}$ conjugated to Alexa647 dye and, consequently, a decreased FRET signal (Figure 3.6). ${ }^{74,114}$

FRET


Figure 3.6 Schematic representation of TR-FRET system for BET proteins and the selected ligand (EML765).

Data presented in Figure 3.7 and in Table 3.3 confirms EML765 as selective ligand of the first bromodomain of BET proteins. In fact, the affinity for BD1 ranges from 0.126 to 0.252 $\mu \mathrm{M}$ while for BD 2 ranges from 2.2 to $22 \mu \mathrm{M}$. The best result is in BRD 2 , in which the selectivity ratio for the binding to BD 1 and BD 2 is 170 (Table 3.1).


Figure 3.7 TR-FRET assay of the compound EML765 (in red) toward the first and the second bromodomain of BET proteins. The compound JQ1 (in blue) was used as reference compound.

Table 3.1 Selectivity profile of EML765

| TARGET PROTEIN | EML765 (KD $\mu \mathrm{M})$ | BD1 SELECTIVITY INDEX |
| :---: | :---: | :---: |
| BRD4 BD1 | 0.252 | 10 |
| BRD4 BD2 | 2.2 | 170 |
| BRD2 BD1 | 0.126 | 40 |
| BRD2 BD2 | 22 |  |
| BRD3 BD1 | 0.18 |  |
| BRD3 BD2 | 7 |  |

Taken together, these data validate EML765 as new lead compound for the development of potent and selective ligands for the first bromodomain of BET proteins.

### 3.3 BET PROTACs compounds

The ability of BET PROTACs (compounds 33 - 48, Figure 2.7) to induce proteindegradation was evaluated in two different cell lines.

Compounds were firstly profiled in HEK-293 cells, a BET-insensitive cell line, which does not undergo to cell-death upon BET inhibition. Firstly, cells were treated for 6 h with compounds at two different concentrations ( $1 \mu \mathrm{M}$ and $0.01 \mu \mathrm{M}$ ) and then the lysates were immunoblotted with BRD2, BRD3 and BRD4 specific antibodies. MZ1 and the inactive cisMZ1 were used as positive and negatively control, respectively. Unfortunately, no significant reduction in protein levels for all the compounds tested was observed (Figure 3.8).

Same results were obtained with longer incubation time (16 hours) (Figure 3.9).


Figure 3.8 Protein degradation profile of BET degraders visualized by immunoblot using antiBRD4, anti-BRD2 and anti-BRD3 antibody. HEK293 cells were treated for 6 h at $1 \mu \mathrm{M}$ and $0.01 \mu \mathrm{M}$ with inhibitors, positive control (MZ1), negative control (cisMZ1) and DMSO control.


Figure 3.9 Protein degradation profile of BET degraders visualized by immunoblot using antiBRD4, anti-BRD2 and anti-BRD3 antibody. HEK293 cells were treated for 16 h at $1 \mu \mathrm{M}$ and $0.01 \mu \mathrm{M}$ with inhibitors, positive control (MZ1), negative control (cisMZ1) and DMSO control.

In order to eliminate potential false-negatives due to the "Hook effect", the compounds were tested also at lower concentrations $(0.1 \mu \mathrm{M}$ and $0.001 \mu \mathrm{M})$. Even in this case, the compounds do not induce protein degradation for BRD2 (Figure 3.10) and BRD3 (Figure 3.11).


Figure 3.10 Protein degradation profile of BET degraders visualized by immunoblot using anti-anti-BRD2 antibody. HEK293 cells were treated for 16 h at $1.0 \mu \mathrm{M}, 0.1 \mu \mathrm{M}, 0.01 \mu \mathrm{M}$ and $0.001 \mu \mathrm{M}$ with inhibitors, positive control (MZ1), negative control (cisMZ1) and DMSO control.


Figure 3.11 Protein degradation profile of BET degraders visualized by immunoblot using anti-anti-BRD3 antibody. HEK293 cells were treated for 16 h at $1.0 \mu \mathrm{M}, 0.1 \mu \mathrm{M}, 0.01 \mu \mathrm{M}$ and $0.001 \mu \mathrm{M}$ with inhibitors, positive control (MZ1), negative control (cisMZ1) and DMSO control.

These results were confirmed using Simple Western ${ }^{\text {TM }}$ Jess, an innovative automated capillary-based immunoassay that is supposed to be more reproducible and more sensitive than traditional method. Using an optimized method, the assay was performed only for BRD4 and the cells were treated for 16 h at four different concentrations $(1.0 \mu \mathrm{M}, 0.1 \mu \mathrm{M}, 0.01 \mu \mathrm{M}$ and $0.001 \mu \mathrm{M}$ ). Also in this assay, however, no protein degradation was observed (Figure 3.12).


Figure 3.12 Protein degradation profile of BET degraders obtained using the Jess, an automated protein separator. HEK293 cells were treated for 16 h at $1 \mu \mathrm{M}, 0.1 \mu \mathrm{M}, 0.01 \mu \mathrm{M}$ and $0.001 \mu \mathrm{M}$ using anti-BRD4 as antibody.

Taken together, these data definitely showed that compounds $\mathbf{3 3}$ - $\mathbf{4 8}$ are not able to induce protein-degradation in in HEK-293 cells.

Our PROTACs were also evaluated in MV4-11 cells, a BET-sensitive cell line which undergoes to cell-death upon BET inhibition. We evaluate the ability of compounds to induce cell-death using the cell viability assay. As depicted in Figure 3.13, compounds $\mathbf{3 3 - 4 8}$ do not induce substantial cell-death.


Figure 3.13 Cell viability assay. MV4-11 cells were treated with the PROTAC compounds, positive control (MZ1) and negative control (cisMZ1) for 48 h . Cell viability was measured with Promega CellTiter-Glo luminescent cell viability assay kit. Data were analyzed with Graphpad Prism software.

Despite the negative results, the designed PROTAC compounds can be considered a starting point for further evaluation and optimization. Indeed, the inactivity of compounds it is not unusual in PROTACs field, considering that several aspects can influence the activity of degraders. We suspect that the length and the nature of the linkers could be not suitable for the proper engagement of target protein and E3 ligase and can be optimized. We can also speculate that the compounds engage just one component of the system (the target protein or the E3 ligase) promoting the formation of a binary system instead the ternary complex or that they do not promote a stable ternary complex. In addition, the compounds could have a low cellular permeability. ${ }^{105-106}$

All these aspects should be better investigated. For examples, we can evaluate possible changes in cellular activity using the electroporation to increase the permeability of the cell membrane or perform in vitro assays to better evaluate the formation of binary and/or ternary complex.

## CHAPTER IV

## CHEMISTRY

### 4.1 Synthesis of bivalent ligands

This Section describes the synthetic procedures adopted for the preparation of target compounds. A general synthetic approach was used for compounds $\mathbf{1} \mathbf{- 1 1}$, while for compounds $\mathbf{1 2 - 1 4}$ other two different approaches have to be used.

### 4.1.1 Synthesis of bivalent ligands 1 - 11

Compounds 1 - 11 were obtained by amide coupling of compound 49, Cl-BzT-7 featuring a carboxylic function, and the proper amino-linkers attached on hydroxy-ethylether group in 4' of RVX-208 (50, 51a - f, 52a,b and 53a,b) using EDC and HOBt as peptide coupling reagents (Scheme 4.1). As summarized in Table 4.1, the yields ranged from moderate to good.

Scheme 4.1 General synthetic scheme for the preparation of compounds 1 - 11


Reagents and conditions: (a) EDC hydrochloride, HOBt, NMM, dry DMF, r. t., 4 h.

Table 4.1 Compounds 1-11
50; $\mathbf{1}$ (EML807)

### 4.1.1.1 Synthesis of triazolobenzotriazepine scaffold 49

The synthetic protocol for the preparation of compound $\mathbf{4 9}$ is reported in Scheme 4.2.

Scheme 4.2 Procedure for preparation of compound 49


Reagents and conditions: (a) $\mathrm{ZnCl}_{2}, 120{ }^{\circ} \mathrm{C}$ to $230{ }^{\circ} \mathrm{C}, 4 \mathrm{~h}$ (70\%); (b) ethyl hydrazinoacetate hydrochloride, dry EtOH, reflux, $18 \mathrm{~h}(79 \%)$; (c) TEA, $\left(\mathrm{CCl}_{3} \mathrm{O}\right)_{2} \mathrm{CO}$, dry DCM, $0^{\circ} \mathrm{C}$ to r. t., 18 h (85\%); (d) Lawesson's reagent, toluene, reflux, 18 h (55\%); (e) $\mathrm{NH}_{2}-\mathrm{NHAc}, \mathrm{Hg}(\mathrm{OAc})_{2}$, THF/AcOH (4:1), $95{ }^{\circ} \mathrm{C}$ (30\%); (f) $\mathrm{LiOH}, \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ (1:1) (93\%).

Reaction of 2-amino-4',5-dichlorobenzophenone (56), prepared from 4-chloroaniline (54) and 4-chloro benzoyl chloride (55) according to previously published procedure ${ }^{115}$, with ethyl hydrazinoacetate hydrochloride in anhydrous ethanol (dry EtOH ) allowed the formation of the corresponding imine (57). Intramolecular cyclization with triphosgene $\left.{ }^{116}\left(\mathrm{CCl}_{3} \mathrm{O}\right)_{2} \mathrm{CO}\right)$ yieldes the corresponding benzotriazepinone (58) which is then thionated with Lawesson's reagent to afford the thioamide 59, a more reactive derivative for the subsequent formation of triazole ring. The moderate yield (55\%) is in part due to poorly soluble byproducts which complicated purification procedures as well as the presence of ethyl ester group which can compete in the thionation reaction.

The triazole ring of compound $\mathbf{6 0}$ was obtained by mercury-mediated cyclization in presence of acetyl hydrazine at $80{ }^{\circ} \mathrm{C}$. The modest yield obtained in this one-step procedure is comparable to the reaction traditionally conducted in a three-step procedure, ${ }^{73}$ which involves
the treatment of compound $\mathbf{5 9}$ with hydrazine in tetrahydrofuran (THF), followed by acetylation with acetyl chloride and further cyclization in acidic conditions. Finally, hydrolysis of the ethyl ester with lithium hydroxide ( LiOH ) furnished the carboxylic acid derivative 49 in good yield (86\%).

### 4.1.1.2 Synthesis of amines 50, 51a - $\boldsymbol{f}$ and 52a,b

The amines 50, 51a - $\mathbf{f}$ and 52a,b were prepared as depicted in Scheme 4.3. The quinazolinone core was formed by a direct cyclocondensation-dehydrogenation of the synthesized aldehydes 61, 62a-f and 63a,b and the 2-amino-4,6-dimethoxybenzamide 64, in presence of sodium bisulfite $\left(\mathrm{NaHSO}_{3} 50 \%\right.$ in mixture with $\left.\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}\right)$ and a catalytic amount of p-toluenesulfonic acid (TsOH) in anhydrous $\mathrm{N}, \mathrm{N}$-dimethylacetamide (DMAc), affording the intermediates 65, 66a-f and 67a,b. Hydrazinolysis of N -substituted phthalimide $\mathbf{6 5}$ and palladium-on-carbon hydrogenation converted the azido group of intermediates $\mathbf{6 6 a}-\mathbf{6 6 f}$ and the nitro group of intermediates 67a,b into the corresponding amino derivatives 50,51a $-\mathbf{f}$ and 52a,b.

It has to be noted that the synthetic protocol involved the construction of RVX-208 scaffold on proper linkers. This strategy was preferred considering that the direct addition of the linkers on quinazolinone scaffold could give a mixture of alkylated byproducts due to the tautomerism of this scaffold.

Scheme 4.3 Procedure for preparation of amines 50, 51a-f, 52a,b


Reagents and conditions: (a) $\mathrm{NaHSO}_{3}\left(50 \%\right.$ in mixture), TsOH, DMAc, $120{ }^{\circ} \mathrm{C}, 18 \mathrm{~h}(40-80 \%)$; (b) $\mathrm{NH}_{2} \mathrm{NH}_{2}, \mathrm{EtOH}, \mathrm{reflux}, 3 \mathrm{~h}(85 \%)$; (c) $\mathrm{H}_{2}$, $\mathrm{Pd} / \mathrm{C}$, EtOH, 3 h ( $81-99 \%)$.

Scheme 4.4 Procedure for preparation of aldehydes 61, 62a-f and 63a,b


Reagents and conditions: (a) $\mathrm{MsCl}, \mathrm{TEA}$, dry DCM $0{ }^{\circ} \mathrm{C}$ to r. $\mathrm{t} .\left(99 \%\right.$ ); (b) $\mathrm{TsCl}, \mathrm{Ag}_{2} \mathrm{O}, \mathrm{KI}$, toluene, $0^{\circ} \mathrm{C}$ to r. $\mathrm{t} ., 18 \mathrm{~h}(53-82 \%)$; (c) $\mathrm{NaN}, \mathrm{DMF}$, r. $\mathrm{t} .$, 18 h ; (50-70\%); (d) p-nitrophenol, $\mathrm{K}_{2} \mathrm{CO}_{3}$, dry DMF, $80^{\circ} \mathrm{C}, 18 \mathrm{~h}(96-99 \%)$; (e) TsCl, TEA, dry DCM, r. t., $18 \mathrm{~h}(41-66 \%)$; (f) $\mathrm{K}_{2} \mathrm{CO}$, dry DMF , $80^{\circ} \mathrm{C}, 18 \mathrm{~h}(51-87 \%)$.

The aldehydes 61, 62a-f and 63a,b were prepared as reported in Scheme 4.4. The general synthetic route is based on the attachment of proper linkers on 4-hydroxy-3,5dimethylbenzaldehyde 76a, commercially available.

All the linkers used are properly activated converting their alcoholic functions in a good leaving group. The ethyl linker of compound $\mathbf{1}$ is prepared starting from the 2-phthalimidoethanol (68) commercially available. The alcoholic group of $\mathbf{6 8}$ is converted into the corresponding mesylate compound (69), using mesyl chloride and triethylamine (TEA) in anhydrous dichloromethane (DCM).

Linkers used for the synthesis of compounds 2-9 are prepared from the commercially available 1,8-octanediol (70a) or the poli(ethylene glycol)s (PEGs) (70b -f).

The starting materials are selectively monotosylated (71a - f) using tosyl chloride ( TsCl ) in presence of silver oxide $\left(\mathrm{Ag}_{2} \mathrm{O}\right)$ and a catalytic amount of potassium iodide (KI). This method allows a selective monotosylation of symmetrical diols using a stoichiometric amount of tosylating agent under mild conditions. ${ }^{117}$

Monotosylated compounds 71a - $\mathbf{f}$ reacted with sodium azide in $\mathrm{N}, \mathrm{N}$-dimethylformamide (DMF) affording the intermediates $\mathbf{7 2 a}-\mathbf{f}$. On the other hand, the intermediates 73a,b are prepared treating the monotosylated intermediates 71c,e with 4-nitrophenol, commercially available, in presence of potassium carbonate $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$ in anhydrous DMF.

Then, the obtained intermediates 72a-f and 73a,b are tosylated in presence of tosyl chloride and triethylamine in anhydrous DCM yielding the intermediates $\mathbf{7 4 a}-\mathbf{f}$ and $\mathbf{7 5 a}, \mathbf{b}$.

The activated linkers (intermediates $\mathbf{6 9}, \mathbf{7 4 a}-\mathbf{f}$ and $\mathbf{7 5 a}, \mathbf{b}$ ) can easily undergo nucleophilic substitution with the 4-hydroxy-3,5-dimethylbenzaldehyde 76a, commercially available, affording the corresponding aldehydes $\mathbf{6 1}, \mathbf{6 2} \mathbf{a}-\mathbf{f}$ and 63a,b in moderate or good yield (5187\%)

The 2-amino-4,6-dimethoxybenzamide (64) was prepared according a previously described procedure (Scheme 4.5). ${ }^{71}$

Scheme 4.5 Synthesis of 2-amino-4,6-dimethoxybenzamide 64


Reagents and conditions: (a) $\mathrm{HCl}_{(\mathrm{g})}, \mathrm{Et}_{2} \mathrm{O}(97 \%)$; (b) $(\mathrm{COCl})_{2}, 0^{\circ} \mathrm{C}$ to $170{ }^{\circ} \mathrm{C}, 90 \mathrm{~min}(61 \%)$; (c) $\mathrm{NaOH}, \mathrm{H}_{2} \mathrm{O}_{2}, 70^{\circ} \mathrm{C}$ to $100{ }^{\circ} \mathrm{C}$ for 2 h then r.t., $18 \mathrm{~h}(60 \%)$; (d) EDC hydrochloride, HOBt , NMM, ammonia hydroxide solution (33\%), dry THF, r. t., 4 h ( $88 \%$ ).

Briefly, the 3,5-dimethoxyaniline hydrochloride salt (78), obtained treating the commercially available amine (77) with hydrochloric acid, reacted with oxalyl chloride at high temperature leading to the formation of the 4,6-dimethoxy indolin-2,3-dione (79). Subsequent reaction with sodium hydroxide $(\mathrm{NaOH})$ and hydrogen peroxide $\left(\mathrm{H}_{2} \mathrm{O}_{2}\right)$ furnished the 2-amino-4,6-dimethoxybenzoic acid (80) which, reacting with ammonia hydroxide solution (33\%), in coupling condition, afford the desired compound 64.

### 4.1.1.3 Synthesis of amines 53a,b

The amines 53a,b were obtained according to a slight modification of the above-described procedure (Scheme 4.6). Indeed, the formation of 1,2,3-triazole scaffold involve a 'click chemistry reaction'. This reaction is a copper(I)-catalyzed variant of the Huisen 1,3-dipolar cycloaddition of azides and terminal alkynes for the 1,2,3-triazole formation. Accordingly, azides intermediates and terminal alkynes have been prepared.

Scheme 4.6 Procedure for preparation of amines 53a,b


Reagents and conditions: (a) $\mathrm{NaN}_{3}$, DMF, r. t., 18 h ( $80 \%$ ); (b) MsCl, TEA, dry DCM $0^{\circ} \mathrm{C}$ to r. t. 3 h ; (c) $\mathrm{K}_{2} \mathrm{CO}_{3}$, dry DMF, $80^{\circ} \mathrm{C}, 18 \mathrm{~h}$ ( 66 - $68 \%$ ); (d) $\mathrm{NaHSO}_{3}\left(50 \%\right.$ in mixture), TsOH, dry DMAc, $120{ }^{\circ} \mathrm{C}, 18 \mathrm{~h}(70-72 \%)$; (e) CuI, DIPEA, AcOH, DCM, r. t., $18 \mathrm{~h}(67-87 \%)$; (f) $\mathrm{NH}_{2} \mathrm{NH}_{2}$, EtOH, reflux, $3 \mathrm{~h}(81-83 \%)$.

Starting from the mesylate intermediate 69, synthesized as described in Scheme 4.3, the azide intermediate $\mathbf{8 1}$ is obtained under classical condition of nucleophilic substitution, using sodium azide as nucleophile. On the other hand, treatment of 3-butyn-1-ol (82a) and 5-hexyn-1-ol (82b), commercially available, with mesyl chloride ( MsCl ) give the mesylates $\mathbf{8 3 a}, \mathbf{b}$ which reacted with the 4-hydroxy-3,5-dimethylbenzaldehyde (76a) under classical conditions of nucleophilic substitution, affording the aldehydes $\mathbf{8 4 a}, \mathbf{b}$. The latters reacted with the 2-amino-4,6-dimethoxybenzamide (64) by a direct cyclocondensation-dehydrogenation affording the quinazolinone-base ligands 85a,b.

Obtained both azide (81) and alkyne intermediates (85a,b), copper iodide and catalytic amount of AcOH and DIPEA were used to obtain the triazole intermediates $\mathbf{8 6 a}, \mathbf{b}$ by click reaction in good yield (67-87\%). Finally, hydrazinolysis of N-substituted phthalimides 86a,b furnished the corresponding amines 53a,b.

### 4.1.2 Synthesis of bivalent ligands 12-13

Compounds $\mathbf{1 2}$ and $\mathbf{1 3}$ were obtained by a final 'click reaction' (Scheme 4.7) between the azido group of the first portion of the linker on Cl-BzT-7 (83) and the alkynyl group on the second portion of the linker attached on RVX-208 (85a,b) prepared as previously described (Scheme 4.6).

Scheme 4.7 Procedure for preparation of compounds 12-13


Reagents and conditions: (a) CuI, DIPEA, AcOH, DCM, r. t., 18 h (52-69\%).

Unlike the synthetic procedures adopted for the preparation of intermediates 86a,b, the triazole ring was formed in the last step of reaction, avoiding the use of free amine-PEG3-azide which could complicate the purification procedures.

The azide intermediate $\mathbf{8 7}$ was prepared as depicted in Scheme 4.8.
Key starting material is the amine-PEG3-azide 90, prepared according to previously published procedures. ${ }^{118}$ Briefly, the triethylene glycol 70c, commercially available, is tosylated (88) using tosyl chloride ( TsCl ) and potassium hydroxide $(\mathrm{KOH})$. The ditosyloxy triethylene glycol $\mathbf{8 8}$ reacts with an excess of sodium azide forming the diazide intermediate 89. Under Staudinger condition, the diazide intermediate 89 is selective reduced furnished the monoamine triethylene glycol as hydrochloride salt 90 in high yield ( $78 \%$ ). The latter was coupled with the intermediate 49 using EDC and HOBT as coupling reagent affording the azide intermediate 87 in good yield (75\%).

Scheme 4.8 Procedure for preparation of compound 87


Reagents and conditions: (a) $\mathrm{TsCl}, \mathrm{KOH}, \mathrm{DCM} 0{ }^{\circ} \mathrm{C}$ to r. t., 3 h (93\%); (b) $\mathrm{NaN}_{3}$, TBAI, DMF, $80^{\circ} \mathrm{C}, 18 \mathrm{~h}(99 \%)$; (c) $\mathrm{PhP}_{3}, \mathrm{HCl}, \mathrm{AcOEt}$, r. t., $18 \mathrm{~h}(78 \%)$; (d) EDC hydrochloride, HOBt , NMM, dry DMF, r. t., 4 h (75\%).

### 4.1.3 Synthesis of bivalent ligand 14

Compound 14 was obtained by a final coupling reaction between the carboxylic group of Cl-BzT-7 (49) and the amino group of the linker attached on quinazolinone side of RVX-208 (91) using EDC and HOBT as coupling reagents (Scheme 4.9).

Scheme 4.9 Procedure for preparation of bivalent ligand 14.


Reagents and conditions: (a) EDC hydrochloride, HOBt, NMM, dry DMF, r. t., 4 h (60\%).

To avoid a mixture of mono- and di-alkylated intermediates, as mentioned above, the first approach involved the insertion of a proper protecting group on the phenolic portion. Tosyl and acetyl group are used as protecting group with negative results.

In both cases, we didn't recover the desired product, with different problems during the attachment of the linker and/or the de-protection steps (Figure 4.1).


Figure 4.1 First synthetic strategy used for the attachment of the linker on quinazolinone scaffold.

These negative results prompted us to directly attach the linker on deprotected quinazolinone intermediate (Scheme 4.10).

Cyclocondensation-dehydrogenation in presence of $\mathrm{NaHSO}_{3}$ of the 2-amino-4,6dimethoxybenzamide (64) and the 4-hydroxy-3,5-dimethylbenzaldehyde 76a allowed the formation of the quinazolinone ligand $\mathbf{9 2}$. Nucleophilic substitution with the linker $\mathbf{7 4} \mathbf{c}$ prepared as previously described (Scheme 4.3) allowed the attachment of the linker on quinazolinone side ( $\mathbf{9 3}$ ). As expected, the modest yield ( $37 \%$ ) is due to the contemporary formation of monoand di-alkylated byproducts separated by flash-chromatography. Palladium-on-carbon hydrogenation converted the azido group into the amino compounds $\mathbf{9 1}$.

Scheme 4.10 Procedure for preparation of amine 91.


Reagents and conditions: (a) 76a, $\mathrm{NaHSO}_{3}$ ( $50 \%$ in mixture), TsOH , dry DMAc, $120^{\circ} \mathrm{C}, 18 \mathrm{~h}$ (94\%); (b) 74c, $\mathrm{K}_{2} \mathrm{CO}_{3}$, dry DMF, $80^{\circ} \mathrm{C}$, $18 \mathrm{~h}(37 \%)$; (c) $\mathrm{H}_{2}, \mathrm{Pd} / \mathrm{C}, \mathrm{EtOH}, 3 \mathrm{~h}(98 \%)$.

### 4.2 Synthesis of benzimidazole-based ligands

This Section describes the synthetic procedures adopted for the preparation of benzimidazole-based ligands $\mathbf{1 5} \mathbf{- 3 2}$. All the compounds were prepared following a synthetic protocol previously developed by us. ${ }^{119}$

Key intermediates in the preparation of benzimidazole-based ligands $\mathbf{1 5} \mathbf{- 3 2}$ are the N -substituted-4-amino-3-nitrobenzenesulfonamides $\mathbf{9 8 a}-\mathbf{h}$, synthesized as illustrated in Scheme 4.11.

Scheme 4.11 Synthesis of N-substituted-4-amino-3-nitrobenzenesulfonamides 98a - $\mathbf{h}$


Reagents and conditions: (a) Ethyl chlorooxoacetate, $\mathrm{Et}_{2} \mathrm{O}$, r. t., 18 h (98\%); (b) $\mathrm{ClSO}_{3} \mathrm{H}, 80$ ${ }^{\circ} \mathrm{C}, 3 \mathrm{~h}\left(99 \%\right.$ ); (c) (i) $\mathbf{9 7 a}$, pyridine, $0^{\circ} \mathrm{C}$ to r. t. (40\%); (ii) $\mathbf{9 7 b}-\mathbf{h}$, THF, r. t., $18 \mathrm{~h}(60-85 \%)$;

Briefly, treatment with ethyl chlorooxoacetate of the 2-nitroaniline 94 in diethyl ether $\left(\mathrm{Et}_{2} \mathrm{O}\right)$ furnished the protected amino compound $\mathbf{9 5}$ which reacted with the chlorosulfonic acid at 80 ${ }^{\circ} \mathrm{C}$. Aqueous workup gave the unprotected sulfonyl chloride $\mathbf{9 6}$ which reacted with appropriate amines (97a $-\mathbf{h})$ to afford the corresponding N -substituted-4-amino-3nitrobenzenesulfonamides $(\mathbf{9 8 a}-\mathbf{h})$. Preparation of $N$-aryl sulphonamide 98a in $40 \%$ yield required the use of pyridine as a solvent at $80^{\circ} \mathrm{C}$. On the other hand, preparation of $N$-alkyl sulfonamides $\mathbf{9 8 b} \mathbf{- h}$ in good yields ( $60-85 \%$ ) involved reaction between one equivalent of 96 and 4 equivalents of the appropriate ammine in THF at room temperature.

From nitro derivatives $\mathbf{9 8} \mathbf{a}-\mathbf{h}$, zinc dust reduction in acetic acid or palladium-catalysed hydrogenation furnishing the corresponding 3,4-diaminobenzenesulfonamides $\mathbf{9 9} \mathbf{a}$ and $\mathbf{9 9 b}$ h, respectively. Under oxidative conditions, the amino compounds were condensed with commercially available aldehydes $\mathbf{7 6 a}-\mathbf{h}$, in dry DMF at $80^{\circ} \mathrm{C}$ to afford the benzimidazolebased compounds $15-17,19-29$ and 100. Finally, deprotection under acidic conditions (DCM/TFA 1:1) of intermediate $\mathbf{1 0 0}$ furnished the primary sulfonamide $\mathbf{1 8}$ (Scheme 4.12).

Scheme 4.12 Synthesis of benzimidazole-based compounds 15 - 29



76a-h
a: $R_{1}, R_{5}=-H ; R_{2}, R_{4},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$
b: $\mathrm{R}_{1}, \mathrm{R}_{2}, \mathrm{R}_{4}, \mathrm{R}_{5},=-\mathrm{H} ; \mathrm{R}_{3}=-\mathrm{OH}$ c: $R_{1}, R_{3}, R_{4}, R_{5},=-H ; R_{2}=-O H$ d: $R_{1}, R_{2}, R_{4}, R_{5},=-H ; R_{3}=-C l$ e: $\mathrm{R}_{2}, \mathrm{R}_{4}, \mathrm{R}_{5}=-\mathrm{H} ; \mathrm{R}_{1}, \mathrm{R}_{3}=-\mathrm{OH}$ f: $R_{1}, R_{4}, R_{5},=-H ; R_{2}, R_{3}=-O H$ g: $\mathrm{R}_{2}, \mathrm{R}_{4}=-\mathrm{H} ; \mathrm{R}_{1}, \mathrm{R}_{5},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$ $h: R_{1}, R_{4}, R_{5},=-H ; R_{2},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$


15-29, 100

15 (EML765): $\mathrm{R}=-2$-pyridinyl; $\mathrm{R}_{1}, \mathrm{R}_{5}=-\mathrm{H} ; \mathrm{R}_{2}, \mathrm{R}_{4},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$ 16 (EML803): $R=-2$-picolyl; $R_{1}, R_{5}=-H ; R_{2}, R_{4},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$ 17 (EML798): $R=-3$-picolyl; $R_{1}, R_{5}=-H ; R_{2}, R_{4},=-\mathrm{CH}_{3} ; R_{3}=-\mathrm{OH}$

100: $R=$-tert-butyl; $R_{1}, R_{5}=-H ; R_{2}, R_{4},=-\mathrm{CH}_{3} ; R_{3}=-\mathrm{OH}$
18: $R=-H \cdot R_{1}, R_{5}=-H ; R_{2}, R_{4}=-\mathrm{CH}_{3} ; R_{3}=-O H$
19 (EML796): R = -methyl; $\mathrm{R}_{1}, \mathrm{R}_{5}=-\mathrm{H} ; \mathrm{R}_{2}, \mathrm{R}_{4},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$
20 (EML801): R = -cyclopropyl; $\mathrm{R}_{1}, \mathrm{R}_{5}=-\mathrm{H} ; \mathrm{R}_{2}, \mathrm{R}_{4},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$ 21 (EML797): $\mathrm{R}=$-cyclopentyl; $\mathrm{R}_{1}, \mathrm{R}_{5}=-\mathrm{H} ; \mathrm{R}_{2}, \mathrm{R}_{4},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$
 23 (EML760): $R_{1}, R_{2}, R_{4}, R_{5},=-H ; R_{3}=-O H$
4 (EML761): $R_{1}, R_{3}, R_{4}, R_{5}=-H ; R_{2}=-O H$
5 (EML762): $R_{1}, R_{2}, R_{4}, R_{5}=-H ; R_{3}=-C l$
26 (EML763): $R_{2}, R_{4}, R_{5},=-H ; R_{1}, R_{3}=-O H$
27 (EML764): $R_{1}, R_{4}, R_{5},=-H ; R_{2}, R_{3}=-O H$
28 (EML766): $R_{2}, R_{4}=-H ; R_{1}, R_{5},=-\mathrm{CH}_{3} ; \mathrm{R}_{3}=-\mathrm{OH}$
29 (EML806): $R_{1}, R_{4}, R_{5}=-H ; R_{2}=-\mathrm{CH}_{3} ; R_{3}=-\mathrm{OH}$

Reagents and conditions: (a) Zn dust, $\mathrm{AcOH}, 4 \mathrm{~h}(64 \%)$ or $\mathrm{H}_{2}, \mathrm{Pd} / \mathrm{C}, \mathrm{EtOH}, 18 \mathrm{~h}(89-97 \%)$; (b) $\mathrm{NaHSO}_{3}$, dry DMF, $80^{\circ} \mathrm{C}, 18 \mathrm{~h}(54-85 \%)$; (c)
DCM/TFA (1:1), r. t. 18 h (74\%).

Preparation of compounds $\mathbf{3 0}$ - $\mathbf{3 2}$ required the preparation of not commercially available aldehydes 103a - cas illustrated in Scheme 4.13. Briefly, aldehyde 76a is selectively brominated at position 2 by means of N -bromosuccinimide (NBS). The brominated aldehyde 101 easily undergoes Suzuki-Miyaura cross-coupling with boronic acids (102a-c) affording the aldehydes 103a - c. Following the synthetic procedures mentioned above, the benzimidazole-based ligands $\mathbf{3 0}$ - $\mathbf{3 2}$ were obtained in good yield (53-67\%) using as starting material the 3,4-diaminobenzenesulfonamide 99a.

Scheme 4.13 Synthesis of benzimidazole-based compounds 30-32


Reagents and conditions: (a) NBS, $\mathrm{H}_{2} \mathrm{SO}_{4}, 60^{\circ} \mathrm{C}, 4 \mathrm{~h}$ (20\%); (b) MW, TETRAKIS, $\mathrm{K}_{2} \mathrm{CO}_{3}$, $\mathrm{EtOH}, 80^{\circ} \mathrm{C}, 30 \mathrm{~min},(67-70 \%)$; (c) $\mathrm{NaHSO}_{3}$, dry DMF, $80^{\circ} \mathrm{C}$, $18 \mathrm{~h}(53-67 \%)$.

### 4.3 Synthesis of BET PROTACs

This Section describes the synthetic procedures adopted for the preparation BET PROTAC compounds 33-48.

Final compounds were obtained via peptide coupling (Scheme 4.14 and Scheme 4.15) between the carboxylic function of proper linkers attached on BD1/BD2 scaffold and the primary amino group of VHL ligand (104) and CRBN ligand (105).

Compounds $\mathbf{3 3}$ - $\mathbf{4 0}$ were obtained using PyOXIM as coupling reagent, DIPEA in anhydrous DMF. On the other hand, COMU was used as coupling reagent for the synthesis of compounds 41 - 48. As summarized in Table 4.2 and in Table 4.3, the yields are moderate.

VHL ligand (104) and CRBN ligand (105) were prepared according to literature. ${ }^{101,120}$ All the linkers and the functionalized BD1/BD2 ligands are prepared as described in the following Sections.

Scheme 4.14 Synthesis of BET PROTACs 33 - 40


Reagents and conditions: (a) PyOXIM, DIPEA, dry DMF, r. t. $30 \mathrm{~min}-1$ h.

Table 4.2 Compounds 33-40

| ENTRY LIGAND | Entry amine | LINKER (L1-L4) | FINAL COMPOUND | YIELD |
| :---: | :---: | :---: | :---: | :---: |
| 106 | 104 |  | 33 | 50\% |
| 107 | 104 |  | 34 | 32\% |
| 108 | 104 |  <br> L3 | 35 | 42\% |
| 109 | 104 |  | 36 | 32\% |
| 106 | 105 |  | 37 | 28\% |
| 107 | 105 |  | 38 | 27\% |
| 108 | 105 |  <br> L3 | 39 | 39\% |
| 109 | 105 |  | 40 | 38\% |

Scheme 4.15 Synthesis of BET PROTACs 41 - 48


Reagents and conditions: (a) COMU, DIPEA, dry DMF, r.t., $30 \mathrm{~min}-1 \mathrm{~h}$.

Table 4.3 Compounds 41-48

| ENTRY LIGAND | ENTRY AMINE | LINKER (L1-L4) | Final compound | YIELD |
| :---: | :---: | :---: | :---: | :---: |
| 110 | 104 |  | 41 | 46\% |
| 111 | 104 |  | 42 | 32\% |
| 112 | 104 |  <br> L3 | 43 | 36\% |
| 113 | 104 |  | 44 | 37\% |
| 110 | 105 |  | 45 | 37\% |
| 111 | 105 |  | 46 | 45\% |
| 112 | 105 |  <br> L3 | 47 | 27\% |
| 113 | 105 |  | 48 | 31\% |

### 4.3.1 Synthesis of PROTAC compounds 33-40

The synthesis of PROTACs $\mathbf{3 3}$ - $\mathbf{4 0}$ required the preparation of intermediates $\mathbf{1 0 6} \mathbf{- 1 0 9}$, obtained as illustrated in Scheme 4.16. These compounds containing a triple bond for proper linker connection. Under Sonogashira coupling conditions, the bromine analogue 114 reacted with the alkynyl linkers $\mathbf{1 1 5}$ - $\mathbf{1 1 8}$ opportune prepared, affording the intermediates $\mathbf{1 1 9 - 1 2 2}$. Straightforward hydrolysis of tert-butyl ester with trifluoroacetic acid in DCM yielded compounds 106-109.

Scheme 4.16 Synthesis of intermediates 106 - 109


Reagents and conditions: (a) $\mathrm{P}\left[\mathrm{P}\left(\mathrm{tBu} u_{3}\right)\right]_{2}, \mathrm{P}\left(\mathrm{tBu}_{3}\right), \mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{CuI}, 100{ }^{\circ} \mathrm{C}, 12 \mathrm{~h}(11-20 \%)$; (b) DCM/TFA 9:1, r. t , $30 \mathrm{~min}(95-96 \%)$.

The Sonogashira reaction conditions required an optimization work. Indeed, considering that this reaction is high context-dependent and could be complex on pyridine, especially using a PEG linker as substrate, different conditions were investigated using the bromine intermediate 114 and the alkynyl linker 115 as model substrate (Table 4.4).

Table 4.4 Different condition of Sonogashira tested. All reactions are set up on 10 mg and carried out at $100^{\circ} \mathrm{C}$. The reactions are monitored by LC-MS.


| $\#$ | $[P D]$ | LIGAND | $[\mathrm{Cu}]$ | BASE | SolVENT | CONVERSION |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{a}$ | $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$ | - | CuI | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | dry DMF | $10 \%$ |
| $\mathbf{b}$ | $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$ | - | CuI | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | dry ACN | No Product |
| $\mathbf{c}$ | $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$ | - | CuI | DIPEA | dry DMF | No Product |
| $\mathbf{d}$ | $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$ | - | CuI | TEA | dry ACN | No Product |
| $\mathbf{e}$ | $\mathrm{Pd}\left(\mathrm{OAc}_{2}\right.$ | $\mathrm{P}(\mathrm{tBu})_{3}$ | CuI | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | dry DMF | No Product |
| $\mathbf{f}$ | $\mathrm{Pd}\left[\mathrm{P}\left(\mathrm{tBu}_{3}\right)\right]_{2}$ | $\mathrm{P}(\mathrm{tBu})_{3}$ | CuI | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | dry ACN | $30 \%$ |
| $\mathbf{g}$ | $\mathrm{Pd}\left[\mathrm{P}(\mathrm{tBu})_{3}\right]_{2}$ | $\mathrm{P}(\mathrm{tBu})_{3}$ | CuI | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | dry DMF | $100 \%$ |

All the conditions differ for the palladium complex, the base and the solvent whereas in all reactions copper iodide is used (Table 4.4).

Firstly, the tetrakis(triphenylphosphine)palladium(0) complex was investigated (a-d), using different bases $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right.$, TEA and DIPEA) and solvents (dry DMF and dry ACN). Unfortunately, trace product was observed only using $\mathrm{K}_{2} \mathrm{CO}_{3}$ as base and dry DMF as solvent (a). Maintaining the base $\left(\mathrm{K}_{2} \mathrm{CO}_{3}\right)$, we decided to investigate different palladium complex: the bis(tri-tert-butylphosphine)palladium(0) and the palladium(II) acetate.

The main difference is the palladium form: palladium(II) acetate required the generation in situ of the reactive form ( Pd 0 ) using tri-tert-butylphosphine (e) whereas the bis(tri-tertbutylphosphine)palladium(0) complex has the palladium in the reactive form ( Pd 0$)(\mathbf{f} \mathbf{- g})$. Nevertheless, the bis(tri-tert-butylphosphine)palladium(0) is more unstable and can easily oxidize in palladium(II). To overcome this issue, tri-tert-butylphosphine was added (f-g).

Using palladium(II) acetate, we recovered only starting material (e). The use of bis(tri-tertbutylphosphine)palladium(0) gave good results. LC-MS analysis reveal the formation of $30 \%$ of product and the $70 \%$ of des-brominated starting material (f) using dry ACN as solvent. Surprisingly, a quantitative conversion of starting material was observed using dry DMF as solvent (g). The good results obtained prompted us to apply these conditions to all the alkynyl linkers 115 - $\mathbf{1 1 8}$ (Scheme 4.16). Unfortunately, despite the quantitative conversion of the starting material, we didn't recover a quantitative amount of products (119-122). As matter of fact, the insoluble palladium by-products and the low solubility of compounds complicated purification and negatively impacted the yield (11-20\%).

### 4.3.1.2 Synthesis of bromine derivative $\mathbf{1 1 4}$

Preparation of bromine analogue of the benzimidazole-based ligand was carried out according to Scheme 4.17 , exploiting the same synthetic strategies previously described (Section 4.2).

Scheme 4.17 Synthesis of bromine analogue 114


Reagents and conditions: (a) 2-amino-5-bromopyridine 97i, dry pyridine, r. t., 18 h (23\%); (b) Zn dust, acetic acid, r. t., 18 h (98\%); (c) 76a, $\mathrm{NaHSO}_{3}$, dry DMF, $100^{\circ} \mathrm{C}, 18 \mathrm{~h}(56 \%)$.

Briefly, the sulphonyl chloride 96, synthesized as previously described (Scheme 4.11), reacted with the commercially available 2 -amino-5-bromopyridine $97 \mathbf{i}$ affording the corresponding N -substituted-4-amino-3-nitrobenzenesulfonamide (123). According to the synthesis of 98a, the reaction featured a low yield (23\%) due to solubility problems which negatively affected purification. Any attempt to optimize this reaction changing solvents (pyridine or $\mathrm{N}, \mathrm{N}$-dimethylacetamide), bases (pyridine and triethylamine) and temperatures ( 0 ${ }^{\circ} \mathrm{C}$ or room temperature) were unsuccessful. The reduction of the nitro group with zinc dust in acetic acid furnished the corresponding amino group (124) which reacts with the commercially available 4-hydroxy-3,5dimethylbenzaldehyde 76a, under oxidative condition, affording the bromine analogue 114.

### 4.3.1.2 Synthesis of alkynyl linkers 115 - 118

The four linkers chosen (L1 - L4) were properly functionalized with the alkynyl group, suitable for the subsequent Sonogashira reaction, and with a carboxylic function, required for the final coupling reaction.

The triethylene glycol 70c, commercially available, reacted with the propargyl bromide in presence of potassium tert-butoxide (tBuOK) to give the alkynyl linker 125. The latter reacted with tert-butyl 2-bromoacetate in presence of sodium hydride $(\mathrm{NaH})$ to afford the proper functionalized PEG (L1) linker 115 (Scheme 4.18).

Scheme 4.18 Synthesis of alkynyl linker 115


Reagents and conditions: (a) propargyl bromide, tBuOK, dry THF, r. t., 18 h (80\%); (b) tertbutyl 2-bromoacetate, NaH ( $60 \%$ mineral oil), dry THF, r. t., 18 h ( $50 \%$ ).

The same synthetic conditions were used to obtain the functionalized aliphatic L2 linker (Scheme 4.19). The commercially available 1,6-hexandiol (126) react with propargyl bromide in presence of sodium hydride affording the alkynyl aliphatic linker $\mathbf{1 2 7}$ in good yield (76\%). Unfortunately, the subsequent reaction with tert-butyl 2-bromoacetate on the alkynyl aliphatic linker failed, both using anhydrous DMF and anhydrous THF as solvents. The same negative result was obtained treating, in the first instance, the 1,6-hexandiol (126) with tert-butyl 2bromoacetate, affording the intermediate 128, which unsuccessfully reacted with propargyl bromide. In all the conditions, we recovered starting materials and byproducts, probably due to rearrangement at alkynyl position.

Finally, using tert-butyl diazoacetate in a rhodium-catalyzed reaction, we were able to obtain the desired product 116. This reaction involved the formation of a metal-carbene complex intermediate which, probably, proceeds better on this linker.

Scheme 4.19 Synthesis of alkynyl linker 116


Reagents and conditions: (a) propargyl bromide, NaH, dry DMF, r. t., 18 h (76\%); (b) tert-butyl 2-bromoacetate, NaH (60\% mineral oil), dry DMF or dry THF, r. t., 18 h (40\%); (c) tert-butyl diazoacetate, rhodium acetate, DCM, r. t., 18 h ( $66 \%$ ).

The synthesis of functionalized linker L3 was carried out treating the 4-bromomethylbenzyl alcohol 129, commercially available, with tert-butyl diazoacetate in presence of rhodiumacetate affording the intermediate $\mathbf{1 3 0}$ (Scheme 4.20).

Under nucleophilic substitution condition, the latter reacted with propargyl alcohol in presence of sodium hydride affording the proper functionalized alkynyl linker 117.

Scheme 4.20 Synthesis of alkynyl linker 117


Reagents and conditions: (a) tert-Butyl diazoacetate, rhodium acetate, DCM, r. t., 18 h (67\%); (b) propargyl alcohol, NaH ( $60 \%$ mineral oil), dry DMF, r. t., 18 h ( $60 \%$ ).

Finally, the akynyl L4 linker is prepared as shown in Scheme 4.21. Briefly, the benzyl 4bromobutyl ether (131) reacted with the 1,4-butanediol (132) in presence of sodium hydride, furnishing the intermediate $\mathbf{1 3 3}$ in good yield ( $84 \%$ ). The resulting intermediate $\mathbf{1 3 3}$ reacted with tert-butyl diazoacetate, in presence of rhodium acetate, forming the intermediate 134 which undergoes a palladium-catalysed hydrogenation affording the O-debenzylated compound 135. Avoiding the issue occurred in the synthesis of linker 116, a different strategy to insert the alkynyl group was used. The primary alcoholic function of $\mathbf{1 3 5}$ was converted into iodide in presence of triphenylphosphine, iodine and catalytic amount of imidazole by Appel reaction, affording the intermediate $\mathbf{1 3 6}$. The latter reacted with propargyl alcohol in presence of sodium hydride affording the alkynyl linker 118.

Scheme 4.21 Synthesis of alkynyl linker 118


Reagents and conditions: (a) NaH (60\% mineral oil), dry DMF, r. t., 18 h (84\%); (b) tert-butyl diazoacetate, rhodium acetate, DCM, r. t., 18 h (75\%); (c) $\mathrm{H}_{2}, \mathrm{Pd} / \mathrm{C} 10 \%$, methanol, r. t., 18 h $(96 \%)(d) \mathrm{I}_{2}, \mathrm{Ph}_{3} \mathrm{P}$, imidazole, dry DCM, $0{ }^{\circ} \mathrm{C}$ to r. t., $3 \mathrm{~h}(56 \%)$; (e) propargyl alcohol, NaH (60\% mineral oil), dry DMF, r. t., 18 h (30\%).

### 4.3.2 Synthesis of PROTAC compounds 41 - 48

The synthesis of PROTAC compounds $\mathbf{4 1}$ - $\mathbf{4 8}$ involved the preparation of intermediates $\mathbf{1 1 0}$ - $\mathbf{1 1 3}$ as depicted in the Scheme 4.22.

Scheme 4.22 Synthesis of intermediates $\mathbf{1 1 0}$ - $\mathbf{1 1 3}$




139


140



110




Reagents and conditions: (a) $\mathrm{NaHSO}_{3}$, dry DMF, $100^{\circ} \mathrm{C}$, 18 h ( $68-86 \%$ ); (b) DCM/TFA 9:1, r. t. $1 \mathrm{~h}(96-97 \%$ ).

According to the synthetic strategies previously optimized (Section 4.1), we decided to build the quinazolinone scaffold on the proper linkers featuring a carboxylic function for the final coupling reaction. The opportune prepared aldehydes ( $\mathbf{1 3 7} \mathbf{- 1 4 0}$ ) reacted with 2-amino-4,6 dimethoxybenzamide 64 by cyclocondensation-dehydrogenation in presence of $\mathrm{NaHSO}_{3}$ to afford the quinazolinone-based ligands ( $\mathbf{1 4 1} \mathbf{- 1 4 4}$ ). Unlike the synthesis of intermediates 65, 66a - $\mathbf{f}$ and $\mathbf{6 7 a , b}$ (Scheme 4.3), we did not use a catalytic amount of p -toluenesulfonic acid to avoid acid conditions which could interfere with the acid-labile tert-butyl ester. Moreover, anhydrous DMF instead of anhydrous DMAc was used and, consequently, a lower temperature $\left(100{ }^{\circ} \mathrm{C}\right)$. Finally, hydrolysis in acidic conditions furnished the carboxylic derivatives $\mathbf{1 1 0}$ 113.

The aldehydes $\mathbf{1 3 7}$ - $\mathbf{1 4 0}$ were prepared as described in Scheme 4.23.
The first step required the activation of the four linkers (L1 - L4) with a proper leaving group (-OTos or -Br ). The triethylene glycol 70c, commercially available, reacted with tert-butyl 2bromoacetate in presence of sodium hydride to give the intermediate $\mathbf{1 4 5}$ in good yield (60\%). Protected linkers $\mathbf{1 2 8}$ and 135, synthesized as previously reported, and $\mathbf{1 4 5}$ were activated in the subsequent nucleophilic substitutions using tosyl chloride and triethylamine in anhydrous DCM affording the tosylated intermediates 146, 147 and 149. The tosylated intermediates 146, 147 and 149 and the bromine intermediate 148, synthesized as previously described (Scheme 4.19), easily reacted with the 4-hydroxy-3,5-dimethylbenzaldehyde 76a, commercially available, under nucleophilic substitution conditions, affording the corresponding aldehyde intermediates $\mathbf{1 3 7} \mathbf{- 1 4 0}$ in good yield ( $70-96 \%$ ).

Scheme 4.23 Synthesis of intermediates 137 - 140


70c


145


146


128



130


135
148



137



139


140

Reagents and conditions: (a) tert-butyl 2-bromoacetate, NaH ( $60 \%$ mineral oil), dry DMF, r. t., 18 h ( $60 \%$ ); (b) TsCl, TEA, dry DCM, r. t., 18 h ( 60 $64 \%$ ); (c) $\mathrm{K}_{2} \mathrm{CO}_{3}$, dry DMF, $80^{\circ} \mathrm{C}$, $18 \mathrm{~h}(70-96 \%)$.

## CHAPTER V

## CONCLUSIONS

This PhD project reported the design, synthesis and biological evaluation of new ligands for the BET family of bromodomains, with attention on selectivity between BET protein isoforms and/or bromodomain modules. Selective ligands could offer valuable chemical probes for these epigenetic reader proteins and putative therapeutic agents. To this aim, three different approaches were applied.

Bisubstrate approach have been exploited to design compounds potentially able to bind the first and the second bromodomain of BET proteins in order to obtain structural information. The compounds were designed connecting a BD1 (Cl-BzT-7) and a BD2 (RVX-208) / selective ligand with a proper linker. A small set of compounds have been designed and synthesized, changing the length and the nature of linkers. DSF assay and sedimentation velocity assay showed that compound EML896 is able to partially engage both BD1 and BD2 domain. These promising results are a valuable starting point for the development of new bivalent chemical probes.

To develop selective ligands for the first bromodomain of BETs, diazobenzene-based compounds were selected as lead compounds. In order to improve metabolic stability and pharmacokinetic properties, the diazobenzene core was rigidized into a benzimidazole scaffold and a small library of compounds was designed and synthesized. DSF, ITC and TR-FRET assays allowed the identification of compound EML765 as potent BET ligand ( $\mathrm{K}_{\mathrm{D}}$ of 70 nM ) with a good selectivity for the first bromodomain of BETs (170-fold for BD1of BRD2). Taken together, the data endorse compound EML765 as promising BD1 selective ligand. Metabolic stability and pharmacokinetic properties are currently being assessed.

Finally, BET PROTACs with BD1 and BD2 selective ligands were designed. EML765 and RVX-208 were chosen as BD1 and BD2 warhead, respectively, and connected with different type of linkers to VHL-based and CRBN-based E3 ligase ligands. Compounds were synthesized and their activity was evaluated in cellular assays. Unfortunately, compounds are not able to induce protein degradation. These results need to be considered in the perspective of the
complexity of protein-induced degradation mechanism and of the multifaceted aspects involved in the PROTACs design.

## CHAPTER VI

## MATERIALS AND METHODS

### 7.1 General information

All chemicals were purchased from Sigma-Aldrich Srl (Milan, Italy) or from Fluorochem Ltd. (Hadfield, UK) and were of the highest purity. All solvents were of reagent grade and, when necessary, were purified and dried by standard methods. All reactions requiring anhydrous conditions were conducted under a positive atmosphere of nitrogen in oven-dried glassware. Standard syringe techniques were used for anhydrous addition of liquids. Reactions were routinely monitored by thin-layer chromatography (TLC) performed on aluminum-backed silica gel plates (Merck DC, Alufolien Kieselgel 60 F254) with spots visualized by UV light ( $\lambda$ $=254,365 \mathrm{~nm}$ ) or using a $\mathrm{KMnO}_{4}$ alkaline solution. Solvents were removed using a rotary evaporator operating at a reduced pressure of $\sim 10$ Torr. Organic solutions were dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Chromatographic purification was done on an automated flash chromatography system (Isolera Dalton 2000, Biotage) using cartridges packed with KP-SIL, $60 \AA(40-63 \mu \mathrm{~m}$ particle size). All microwave-assisted reactions were conducted in a CEM Discover SP microwave synthesizer equipped with vertically focused IR temperature sensor. Analytical high-performance liquid chromatography (HPLC) was performed on a Shimadzu SPD 20A UV/vis detector ( $\lambda=220$ and 254 nm ) using C-18 column Phenomenex Synergi Fusion-RP $80 \mathrm{~A}\left(75 \times 4.60 \mathrm{~mm}^{2} ; 4 \mu \mathrm{~m}\right)$ at $25^{\circ} \mathrm{C}$ using mobile phases A (water $+0.1 \%$ TFA) and $B$ (acetonitrile $(A C N)+0.1 \% \mathrm{TFA})$ at a flow rate of $1 \mathrm{~mL} / \mathrm{min}$. Preparative HPLC was performed using a Shimadzu Prominence LC20AP with the UV detector set to 220 and 254 nm . Samples were injected onto a Phenomenex Synergi Fusion-RP $80 \mathrm{~A}\left(150 \times 21 \mathrm{~mm}^{2} ; 4 \mu \mathrm{~m}\right) \mathrm{C}$ 18 column at room temperature. Mobile phases of A (water $+0.1 \%$ TFA) and $\mathrm{B}(\mathrm{ACN}+0.1 \%$ TFA) were used with a flow rate of $20 \mathrm{~mL} / \mathrm{min}$.
${ }^{1} \mathrm{H}$ spectra were recorded at 400 MHz on a Bruker Ascend 400 spectrometer, while ${ }^{13} \mathrm{C}$ NMR spectra were obtained by distortionless enhancement by polarization transfer quaternary
spectroscopy on the same spectrometer. Chemical shifts are reported in $\delta(\mathrm{ppm})$ relative to the internal reference tetramethylsilane. Due to the existence of tautomers, some ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR signals could not be detected for some of the prepared benzimidazoles so only the distinct signals are reported. Low-resolution mass spectra were recorded on a Finnigan LCQ DECA TermoQuest mass spectrometer in electrospray positive and negative ionization modes (ESIMS). VHL ligand $\mathbf{1 0 4}$ and CRBN ligand 105 were prepared in according with the reported procedure and all the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra are consistent with those already reported to literature. ${ }^{90,101}$ Only the biological experiments done by myself are reported.

### 7.2 Synthetic procedures

Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4] triazepin -4-yl)-N-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy) ethyl)acetamide (1, EML807)


To a stirred solution of intermediate $49(22 \mathrm{mg}, 0.05 \mathrm{mmol})$ and amine derivative $50(22 \mathrm{mg}$, 0.06 mmol ) in 1 mL of dry DMF, EDC hydrochloride ( $17 \mathrm{mg}, 0.085 \mathrm{mmol}$ ), HOBt ( 13 mg , $0.085 \mathrm{mmol})$ and 4 -methylmorpholine ( $0.024 \mathrm{~mL}, 0.22 \mathrm{mmol}$ ) were added. The reaction was stirred at room temperature for 18 h . Then, water $(15 \mathrm{~mL})$ was added the resulting mixture was extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ). The combined organic phases were washed with saturated solution of $\mathrm{NaHCO}_{3}(3 \times 10 \mathrm{~mL})$, brine ( 15 mL ), dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and
evaporated under reduced pressure. The crude was purified by silica gel chromatography (DCM-MeOH 97:3 to 90:10) yielding the title compound as white solid ( $25 \mathrm{mg}, 66 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.65(\mathrm{~s}, 1 \mathrm{H}), 7.65(\mathrm{~s}, 2 \mathrm{H}), 7.57(\mathrm{dd}, \mathrm{J}=8.6,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.48$ $(\mathrm{d}, \mathrm{J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.29(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.20(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $7.21-7.15(\mathrm{~m}, 1 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.64-4.57(\mathrm{~m}, 2 \mathrm{H})$, $3.96(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.90-3.85(\mathrm{~m}, 2 \mathrm{H}), 3.73-3.64(\mathrm{~m}, 2 \mathrm{H}), 2.60(\mathrm{~s}, 3 \mathrm{H}), 2.18$ (s, 6H). ${ }^{13}{ }^{13}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 168.41,165.16,161.43,160.76,160.61,159.32,158.38,153.82$, $152.12,148.47,137.35,133.65,133.38,133.07,132.28,131.61,131.22,130.44,129.56$, 128.97, 127.96, 127.71, 123.95, 101.39, 98.26, 70.62, 57.61, 56.40, 55.74, 39.72, 16.43, 12.66. MS (ESI) $m / z: 753(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4] triazepin-4-yl)-N-(8-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy) octyl)acetamide (2, EML901)


Compound 2 (EML901) was obtained as a light yellow solid ( $35 \mathrm{mg}, 60 \%$ ) from derivative 49 $(28 \mathrm{mg}, 0.07 \mathrm{mmol})$ and $\mathbf{5 1 a}(35 \mathrm{mg}, 0.077 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.40(\mathrm{~s}, 1 \mathrm{H}), 7.68(\mathrm{~s}, 2 \mathrm{H}), 7.64(\mathrm{dd}, \mathrm{J}=8.7,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.49$ $(\mathrm{d}, \mathrm{J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.32(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.27(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $6.82(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{t}, \mathrm{J}=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.54-4.49(\mathrm{~m}, 2 \mathrm{H})$, 3.97 (s, 3H), $3.93(\mathrm{~s}, 3 \mathrm{H}), 3.78(\mathrm{t}, \mathrm{J}=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.37-3.17(\mathrm{~m}, 2 \mathrm{H}), 2.60(\mathrm{~s}, 3 \mathrm{H}), 2.35(\mathrm{~s}$, $6 \mathrm{H}), 1.85-1.72(\mathrm{~m}, 4 \mathrm{H}), 1.51-1.35(\mathrm{~m}, 8 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 167.94, 165.14,
$161.43,160.53,159.39,153.90,152.26,148.38,137.32,133.70,133.33,133.20,132.31$, $132.11,131.10,130.46,129.64,128.98,127.46,124.06,101.39,98.18,72.49,57.65,56.39$, $55.73,39.26,30.35,29.71,29.50,29.40,29.20,26.76,26.05,16.54,12.68$. MS (ESI) $m / z: 837$ $(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4]

 triazepin-4-yl)-N-(2-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy)ethoxy)ethyl)acetamide (3, EML896)

Compound 3 (EML896) was obtained as a white solid ( $47 \mathrm{mg}, 60 \%$ ) from derivative 49 (40 $\mathbf{m g}, 0.1 \mathrm{mmol})$ and $\mathbf{5 1 b}(46 \mathrm{mg}, 0.11 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.62(\mathrm{~s}, 1 \mathrm{H}), 7.68(\mathrm{~s}, 2 \mathrm{H}), 7.57(\mathrm{dd}, \mathrm{J}=8.7,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.47$ $(\mathrm{d}, \mathrm{J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.28(\mathrm{~d}, \mathrm{~J}=5.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $6.80(\mathrm{~d}, \mathrm{~J}=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.60-4.50(\mathrm{~m}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}$, $3 \mathrm{H}), 3.89-3.85(\mathrm{~m}, 2 \mathrm{H}), 3.74-3.67(\mathrm{~m}, 2 \mathrm{H}), 3.61(\mathrm{t}, \mathrm{J}=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.57(\mathrm{~s}, 3 \mathrm{H}), 2.30(\mathrm{~s}$, $6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 168.13,165.15,161.44,160.57,160.49,159.28,158.81$, 153.86, 152.20, 148.43, 137.24, 133.78, 133.22, 133.14, 132.26, 131.93, 131.19, 130.43, 129.57, 128.93, 127.83, 127.62, 124.00, 105.02, 101.40, 98.22, 71.42, 70.26, 69.91, 57.55, 56.39, 55.73, 39.31, 16.47, 12.66. MS (ESI) $m / z: 797(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4] triazepin-4-yl)-N-(2-(2-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy)ethoxy)ethoxy)ethyl)acetamide (4, EML730)


Compound 4 (EML730) was obtained as a white solid ( 45 mg , 54\%) from derivative 49 ( 40 $\mathrm{mg}, 0.1 \mathrm{mmol})$ and $\mathbf{5 1 c}(50 \mathrm{mg}, 0.11 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.74(\mathrm{~s}, 1 \mathrm{H}), 7.67(\mathrm{~s}, 2 \mathrm{H}), 7.61(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{~d}, J=$ $8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.37(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.28(\mathrm{~m}, 1 \mathrm{H}), 7.25-7.25(\mathrm{~m}, 1 \mathrm{H}), 6.83(\mathrm{~s}, 1 \mathrm{H}), 6.76(\mathrm{~s}$, $1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.70-4.37(\mathrm{~m}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.82-3.76(\mathrm{~m}$, 2H), $3.67-3.43(\mathrm{~m}, 10 \mathrm{H}), 2.61(\mathrm{~s}, 3 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 167.97, $165.10,161.43,160.28,137.15,133.93,133.20,132.23,131.99,131.25,130.49,129.65$, 128.92, 127.77, 124.06, 101.33, 98.19, 77.34, 77.02, 76.70, 71.63, 70.70, 70.42, 70.28, 69.83, 57.39, 57.39, 56.41, 39.15, 16.52, 12.73. MS (ESI) $m / z: 841(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4] triazepin-4-yl)-N-(2-(2-(2-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy)ethoxy)ethoxy)ethoxy)ethyl)acetamide (5, EML897)


Compound 5 (EML897) was obtained as a white solid ( 53 mg , $60 \%$ ) from derivative 49 (40 $\mathrm{mg}, 0.1 \mathrm{mmol}$ ) and 51d ( $55 \mathrm{mg}, 0.11 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.79(\mathrm{~s}, 1 \mathrm{H}), 7.68(\mathrm{~s}, 2 \mathrm{H}), 7.60(\mathrm{dd}, \mathrm{J}=8.7,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.46$ (d, J = $8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.37(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.30(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $6.95(\mathrm{t}, \mathrm{J}=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.56-4.47(\mathrm{~m}, 2 \mathrm{H})$, $4.02-3.96(\mathrm{~m}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.85-3.79(\mathrm{~m}, 2 \mathrm{H}), 3.73-3.69(\mathrm{~m}, 2 \mathrm{H}), 3.66$ $-3.62(\mathrm{~m}, 2 \mathrm{H}), 3.58-3.45(\mathrm{~m}, 8 \mathrm{H}), 2.61(\mathrm{~s}, 3 \mathrm{H}), 2.37(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta$ 168.01, 165.07, 161.43, 160.54, 160.27, 159.28, 158.91, 153.86, 152.53, 148.49, 137.10, 133.97, 133.18, 133.15, 132.19, 131.99, 131.23, 130.49, 129.65, 128.90, 127.72, 105.07, 101.34, 98.15, 77.34, 71.66, 70.84, 70.63, 70.57, 70.51, 70.25, 69.82, 57.29, 56.39, 55.72, 39.18, 16.53, 12.67. MS (ESI) $m / z: 885(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4]

triazepin-4-yl)-N-(14-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy)-3,6,9,12-tetraoxatetradecyl)acetamide (6, EML731)


Compound 6 (EML731) was obtained as a white solid ( $60 \mathrm{mg}, 48 \%$ ) from derivative 49 (55 $\mathbf{m g}, 0.135 \mathrm{mmol})$ and $\mathbf{5 1 e}(81 \mathrm{mg}, 0.149 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.71(\mathrm{~s}, 1 \mathrm{H}), 7.68(\mathrm{~s}, 2 \mathrm{H}), 7.59(\mathrm{dd}, \mathrm{J}=8.7,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.46$ (d, J = 8.6 Hz, 2H), $7.37(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.30(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $6.92(\mathrm{t}, \mathrm{J}=5.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.44(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.60-4.46(\mathrm{~m}, 2 \mathrm{H})$, 3.99 (dd, J = 5.7, 3.7 Hz, 2H), 3.96 (s, 3H), 3.92 (s, 3H), 3.83 (dd, J = 5.7, 3.7 Hz, 2H), 3.72
(dd, J = 5.7, 3.2 Hz, 2H), 3.67 (dd, J = 6.3, 3.7 Hz, 2H), 3.63 (dd, J = 5.7, 3.7 Hz, 2H), 3.59 (dd, $\mathrm{J}=6.3,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.53(\mathrm{~s}, 8 \mathrm{H}), 2.59(\mathrm{~s}, 3 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $168.05,165.09,163.35,161.43,160.54,160.29,159.29,159.00,153.86,152.42,148.37$, 137.11, 133.96, 133.17, 132.20, 132.03, 131.23, 130.49, 129.62, 128.91, 127.62, 124.02, 105.06, 101.34, 98.15, 71.67, 70.91, 70.65, 70.56, 70.47, 70.22, 69.80, 57.36, 56.38, 55.72, 39.18, 16.52, 12.67. MS (ESI) $m / z: 929(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4] triazepin-4-yl)-N-(17-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy)-3,6,9,12,15-pentaoxaheptadecyl)acetamide (7, EML898)


Compound 7 (EML898) was obtained as a white solid ( 20 mg , 49\%) from derivative 49 (17 $\mathrm{mg}, 0.042 \mathrm{mmol})$ and $\mathbf{5 1 f}(25 \mathrm{mg}, 0.042 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.85(\mathrm{~s}, 1 \mathrm{H}), 7.71(\mathrm{~s}, 2 \mathrm{H}), 7.60(\mathrm{dd}, \mathrm{J}=8.7,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.46$ (d, J = 8.6 Hz, 2H), $7.37(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.30(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $6.90(\mathrm{t}, \mathrm{J}=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.44(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.59-4.48(\mathrm{~m}, 2 \mathrm{H})$, $3.99(\mathrm{dd}, \mathrm{J}=5.8,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.82(\mathrm{dd}, \mathrm{J}=5.8,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.74-$ $3.55(\mathrm{~m}, 20 \mathrm{H}), 2.58(\mathrm{~s}, 3 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 168.03, 165.09, $161.43,160.69,160.31,159.29,159.03,153.89,152.43,148.36,137.11,133.95,133.16$, $132.22,132.00,131.22,130.49,128.91,127.66,124.04,105.03,101.33,98.15,71.67,70.90$, $70.67,70.60,70.53,70.46,70.44,70.20,69.78,57.37,56.37,55.72,39.17,29.70,16.51,12.66$. MS (ESI) $m / z: 973(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4] triazepin-4-yl)-N-(4-(2-(2-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy)ethoxy)ethoxy)ethoxy)phenyl)acetamide (8, EML729)


Compound 8 (EML729) was obtained as a white solid ( $35 \mathrm{mg}, 74 \%$ ) from derivative 49 (20 $\mathrm{mg}, 0.051 \mathrm{mmol}$ ) and $\mathbf{5 2 a}(31 \mathrm{mg}, 0.056 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 10.08(\mathrm{~s}, 1 \mathrm{H}), 8.87(\mathrm{~s}, 1 \mathrm{H}), 7.71(\mathrm{~s}, 2 \mathrm{H}), 7.62(\mathrm{dd}, \mathrm{J}=8.7,2.4$ $\mathrm{Hz}, 1 \mathrm{H}), 7.47(\mathrm{~d}, \mathrm{~J}=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, \mathrm{~J}=7.9 \mathrm{~Hz}, 3 \mathrm{H}), 7.31(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.26(\mathrm{~d}, \mathrm{~J}$ $=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.83-6.80(\mathrm{~m}, 2 \mathrm{H}), 6.43(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.64(\mathrm{~s}, 2 \mathrm{H}), 4.09(\mathrm{t}, \mathrm{J}=4.9 \mathrm{~Hz}$, 2H), 3.97 (d, J = $4.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), $3.94(\mathrm{~s}, 3 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H}), 3.84(\mathrm{t}, \mathrm{J}=4.9 \mathrm{~Hz}, 4 \mathrm{H}), 3.77-3.69$ (m, 4H), $2.59(\mathrm{~s}, 3 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta$ 166.26, 165.07, 161.39, $160.88,160.76,159.43,158.98,155.59,153.92,152.53,148.47,137.28,133.69,133.41$, $133.07,132.34,131.96,131.25,131.08,130.51,129.65,128.96,127.75,124.09,121.61$, 114.96, 105.02, 101.33, 98.12, 71.62, 70.95, 70.50, 69.83, 67.79, 65.85, 58.31, 56.33, 55.72, 16.46, 12.65. MS (ESI) $m / z: 933(\mathrm{M}+\mathrm{H})^{+}$. 3,6,9,12-tetraoxatetradecyl)oxy)phenyl)acetamide (9, EML742)


Compound 9 (EML742) was obtained as a white solid ( 64 mg , 52\%) from derivative 49 (49 $\mathrm{mg}, 0.121 \mathrm{mmol})$ and $\mathbf{5 2 b}(85 \mathrm{mg}, 0.133 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.70(\mathrm{~s}, 1 \mathrm{H}), 8.79(\mathrm{~s}, 1 \mathrm{H}), 7.69(\mathrm{~s}, 2 \mathrm{H}), 7.63(\mathrm{dd}, \mathrm{J}=8.7,2.4 \mathrm{~Hz}$, $1 \mathrm{H}), 7.48(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.35-7.29(\mathrm{~m}, 2 \mathrm{H}), 7.28(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}$, $1 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=2.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.79(\mathrm{~s}, 1 \mathrm{H}), 6.44(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.63(\mathrm{~s}, 2 \mathrm{H}), 4.06(\mathrm{dd}, \mathrm{J}=$ $5.8,3.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.98(\mathrm{~d}, \mathrm{~J}=4.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.86-3.78(\mathrm{~m}, 4 \mathrm{H}), 3.75$ $-3.64(\mathrm{~m}, 12 \mathrm{H}), 2.60(\mathrm{~s}, 3 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 166.19, 165.09, $161.41,160.80,160.60,159.44,159.04,155.60,153.88,152.40,148.47,137.31,133.67$, 133.44, 133.08, 132.35, 132.04, 131.27, 131.03, 130.52, 129.66, 128.97, 127.63, 124.07, $121.52,114.93,101.35,98.14,71.66,70.92,70.85,70.69,70.65,70.47,69.74,67.76,61.49$, 58.37, 56.37, 55.72, 16.50, 12.67. MS (ESI) $m / z: 1022(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4] triazepin-4-yl)-N-(2-(4-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy)ethyl)-1H-1,2,3-triazol-1-yl)ethyl)acetamide (10, EML823)


Compound 10 (EML823) was obtained as a white solid ( $45 \mathrm{mg}, 78 \%$ ) from derivative 49 (27 $\mathrm{mg}, 0.068 \mathrm{mmol})$ and $\mathbf{5 3} \mathbf{a}(32 \mathrm{mg}, 0.075 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.69$ (s, 1H), 7.67 (s, 2H), $7.62(\mathrm{dd}, \mathrm{J}=8.6,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.49$ $(\mathrm{s}, 1 \mathrm{H}), 7.44(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.39(\mathrm{~d}, \mathrm{~J}=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.12(\mathrm{t}, \mathrm{J}=$ $5.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.60(\mathrm{~s}, 2 \mathrm{H}), 4.44(\mathrm{t}, \mathrm{J}=5.6 \mathrm{~Hz}$, $2 \mathrm{H}), 4.04(\mathrm{t}, \mathrm{J}=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.85-3.72(\mathrm{~m}, 2 \mathrm{H}), 3.13(\mathrm{t}, \mathrm{J}=6.6 \mathrm{~Hz}$, 2H), 2.58 ( $\mathrm{s}, 3 \mathrm{H}$ ), $2.24(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 168.82, 165.13, 161.42, 160.81, $160.60,159.10,158.85,153.85,152.27,148.62,144.27,137.20,133.87,133.37,132.93$, $132.38,131.90,131.70,130.46,129.33,128.91,127.77,124.17,122.86,105.01,101.37,98.21$, $70.92,57.47,56.38,55.73,49.30,39.27,26.98,16.50,12.65 . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / z: 848(\mathrm{M}+\mathrm{H})^{+}$ triazepin-4-yl)-N-(2-(4-(4-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy)butyl)-1H-1,2,3-triazol-1-yl)ethyl)acetamide (11, EML899)


Compound 11 (EML899) was obtained as a white solid ( 37 mg , $57 \%$ ) from derivative 49 (30 $\mathrm{mg}, 0.074 \mathrm{mmol})$ and $\mathbf{5 3 b}(40 \mathrm{mg}, 0.081 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$. ${ }^{1}{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.58(\mathrm{~s}, 1 \mathrm{H}), 7.69(\mathrm{~s}, 2 \mathrm{H}), 7.64(\mathrm{dd}, \mathrm{J}=8.6,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.43$ $(\mathrm{d}, \mathrm{J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.40(\mathrm{~d}, \mathrm{~J}=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.36(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.29(\mathrm{~s}, 1 \mathrm{H}), 7.10(\mathrm{t}, \mathrm{J}=$ $5.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.68-4.45(\mathrm{~m}, 2 \mathrm{H}), 4.40(\mathrm{t}, \mathrm{J}=$ $5.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.82(\mathrm{~d}, \mathrm{~J}=5.0 \mathrm{~Hz}, 4 \mathrm{H}), 2.73-2.65(\mathrm{~m}, 2 \mathrm{H}), 2.59(\mathrm{~s}$, $3 \mathrm{H}), 2.34(\mathrm{~s}, 6 \mathrm{H}), 1.90-1.81(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 168.80,165.14,161.43$, $160.85,160.58,159.25,159.10,153.89,152.29,148.59,147.72,137.19,133.89,133.38$, $132.95,132.40,132.02,131.71,130.45,129.32,128.91,127.57,124.12,121.57,105.00$, $101.38,98.19,72.04,57.47,56.39,55.73,49.24,39.13,29.96,29.70,25.95,25.41,16.56$, 12.67. MS (ESI) $m / z: 876(\mathrm{M}+\mathrm{H})^{+}$. phenoxy)ethyl)-1H-1,2,3-triazol-1-yl)ethoxy)ethoxy)ethyl)acetamide (12, EML809)


A solution of compound $\mathbf{8 7}(30 \mathrm{mg}, 0.054 \mathrm{mmol})$ in 0.050 mL of dry DCM was added to a solution of $\mathrm{CuI}(1 \mathrm{mg}, 0.0012 \mathrm{mmol})$, DIPEA ( $0.0036 \mathrm{~mL}, 0.0021 \mathrm{mmol}$ ) and AcOH ( 0.0012 $\mathrm{mL}, 0.0021 \mathrm{mmol}$ ) in 0.015 mL of dry DCM. After $5 \mathrm{~min}, \mathbf{8 5 a}$ ( $20 \mathrm{mg}, 0.053 \mathrm{mmol}$ ) was added and the resulting mixture was stirred at room temperature for 18 h . Then, the solvent was evaporated and the crude was purified by silica gel chromatography (DCM/MeOH 95:5 to 80:20) to give the title compound $\mathbf{1 2}$ as white solid ( $34 \mathrm{mg}, 69 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.66(\mathrm{~s}, 2 \mathrm{H}), 7.64(\mathrm{~s}, 1 \mathrm{H}), 7.59(\mathrm{dd}, \mathrm{J}=8.6,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.46$ $(\mathrm{d}, \mathrm{J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.30(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $6.93(\mathrm{t}, \mathrm{J}=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.44(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.54-4.46(\mathrm{~m}, 4 \mathrm{H})$, $4.08(\mathrm{t}, \mathrm{J}=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.83(\mathrm{t}, \mathrm{J}=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.51-3.37(\mathrm{~m}, 8 \mathrm{H})$, $3.23(\mathrm{t}, \mathrm{J}=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.59(\mathrm{~s}, 3 \mathrm{H}), 2.25(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 168.23$, 165.07, 161.40, 160.53, 160.37, 159.25, 158.80, 153.83, 152.40, 148.51, 144.34, 137.17, 133.87, 133.23, 133.08, 132.25, 131.94, 131.20, 130.64, 130.49, 129.59, 128.92, 124.10, $123.26,105.05,101.35,98.17,71.04,70.43,70.16,69.74,69.54,57.52,56.38,55.72,50.15$, 39.13, 27.10, 16.48, 12.72. MS (ESI) $m / z: 876(\mathrm{M}+\mathrm{H})^{+}$


Compound 13 (EML900) was obtained as a white solid ( 38 mg , $52 \%$ ) from derivative 87 ( 45 $\mathrm{mg}, 0.081 \mathrm{mmol}$ ) and $\mathbf{8 5 b}(31 \mathrm{mg}, 0.077 \mathrm{mmol})$ according to the procedure described for $\mathbf{1 2}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.63(\mathrm{~s}, 1 \mathrm{H}), 7.69(\mathrm{~s}, 2 \mathrm{H}), 7.60(\mathrm{dd}, \mathrm{J}=8.7,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.47$ $(\mathrm{d}, \mathrm{J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.43(\mathrm{~s}, 1 \mathrm{H}), 7.37(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.31(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{~d}, \mathrm{~J}=$ $2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{~d}, \mathrm{~J}=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.71-$ $4.48(\mathrm{~m}, 2 \mathrm{H}), 4.47(\mathrm{t}, \mathrm{J}=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.88-3.78(\mathrm{~m}, 4 \mathrm{H}), 3.56-$ $3.39(\mathrm{~m}, 8 \mathrm{H}), 2.85-2.71(\mathrm{~m}, 2 \mathrm{H}), 2.59(\mathrm{~s}, 3 \mathrm{H}), 2.34(\mathrm{~s}, 6 \mathrm{H}), 1.97-1.86(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 168.15,165.10,161.42,160.55,160.38,159.22,153.88,152.36,137.20$, $133.87,133.23,133.13,132.27,131.99,131.21,130.49,129.60,128.94,127.60,124.31$, $121.91,101.36,98.16,72.07,70.42,70.18,69.77,69.57,57.55,56.38,55.72,50.09,39.15$, 29.89, 26.11, 25.49, 16.55, 12.71. MS (ESI) $m / z: 964(\mathrm{M}+\mathrm{H})^{+}$.


Compound 14 (EML824) was obtained as a white solid ( $25 \mathrm{mg}, 60 \%$ ) from derivative 49 (20 $\mathrm{mg}, 0.050 \mathrm{mmol})$ and $\mathbf{9 1}(25 \mathrm{mg}, 0.055 \mathrm{mmol})$ according to the procedure described for $\mathbf{1}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.15(\mathrm{~s}, 2 \mathrm{H}), 7.55(\mathrm{dd}, \mathrm{J}=8.6,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}$, $2 \mathrm{H}), 7.36(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.25(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.83-4.76(\mathrm{~m}$, 2H), $4.56-4.46(\mathrm{~m}, 2 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.93-3.91(\mathrm{~m}, 2 \mathrm{H}), 3.89(\mathrm{~s}, 3 \mathrm{H}), 3.68(\mathrm{~d}, \mathrm{~J}=5.5 \mathrm{~Hz}$, $2 \mathrm{H}), 3.58-3.41(\mathrm{~m}, 6 \mathrm{H}), 2.53(\mathrm{~s}, 3 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.98$, 166.17, 164.07, 160.49, 160.43, 159.20, 158.34, 156.45, 154.81, 148.33, 137.18, 133.85, $133.17,133.09,132.23,131.21,130.46,129.81,129.55,129.01,128.92,123.98,123.02,99.51$, 98.57, 70.66, 70.41, 69.84, 69.34, 66.02, 57.44, 56.10, 55.69, 39.16, 16.12, 12.59. MS (ESI) $m / z: 841(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-hydroxy-3,5-dimethylphenyl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide (15, EML765)


3,4-diamino-N-(pyridin-2-yl)benzenesulfonamide $99 \mathbf{a}$ ( $200 \mathrm{mg}, 0.75 \mathrm{mmol}$ ) was solubilized in 6 mL of dry DMF and 4-hydroxy-3,5-dimethylbenzaldehyde 76a ( $114 \mathrm{mg}, 0.75 \mathrm{mmol}$ ) and $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{5}(187 \mathrm{mg}, 0.98 \mathrm{mmol})$ were added. The resulting mixture was heated at $80{ }^{\circ} \mathrm{C}$ and
stirred at this temperature for 18 h . After cooling at room temperature, water was added. The brown precipitate formed was recovered by filtration and washed several times with water. After recrystallization from EtOH , compound $\mathbf{1 5}$ was obtained as light yellow solid ( 250 mg , $85 \%)$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.98(\mathrm{~s}, 1 \mathrm{H}), 8.07-8.00(\mathrm{~m}, 2 \mathrm{H}), 7.78(\mathrm{~s}, 2 \mathrm{H}), 7.74-7.64$ (m, 2H), $7.68-7.60(\mathrm{~m}, 1 \mathrm{H}), 7.18(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{t}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.26(\mathrm{~s}, 6 \mathrm{H})$. ${ }^{13}$ C NMR (101 MHz, DMSO- $d_{6}$ ) $\delta 156.60,154.92,140.40,127.78,125.30,121.17,119.76$, 113.87, 17.17. MS (ESI) $m / z: 395(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of 2-(4-hydroxy-3,5-dimethylphenyl)-N-(pyridin-2-ylmethyl)-1H-benzo[d]imidazole-6-

 sulfonamide (16, EML803)

Compound 16 (EML803) was obtained as a light yellow solid ( $245 \mathrm{mg}, 56 \%$ ) from derivative 99b ( $298 \mathrm{mg}, 1.07 \mathrm{mmol}$ ) and 4-hydroxy-3,5-dimethylbenzaldehyde 76a (161 mg, 1.07 mmol ) according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 13.01(\mathrm{~s}, 1 \mathrm{H}), 8.88(\mathrm{~s}, 1 \mathrm{H}), 8.43-8.39(\mathrm{~m}, 1 \mathrm{H}), 8.13(\mathrm{~s}, 1 \mathrm{H})$, $8.04-7.85(\mathrm{~m}, 1 \mathrm{H}), 7.80(\mathrm{~s}, 2 \mathrm{H}), 7.74-7.67(\mathrm{~m}, 1 \mathrm{H}), 7.61(\mathrm{~s}, 1 \mathrm{H}), 7.37(\mathrm{~d}, \mathrm{~J}=7.9 \mathrm{~Hz}, 1 \mathrm{H})$, $7.21(\mathrm{dd}, \mathrm{J}=7.5,4.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.05(\mathrm{~d}, \mathrm{~J}=5.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz , DMSO- $d_{6}$ ) $\delta 157.80,156.22,149.15,137.10,133.82,127.63,125.21,122.78,122.07,120.69$, 48.53, 17.19. MS (ESI) $m / z: 409(\mathrm{M}+\mathrm{H})^{+}$ sulfonamide (17, EML798)


Compound 17 (EML798) was obtained as a light yellow solid ( $68 \mathrm{mg}, 72 \%$ ) from derivative 99c ( $64 \mathrm{mg}, 0.230 \mathrm{mmol}$ ) and 4-hydroxy-3,5-dimethylbenzaldehyde 76a ( $34 \mathrm{mg}, 0.230 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 13.03(\mathrm{~s}, 1 \mathrm{H}), 8.89(\mathrm{~s}, 1 \mathrm{H}), 8.45-8.37(\mathrm{~m}, 1 \mathrm{H}), 8.12(\mathrm{t}, J=$ $6.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.95(\mathrm{~s}, 1 \mathrm{H}), 7.81(\mathrm{~s}, 2 \mathrm{H}), 7.70-7.57(\mathrm{~m}, 3 \mathrm{H}), 7.28(\mathrm{dd}, J=7.9,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.02$ $(\mathrm{d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.28(\mathrm{~s}, 7 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO- $d_{6}$ ) $\delta$ 156.24, 149.33, 148.75, 135.83, 133.89, 127.64, 125.21, 123.74, 120.67, 120.61, 44.26, 17.19. MS (ESI) m/z: 409 $(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide (18, EML795)


N-(tert-butyl)-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide $\mathbf{1 0 0}$ ( $136 \mathrm{mg}, 0.364 \mathrm{mmol}$ ) was dissolved in 3.6 mL of a solution DCM/TFA (1:1) and the mixture was stirred for 18 h . The solvent was evaporated, and the resulting solid was crystallized with ethanol to give the title compound as a light brown solid ( $85 \mathrm{mg}, 74 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta 12.98(\mathrm{~s}, 1 \mathrm{H}), 8.87(\mathrm{~s}, 1 \mathrm{H}), 8.06-7.87(\mathrm{~m}, 1 \mathrm{H}), 7.79(\mathrm{~d}, J=$ $2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.74-7.56(\mathrm{~m}, 2 \mathrm{H}), 7.28-7.22(\mathrm{~m}, 2 \mathrm{H}), 2.27(\mathrm{~d}, J=2.9 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO- $d_{6}$ ) $\delta 156.18,156.09,155.35,154.84,146.49,143.63,138.01,137.92,137.53$,
134.60, 127.56, 125.18, 120.76, 120.04, 119.68, 118.63, 116.57, 111.48, 109.56, 17.17. MS (ESI) $m / z: 318(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(4-hydroxy-3,5-dimethylphenyl)-N-methyl-1H-benzo[d]imidazole-6-sulfonamide (19, EML796)


Compound 19 (EML796) was obtained as a white solid ( 60 mg , 73\%) from derivative 99e (50 $\mathrm{mg}, 0.248 \mathrm{mmol}$ ) and 4-hydroxy-3,5-dimethylbenzaldehyde 76a ( $37 \mathrm{mg}, 0.248 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta 7.91(\mathrm{~s}, 1 \mathrm{H}), 7.79(\mathrm{~s}, 2 \mathrm{H}), 7.68(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.57$ (dd, $J=8.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.36-7.28(\mathrm{~m}, 1 \mathrm{H}), 2.40(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 3 \mathrm{H}), 2.27(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO- $d_{6}$ ) $\delta 156.23,132.61,127.63,125.21,120.68,29.22,17.19$. MS (ESI) $m / z: 332$ $(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of N-cyclopropyl-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide

 (20, EML801)

Compound 20 (EML801) was obtained as a light yellow solid ( $64 \mathrm{mg}, 63 \%$ ) from derivative $\mathbf{9 9 f}(64 \mathrm{mg}, 0.281 \mathrm{mmol})$ and 4-hydroxy-3,5-dimethylbenzaldehyde $7 \mathbf{6 a}(25 \mathrm{mg}, 0.281 \mathrm{mmol})$ according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 13.02(\mathrm{~s}, 1 \mathrm{H}), 8.88(\mathrm{~s}, 1 \mathrm{H}), 8.01-7.88(\mathrm{~m}, 1 \mathrm{H}), 7.79(\mathrm{~s}, 2 \mathrm{H})$, $7.73-7.62(\mathrm{~m}, 1 \mathrm{H}), 7.61(\mathrm{~d}, \mathrm{~J}=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.27(\mathrm{~s}, 6 \mathrm{H}), 2.07(\mathrm{dd}, \mathrm{J}=6.5,3.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.49$
$-0.41(\mathrm{~m}, 2 \mathrm{H}), 0.41-0.34(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO- $\left.d_{6}\right) \delta 156.22,133.69,127.62$, 125.21, 120.69, 24.64, 17.19, 5.56. MS (ESI) $m / z: 358(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of N-cyclopentyl-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide

 (21, EML797)

Compound 21 (EML797) was obtained as a yellow solid ( 345 mg , $58 \%$ ) from derivative 99 g ( $390 \mathrm{mg}, 1.53 \mathrm{mmol}$ ) and 4-hydroxy-3,5-dimethylbenzaldehyde 76a ( $230 \mathrm{mg}, 1.53 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 9.25-9.11(\mathrm{~m}, 1 \mathrm{H}), 7.99(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.82(\mathrm{~s}, 2 \mathrm{H})$, $7.77(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{dd}, J=8.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.66(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.45-3.35$ $(\mathrm{m}, 1 \mathrm{H}), 2.28(\mathrm{~s}, 6 \mathrm{H}), 1.60-1.49(\mathrm{~m}, 4 \mathrm{H}), 1.39-1.26(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO$\left.d_{6}\right) \delta 157.40,154.22,136.55,128.19,125.56,121.94,118.01,114.96,113.79,65.38,54.97$, 32.92, 23.27, 17.18. MS (ESI) $m / z: 386(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of N-cyclohexyl-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide

 (22, EML802)

Compound 22 (EML802) was obtained as a yellow solid ( 160 mg , 54\%) from derivative 99h ( $200 \mathrm{mg}, 0.742 \mathrm{mmol}$ ) and 4-hydroxy-3,5-dimethylbenzaldehyde $76 \mathbf{a}$ ( $111 \mathrm{mg}, 0.742 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO- $d_{6}$ ) $\delta 13.00(\mathrm{~s}, 1 \mathrm{H}), 8.88(\mathrm{~s}, 1 \mathrm{H}), 8.06-7.85(\mathrm{~m}, 1 \mathrm{H}), 7.79(\mathrm{~s}, 2 \mathrm{H})$, $7.73-7.41(\mathrm{~m}, 3 \mathrm{H}), 2.91(\mathrm{~s}, 1 \mathrm{H}), 2.27(\mathrm{~s}, 6 \mathrm{H}), 1.59-1.34(\mathrm{~m}, 5 \mathrm{H}), 1.19-0.95(\mathrm{~m}, 5 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO) $\delta 156.19,135.76,127.60,125.20,120.72,33.67,25.36,24.82,17.19$. MS (ESI) $m / z: 400(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-hydroxyphenyl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide (23, EML760)


Compound 23 (EML760) was obtained as a white solid ( 180 mg , 65\%) from derivative 99a ( $200 \mathrm{mg}, 0.757 \mathrm{mmol}$ ) and 4-hydroxybenzaldehyde $\mathbf{7 6 b}$ ( $92 \mathrm{mg}, 0.757 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 13.04(\mathrm{~s}, 1 \mathrm{H}), 10.06(\mathrm{~s}, 1 \mathrm{H}), 8.12-8.01(\mathrm{~m}, 2 \mathrm{H}), 8.00(\mathrm{~d}, J$ $=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.72-7.55(\mathrm{~m}, 3 \mathrm{H}), 7.17(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.86(\mathrm{t}, J$ $=6.3 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO- $d_{6}$ ) $\delta 160.21,129.04,120.76,116.28$. MS (ESI) $m / z: 367(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(3-hydroxyphenyl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide (24, EML761)


Compound 24 (EML761) was obtained as a white solid ( 270 mg , $65 \%$ ) from derivative 99a ( $300 \mathrm{mg}, 1.13 \mathrm{mmol}$ ) and 3-hydroxybenzaldehyde 76c ( $137 \mathrm{mg}, 1.13 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 9.91(\mathrm{~s}, 1 \mathrm{H}), 8.14(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.01(\mathrm{dd}, J=5.5,1.7$ $\mathrm{Hz}, 1 \mathrm{H}), 7.80(\mathrm{dd}, J=8.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.76(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.74-7.67(\mathrm{~m}, 1 \mathrm{H}), 7.62-$ $7.55(\mathrm{~m}, 2 \mathrm{H}), 7.41(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.20(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.04-6.96(\mathrm{~m}, 1 \mathrm{H}), 6.87(\mathrm{t}, J$ $=6.4 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz , DMSO- $d_{6}$ ) $\delta$ 158.41, 153.98, 153.47, 140.73, 130.87, 129.34, 122.03, 119.02, 118.35, 114.35, 114.06. MS (ESI) $m / z: 367(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(4-chlorophenyl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide (25, EML762)


Compound 25 (EML762) was obtained as a pale yellow solid ( $135 \mathrm{mg}, 52 \%$ ) from derivative 99a ( $180 \mathrm{mg}, 0.681 \mathrm{mmol}$ ) and 4-chlorobenzaldehyde $76 \mathbf{d}(100 \mathrm{mg}, 0.681 \mathrm{mmol})$ according to the procedure described for 15.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.18(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 8.13-8.12(\mathrm{~m}, 1 \mathrm{H}), 8.02(\mathrm{dd}, J=$ $5.4,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.77-7.73(\mathrm{~m}, 2 \mathrm{H}), 7.72-7.69(\mathrm{~m}, 1 \mathrm{H}), 7.67(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.22-7.16$ $(\mathrm{m}, 1 \mathrm{H}), 6.91-6.84(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta$ 153.38, 140.43, 135.86, 129.72, 129.04, 128.53, 121.51, 113.86. MS (ESI) $m / z: 385(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of 2-(2,4-dihydroxyphenyl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide (26,

 EML763)

Compound 26 (EML763) was obtained as a light yellow solid ( $260 \mathrm{mg}, 60 \%$ ) from derivative 99a ( $300 \mathrm{mg}, 1.13 \mathrm{mmol}$ ) and 2,4-dihydroxybenzaldehyde $\mathbf{7 6 e}(157 \mathrm{mg}, 1.13 \mathrm{mmol})$ according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta 10.26(\mathrm{~s}, 2 \mathrm{H}), 8.12(\mathrm{~s}, 1 \mathrm{H}), 8.01(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.87(\mathrm{~d}$, $J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.77(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.76-7.66(\mathrm{~m}, 3 \mathrm{H}), 7.19(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{t}$, $J=6.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.52-6.45(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz, DMSO- $d_{6}$ ) $\delta 162.48,160.25,140.72$, 129.22, 121.88, 114.08, 108.73, 103.48. MS (ESI) $m / z: 383(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(3,4-dihydroxyphenyl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide (27, EML764)


Compound 27 (EML764) was obtained as a pale yellow solid ( $255 \mathrm{mg}, 59 \%$ ) from derivative 99a ( $300 \mathrm{mg}, 1.14 \mathrm{mmol}$ ) and 3,4-dihydroxybenzaldehyde $\mathbf{7 6 f}(157 \mathrm{mg}, 1.14 \mathrm{mmol})$ according to the procedure described for 15.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 10.01$ ( $\mathrm{s}, 1 \mathrm{H}$ ), $9.52(\mathrm{~s}, 1 \mathrm{H}), 8.10(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.01$ $(\mathrm{dd}, J=5.7,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{dd}, J=8.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.78-7.67(\mathrm{~m}, 2 \mathrm{H}), 7.58(\mathrm{~d}, J=2.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.51(\mathrm{dd}, J=8.3,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.20(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.97(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.91$ $-6.83(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO- $d_{6}$ ) $\delta$ 153.94, 153.58, 150.24, 146.43, 140.91, 122.34, 120.31, 116.60, 116.20, 115.16, 114.83, 114.20, 113.77. MS (ESI) $m / z: 383(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(4-hydroxy-2,6-dimethylphenyl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide (28, EML766)


Compound 28 (EML766) was obtained as a pale yellow solid (190 mg, 67\%) from derivative 99a ( $190 \mathrm{mg}, 0.719 \mathrm{mmol}$ ) and 4-hydroxy-2,6-dimethylbenzaldehyde 76 g ( $108 \mathrm{mg}, 0.719$ mmol ) according to the procedure described for $\mathbf{1 5}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 12.83(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 1 \mathrm{H}), 9.60(\mathrm{~d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.19-$ $7.98(\mathrm{~m}, 2 \mathrm{H}), 7.79-7.66(\mathrm{~m}, 2 \mathrm{H}), 7.59(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.25-7.16(\mathrm{~m}, 1 \mathrm{H}), 6.88(\mathrm{t}, J=$ $6.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.58(\mathrm{~s}, 2 \mathrm{H}), 2.01(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz, DMSO- $d_{6}$ ) $\delta$ 158.48, 143.10, 122.21, 121.02, 120.13, 119.23, 118.27, 114.71, 111.89, 20.41. MS (ESI) $m / z: 395(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of 2-(4-hydroxy-3-methylphenyl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide

 (29, EML806)

Compound 29 (EML806) was obtained as a light yellow solid ( $130 \mathrm{mg}, 54 \%$ ) from derivative 99a ( $167 \mathrm{mg}, 0.632 \mathrm{mmol}$ ) and 4-hydroxy-3-methylbenzaldehyde $76 \mathrm{~h}(86 \mathrm{mg}, 0.632 \mathrm{mmol})$ according to the procedure described for 15.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 10.39(\mathrm{~s}, 1 \mathrm{H}), 8.10(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.01(\mathrm{dd}, J=5.6,1.9$ $\mathrm{Hz}, 1 \mathrm{H}), 7.94(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.87(\mathrm{dd}, J=8.4,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.82(\mathrm{dd}, J=8.6,1.7 \mathrm{~Hz}, 1 \mathrm{H})$, $7.75(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.75-7.67(\mathrm{~m}, 1 \mathrm{H}), 7.21(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.00(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H})$, $6.91-6.83(\mathrm{~m}, 1 \mathrm{H}), 2.23(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{DMSO}$ ) $\delta$ 159.91, 153.88, 153.59, 140.94, 130.55, 127.36, 125.70, 122.39, 116.30, 116.15, 114.78, 114.75, 113.72, 16.45. MS (ESI) $m / z: 381(\mathrm{M}+\mathrm{H})^{+}$


Compound $\mathbf{3 0}$ (EML799) was obtained as a yellow solid ( 85 mg , 53\%) from derivative 99a (91 $\mathrm{mg}, 0.345 \mathrm{mmol}$ ) and the intermediate $\mathbf{1 0 3 a}(78 \mathrm{mg}, 0.345 \mathrm{mmol})$ according to the procedure described for 15.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO-d6) $\delta 11.96-11.86(\mathrm{~m}, 2 \mathrm{H}), 8.80(\mathrm{~d}, \mathrm{~J}=6.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.04-7.91$ (m, 2H), $7.66(\mathrm{~s}, 1 \mathrm{H}), 7.60-7.47(\mathrm{~m}, 2 \mathrm{H}), 7.34(\mathrm{t}, \mathrm{J}=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.27-7.18(\mathrm{~m}, 3 \mathrm{H}), 7.14$ - $7.03(\mathrm{~m}, 3 \mathrm{H}), 6.86(\mathrm{~s}, 1 \mathrm{H}), 2.28(\mathrm{~s}, 3 \mathrm{H}), 1.95(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO) $\delta$ 190.79, $159.28,155.38,140.49,139.70,137.26,130.36,130.27,130.17,128.67,128.26,128.03$, $127.50,127.23,124.53,123.88,123.77,17.02,14.58,13.86$. MS (ESI) $m / z: 471(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of 2-(5-hydroxy-3'-methoxy-4,6-dimethyl-[1,1'-biphenyl]-2-yl)-N-(pyridin-2-yl)-1H-

## benzo[d]imidazole-6-sulfonamide (31, EML804)



Compound 31 (EML804) was obtained as a pale yellow solid ( $95 \mathrm{mg}, 65 \%$ ) from derivative 99a ( $77 \mathrm{mg}, 0.291 \mathrm{mmol}$ ) and the intermediate $\mathbf{1 0 3 b}(75 \mathrm{mg}, 0.291 \mathrm{mmol})$ according to the procedure described for 15.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO-d6) $\delta 11.95-11.82(\mathrm{~m}, 1 \mathrm{H}), 8.81(\mathrm{~d}, \mathrm{~J}=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.06-7.94$ $(\mathrm{m}, 1 \mathrm{H}), 7.69-7.62(\mathrm{~m}, 1 \mathrm{H}), 7.62-7.55(\mathrm{~m}, 2 \mathrm{H}), 7.41-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.17-7.08(\mathrm{~m}, 2 \mathrm{H})$,
$6.89-6.83(\mathrm{~m}, 1 \mathrm{H}), 6.81-6.73(\mathrm{~m}, 1 \mathrm{H}), 6.64-6.63(\mathrm{~m}, 1 \mathrm{H}), 6.60(\mathrm{~d}, \mathrm{~J}=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.61$ $(\mathrm{dd}, \mathrm{J}=8.0,2.0 \mathrm{~Hz}, 3 \mathrm{H}), 2.27(\mathrm{~s}, 3 \mathrm{H}), 1.97(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO) $\delta$ 159.04, $158.96,157.34,156.49,155.91,155.26,155.19,146.15,142.82,141.06,141.01,140.18$, 140.14, 133.91, 130.23, 129.38, 129.28, 123.88, 123.62, 122.53, 121.97, 120.74, 119.97, 118.91, 118.04, 116.13, 112.66, 111.71, 111.04, 55.32, 17.00, 14.57. MS (ESI) m/z: 501 $(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(3'-(dimethylamino)-5-hydroxy-4,6-dimethyl-[1,1'-biphenyl]-2-yl)-N-(pyridin-2-yl)-1H-benzo[d]imidazole-6-sulfonamide (32, EML805)


Compound 32 (EML805) was obtained as a yellow solid ( 60 mg , $67 \%$ ) from derivative 99a (54 $\mathrm{mg}, 0.204 \mathrm{mmol})$ and the intermediate $\mathbf{1 0 3 c}(55 \mathrm{mg}, 0.204 \mathrm{mmol})$ according to the procedure described for 15.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 11.69-11.57(\mathrm{~m}, 1 \mathrm{H}), 8.74(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.04-7.95$ (m, 1H), $7.71-7.62(\mathrm{~m}, 1 \mathrm{H}), 7.57(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.39-7.24(\mathrm{~m}, 2 \mathrm{H}), 7.18-7.00(\mathrm{~m}$, 2H), $6.87(\mathrm{~s}, 1 \mathrm{H}), 6.61-6.50(\mathrm{~m}, 1 \mathrm{H}), 6.42-6.34(\mathrm{~m}, 2 \mathrm{H}), 2.71(\mathrm{~s}, 3 \mathrm{H}), 2.68(\mathrm{~s}, 3 \mathrm{H}), 2.26(\mathrm{~s}$, 3H), 1.99 (s, 3H). ${ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO) $\delta 156.88,156.19,155.23,155.16,150.27$, 146.14, 142.79, 141.14, 130.14, 128.85, 128.74, 123.56, 123.49, 122.05, 120.70, 119.96, 118.83, 118.37, 118.27, 117.99, 114.59, 114.48, 111.41, 40.48, 16.98, 14.65. MS (ESI) $m / z:$ $514(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of (2S,4R)-1-((S)-2-(tert-butyl)-18-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo
[d]imidazole)-6-sulfonamido)pyridin-3-yl)-4-oxo-6,9,12,15-tetraoxa-3-azaoctadec-17-ynoyl)-4-hydroxy-N-(4-(4-methylthiazol-5-yl)benzyl)pyrrolidine-2-carboxamide (33)


To a solution of compound $\mathbf{1 0 6}(3.45 \mathrm{mg}, 0.0054 \mathrm{mmol})$ in dry 0.300 mL DMF mL , a solution of compound 104 ( $2.52 \mathrm{mg}, 0.0054 \mathrm{mmol}$ ) in 0.100 mL of dry DMF, PyOXIM ( $2.65 \mathrm{mg}, 0.0054$ $\mathrm{mmol})$, and DIPEA ( $0.009 \mathrm{~mL}, 0.0504 \mathrm{mmol}$ ) were added. The reaction mixture was left to stir for 1 h and monitored by LC-MS (acidic method). When completed, the crude reaction was purified by HPLC with a gradient from $5 \%$ to $90 \% \mathrm{v} / \mathrm{v}$ acetonitrile with $0.01 \% \mathrm{v} / \mathrm{v}$ aqueous solution of formic acid over 20 min to yield the title compound ( $2.81 \mathrm{mg}, 50 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 8.82(\mathrm{~s}, 1 \mathrm{H}), 8.18(\mathrm{~s}, 1 \mathrm{H}), 8.14(\mathrm{~s}, 1 \mathrm{H}), 7.81(\mathrm{dd}, \mathrm{J}=8.6,1.8$ $\mathrm{Hz}, 1 \mathrm{H}), 7.71(\mathrm{~s}, 2 \mathrm{H}), 7.66-7.59(\mathrm{~m}, 2 \mathrm{H}), 7.43-7.38(\mathrm{~m}, 2 \mathrm{H}), 7.35(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.17$ $(\mathrm{d}, \mathrm{J}=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.66(\mathrm{~s}, 1 \mathrm{H}), 4.59-4.54(\mathrm{~m}, 1 \mathrm{H}), 4.52-4.46(\mathrm{~m}, 2 \mathrm{H}), 4.33(\mathrm{~s}, 2 \mathrm{H}), 4.32-$ $4.26(\mathrm{~m}, 1 \mathrm{H}), 4.04-3.94(\mathrm{~m}, 2 \mathrm{H}), 3.89-3.82(\mathrm{~m}, 1 \mathrm{H}), 3.80-3.73(\mathrm{~m}, 1 \mathrm{H}), 3.67-3.57(\mathrm{~m}$, $12 \mathrm{H}), 2.43(\mathrm{~s}, 3 \mathrm{H}), 2.30(\mathrm{~s}, 6 \mathrm{H}), 2.21-2.18(\mathrm{~m}, 1 \mathrm{H}), 2.06-2.03(\mathrm{~m}, 1 \mathrm{H}), 1.00(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 173.00,170.73,170.31,156.40,151.42,147.59,138.77,132.00$, $130.05,128.94,127.51,127.21,124.93,121.15,119.59,87.41,81.87,70.88,70.23,70.19$, $70.05,70.02,69.67,69.65,68.86,59.42,58.19,56.73,45.96,42.30,37.54,35.69,26.00,15.42$, 14.43. LC/MS 3.5-3.8 min, $m / z: 526.2072$, $1051.2073(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of (2S,4R)-4-hydroxy-1-((S)-2-(2-((6-((3-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole)-6-sulfonamido)pyridin-3-yl)prop-2-yn-1-yl)oxy)hexyl)oxy)acetamido)-3,3-dimethylbutanoyl)-N-(4-(4-methylthiazol-5-yl)benzyl)pyrrolidine-2-carboxamide (34)


Compound $\mathbf{3 4}$ was obtained as a white solid ( $1.5 \mathrm{mg}, 32 \%$ ) from derivative $107(3 \mathrm{mg}, 0.0045$ $\mathrm{mmol})$ and the intermediate $\mathbf{1 0 4}(2.31 \mathrm{mg}, 0.0045 \mathrm{mmol})$ according to the procedure described for 33.
${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 8.80(\mathrm{~s}, 1 \mathrm{H}), 8.23-8.15(\mathrm{~m}, 1 \mathrm{H}), 8.14-8.09(\mathrm{~m}, 1 \mathrm{H}), 7.81$ $(\mathrm{dd}, \mathrm{J}=8.6,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.70(\mathrm{~s}, 2 \mathrm{H}), 7.58(\mathrm{dd}, \mathrm{J}=8.7,2.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.42-7.38(\mathrm{~m}, 2 \mathrm{H}), 7.37$ - $7.31(\mathrm{~m}, 2 \mathrm{H}), 7.16-7.12(\mathrm{~m}, 1 \mathrm{H}), 4.66(\mathrm{~s}, 1 \mathrm{H}), 4.61-4.55(\mathrm{~m}, 1 \mathrm{H}), 4.53-4.46(\mathrm{~m}, 2 \mathrm{H})$, $4.30(\mathrm{~d}, \mathrm{~J}=1.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.26(\mathrm{~s}, 2 \mathrm{H}), 3.97-3.88(\mathrm{~m}, 2 \mathrm{H}), 3.88-3.83(\mathrm{~m}, 2 \mathrm{H}), 3.53-3.47(\mathrm{~m}$, $6 \mathrm{H}), 2.42(\mathrm{~s}, 3 \mathrm{H}), 2.30(\mathrm{~s}, 6 \mathrm{H}), 2.25-2.17(\mathrm{~m}, 1 \mathrm{H}), 2.10-2.02(\mathrm{~m}, 1 \mathrm{H}), 1.65-1.53(\mathrm{~m}, 6 \mathrm{H})$, $0.99(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 172.96,170.63,170.27,151.39,147.56,138.74$, 132.01, 130.04, 128.92, 127.50, 127.27, 124.92, 119.66, 71.44, 69.68, 69.55, 69.31, 59.42, $57.89,56.73,56.56,42.30,37.53,35.83,29.17,29.08,25.58,25.55,25.49,15.41,14.42$. LC/MS 3.7-3.9 min, $m / z: 510.2151,1019.4473(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of (2S,4R)-4-hydroxy-1-((S)-2-(2-((4-(((3-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo [d]imidazole)-6-sulfonamido)pyridin-3-yl)prop-2-yn-1-yl)oxy)methyl)benzyl)oxy)acetamido)-3,3-dimethylbutanoyl)-N-(4-(4-methylthiazol-5-yl)benzyl)pyrrolidine-2-carboxamide (35)


Compound 35 was obtained as a white solid ( $1.52 \mathrm{mg}, 42 \%$ ) from derivative 108 ( 2.2 mg , $0.0035 \mathrm{mmol})$ and the intermediate $104(1.64 \mathrm{mg}, 0.0035 \mathrm{mmol})$ according to the procedure described for 33
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 8.80(\mathrm{~s}, 1 \mathrm{H}), 8.22-8.13(\mathrm{~m}, 1 \mathrm{H}), 8.12(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.81$ $(\mathrm{dd}, J=8.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.70(\mathrm{~s}, 2 \mathrm{H}), 7.62(\mathrm{dd}, J=8.7,2.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.44-7.39(\mathrm{~m}, 2 \mathrm{H}), 7.40$ $-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.36-7.30(\mathrm{~m}, 4 \mathrm{H}), 7.17(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.68(\mathrm{~s}, 1 \mathrm{H}), 4.64-4.58(\mathrm{~m}, 1 \mathrm{H})$, $4.58(\mathrm{~s}, 2 \mathrm{H}), 4.56(\mathrm{~s}, 2 \mathrm{H}), 4.55-4.47(\mathrm{~m}, 2 \mathrm{H}), 4.32(\mathrm{~s}, 2 \mathrm{H}), 4.31-4.24(\mathrm{~m}, 1 \mathrm{H}), 4.06-3.93$ $(\mathrm{m}, 2 \mathrm{H}), 3.90-3.84(\mathrm{~m}, 1 \mathrm{H}), 3.82-3.75(\mathrm{~m}, 1 \mathrm{H}), 2.41(\mathrm{~s}, 3 \mathrm{H}), 2.30(\mathrm{~s}, 6 \mathrm{H}), 2.26-2.17(\mathrm{~m}$, $1 \mathrm{H}), 2.11-2.05(\mathrm{~m}, 1 \mathrm{H}), 1.00(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 172.97,170.33,156.38$, $151.38,147.57,138.76,137.58,136.86,132.02,130.02,128.90,128.02,127.77,127.50$, $127.20,124.92,119.61,87.24,72.81,71.06,69.70,68.53,59.48,57.18,56.76,56.69,42.28$ $37.53,35.84,35.13,31.68,29.43,29.22,29.07,28.96,28.84,26.73,25.49,15.41,14.42,13.05$. LC/MS 3.7-3.9 min, $m / z: 520.2514,1039.5043(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of (2S,4R)-1-((S)-2-(tert-butyl)-19-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-
benzo[d]imidazole)-6-sulfonamido)pyridin-3-yl)-4-oxo-6,11,16-trioxa-3-azanonadec-18-ynoyl)-4-hydroxy-N-(4-(4-methylthiazol-5-yl)benzyl)pyrrolidine-2-carboxamide (36)


Compound 36 was obtained as a white solid ( $2.41 \mathrm{mg}, 32 \%$ ) from derivative $\mathbf{1 0 9}$ ( 4.75 mg , $0.0073 \mathrm{mmol})$ and the intermediate $\mathbf{1 0 4}(3.4 \mathrm{mg}, 0.0073 \mathrm{mmol})$ according to the procedure described for 33.
${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 8.83(\mathrm{~s}, 1 \mathrm{H}), 8.23-8.15(\mathrm{~m}, 1 \mathrm{H}), 8.14(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.81$ (dd, J = 8.6, 1.8 Hz, 1H), $7.70(\mathrm{~s}, 2 \mathrm{H}), 7.62(\mathrm{dd}, \mathrm{J}=8.7,2.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.42(\mathrm{~d}, \mathrm{~J}=3.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.40(\mathrm{~s}, 1 \mathrm{H}), 7.38-7.33(\mathrm{~m}, 2 \mathrm{H}), 7.16(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.68(\mathrm{~s}, 1 \mathrm{H}), 4.62-4.55(\mathrm{~m}, 1 \mathrm{H})$, $4.52-4.48(\mathrm{~m}, 2 \mathrm{H}), 4.31(\mathrm{~d}, \mathrm{~J}=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.28(\mathrm{~s}, 2 \mathrm{H}), 3.97-3.90(\mathrm{~m}, 2 \mathrm{H}), 3.89-3.84(\mathrm{~m}$, $1 \mathrm{H}), 3.82-3.75(\mathrm{~m}, 1 \mathrm{H}), 3.53-3.50(\mathrm{~m}, 2 \mathrm{H}), 3.48(\mathrm{t}, \mathrm{J}=5.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.40-3.36(\mathrm{~m}, 4 \mathrm{H}), 2.43$ (s, 3H), $2.30(\mathrm{~s}, 6 \mathrm{H}), 2.26-2.19(\mathrm{~m}, 1 \mathrm{H}), 2.11-2.05(\mathrm{~m}, 1 \mathrm{H}), 1.62-1.57(\mathrm{~m}, 8 \mathrm{H}), 1.01(\mathrm{~s}$, 9H). ${ }^{13}$ C NMR ( $126 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta$ 172.95, 170.59, 170.33, 156.37, 151.41, 147.58, 140.88, 138.76, 132.02, 130.05, 128.93, 128.09, 127.51, 127.19, 124.92, 119.63, 87.56, 81.71, 71.25, $70.09,70.04,69.68,69.42,69.27,59.45,57.87,56.76,56.60,42.30,37.55,35.81,26.05,25.98$, 25.97, 25.94, 25.52, 15.42, 14.43. LC/MS 3.7-3.9 min, $m / z: 532.2278,1063.4492(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of N-(5-(1-(4-(2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-5-yl)piperazin-1-yl)-1-oxo-3,6,9,12-tetraoxapentadec-14-yn-15-yl)pyridin-2-yl)-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide (37)


Compound $\mathbf{3 7}$ was obtained as a yellow solid ( $1.46 \mathrm{mg}, 28 \%$ ) from derivative 106 ( 3.51 mg , $0.0055 \mathrm{mmol})$ and the intermediate $\mathbf{1 0 5}(2.3 \mathrm{mg}, 0.0055 \mathrm{mmol})$ according to the procedure described for 33.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO-d6) $\delta 11.08(\mathrm{~s}, 1 \mathrm{H}), 8.85(\mathrm{~s}, 1 \mathrm{H}), 8.15(\mathrm{~s}, 1 \mathrm{H}), 7.93(\mathrm{~s}, 1 \mathrm{H}), 7.76$ (d, J = 3.3 Hz, 2H), $7.68(\mathrm{~d}, \mathrm{~J}=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.63-7.49(\mathrm{~m}, 2 \mathrm{H}), 7.34(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.21$ (dd, J = 8.7, 2.3 Hz, 1H), $5.10-5.03(\mathrm{~m}, 1 \mathrm{H}), 4.32(\mathrm{~s}, 2 \mathrm{H}), 4.18(\mathrm{~s}, 2 \mathrm{H}), 3.57-3.48(\mathrm{~m}, 12 \mathrm{H})$, $3.44(\mathrm{~s}, 4 \mathrm{H}), 3.05-2.97(\mathrm{~m}, 2 \mathrm{H}), 2.93-2.82(\mathrm{~m}, 2 \mathrm{H}), 2.25(\mathrm{~s}, 6 \mathrm{H}), 2.08-1.92(\mathrm{~m}, 4 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 171.76,169.37,167.66,167.62,167.02,155.39,151.20$, $150.80,140.88,134.37,127.47,124.69,124.45,120.87,119.75,118.10,111.40,108.19,88.06$, 82.12, 70.27, 70.25, 70.18, 70.11, 70.09, 70.03, 68.97, 58.33, 49.21, 47.46, 46.92, 44.09, 40.86, 31.12, 22.55, 15.83, 13.48. LC/MS 3.2-3.3 min, $m / z: 482.1710,963.3276(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of N-(5-(3-((6-(2-(4-(2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-5-yl)piperazin-1-yl)-2-oxoethoxy)hexyl)oxy)prop-1-yn-1-yl)pyridin-2-yl)-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide (38)


Compound $\mathbf{3 8}$ was obtained as a yellow solid ( $1.5 \mathrm{mg}, 27 \%$ ) from derivative 107 ( 3.64 mg , $0.006 \mathrm{mmol})$ and the intermediate $\mathbf{1 0 5}(2.49 \mathrm{mg}, 0.006 \mathrm{mmol})$ according to the procedure described for 33.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 11.08(\mathrm{~s}, 1 \mathrm{H}), 8.90(\mathrm{~s}, 1 \mathrm{H}), 8.22(\mathrm{~s}, 1 \mathrm{H}), 8.06(\mathrm{~s}, 1 \mathrm{H}), 7.77$ (s, 2H), $7.73(\mathrm{dd}, \mathrm{J}=8.7,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.72-7.65(\mathrm{~m}, 2 \mathrm{H}), 7.64(\mathrm{~s}, 1 \mathrm{H}), 7.34(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}$, $1 \mathrm{H}), 7.22(\mathrm{dd}, \mathrm{J}=8.7,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.12(\mathrm{~d}, \mathrm{~J}=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.13-5.02(\mathrm{~m}, 1 \mathrm{H}), 4.29(\mathrm{~s}, 2 \mathrm{H})$, $4.12(\mathrm{~s}, 2 \mathrm{H}), 3.59-3.53(\mathrm{~m}, 4 \mathrm{H}), 3.52-3.45(\mathrm{~m}, 4 \mathrm{H}), 2.94-2.81(\mathrm{~m}, 2 \mathrm{H}), 2.69-2.60(\mathrm{~m}$, $2 H), 2.25(\mathrm{~s}, 6 \mathrm{H}), 2.07-1.93(\mathrm{~m}, 4 \mathrm{H}), 1.53-1.44(\mathrm{~m}, 4 \mathrm{H}), 1.34-1.26(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 171.84,171.80,169.45,168.34,167.57,167.00,155.37,152.89$, $150.86,141.24,134.40,128.77,125.61,124.72,124.17,119.88,118.18,114.41,108.20,88.75$, 81.63, 70.85, 69.90, 69.25, 57.97, 49.20, 47.44, 41.01, 40.46, 31.10, 25.60, 25.49, 22.53, 15.78. LC/MS 3.6-3.7 min, $m / z: ~ 931.3247(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of N-(5-(3-((4-((2-(4-(2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-5-yl)piperazin-1-yl)-2-oxoethoxy)methyl)benzyl)oxy)prop-1-yn-1-yl)pyridin-2-yl)-2-(4-hydroxy-3,5-dimethylphenyl)-1H-
benzo[d]imidazole-6-sulfonamide (39)


Compound 39 was obtained as a yellow solid ( $2.53 \mathrm{mg}, 76 \%$ ) from derivative $\mathbf{1 0 8}(2.2 \mathrm{mg}$, $0.0035 \mathrm{mmol})$ and the intermediate $\mathbf{1 0 5}(1.45 \mathrm{mg}, 0.0035 \mathrm{mmol})$ according to the procedure described for 33.
${ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO- $d_{6}$ ) $\delta 11.05(\mathrm{~s}, 1 \mathrm{H}), 8.82(\mathrm{~s}, 1 \mathrm{H}), 8.19(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.08(\mathrm{~s}$, $1 \mathrm{H}), 7.96(\mathrm{~s}, 1 \mathrm{H}), 7.76(\mathrm{~s}, 2 \mathrm{H}), 7.71-7.53$ (m, 4H), 7.33 (d, $J=8.8 \mathrm{~Hz}, 4 \mathrm{H}), 7.23(\mathrm{dd}, J=8.6$,
$2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.08-6.99(\mathrm{~m}, 1 \mathrm{H}), 5.10-5.01(\mathrm{~m}, 1 \mathrm{H}), 4.54(\mathrm{~s}, 2 \mathrm{H}), 4.52(\mathrm{~s}, 2 \mathrm{H}), 4.37(\mathrm{~s}, 2 \mathrm{H})$, $4.24(\mathrm{~s}, 2 \mathrm{H}), 3.59(\mathrm{~s}, 4 \mathrm{H}), 3.49(\mathrm{~s}, 4 \mathrm{H}), 2.94-2.82(\mathrm{~m}, 2 \mathrm{H}), 2.26(\mathrm{~s}, 6 \mathrm{H}), 2.07-1.95(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 171.87,169.41,167.59,167.44,167.03,155.86,155.44$, $151.33,150.78,140.91,137.59,134.41,133.43,127.93,127.41,124.71,124.51,120.99$, $120.85,119.83,118.18,111.41,108.22,87.87,82.41,72.38,71.07,69.41,57.42,49.21,47.48$, 47.01, 44.16, 40.83, 31.13, 22.54, 15.83. LC/MS 3.4-3.6 min, $m / z: ~ 951.2(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of N-(5-(3-(4-(4-(2-(4-(2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-5-yl)piperazin-1-yl)-2-oxoethoxy)butoxy)butoxy)prop-1-yn-1-yl)pyridin-2-yl)-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide (40)


Compound $\mathbf{4 0}$ was obtained as a yellow solid ( $2.67 \mathrm{mg}, 38 \%$ ) from derivative $\mathbf{1 0 9}$ ( 4.75 mg , $0.0073 \mathrm{mmol})$ and the intermediate $\mathbf{1 0 5}(3.03 \mathrm{mg}, 0.0073 \mathrm{mmol})$ according to the procedure described for 33.
${ }^{1} \mathrm{H}$ NMR ( 500 MHz, DMSO- $d_{6}$ ) $\delta 11.04(\mathrm{~s}, 1 \mathrm{H}), 8.81(\mathrm{~s}, 1 \mathrm{H}), 8.17-8.11(\mathrm{~m}, 1 \mathrm{H}), 8.06(\mathrm{~s}, 1 \mathrm{H})$, $7.94(\mathrm{~s}, 1 \mathrm{H}), 7.76(\mathrm{~s}, 2 \mathrm{H}), 7.68(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.64-7.48(\mathrm{~m}, 2 \mathrm{H}), 7.34(\mathrm{~s}, 1 \mathrm{H}), 7.23(\mathrm{~d}, \mathrm{~J}$ $=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.03-6.93(\mathrm{~m}, 1 \mathrm{H}), 5.11-5.01(\mathrm{~m}, 1 \mathrm{H}), 4.28(\mathrm{~s}, 2 \mathrm{H}), 4.13(\mathrm{~s}, 2 \mathrm{H}), 3.60-3.56$ (m, 4H), $3.53-3.38(\mathrm{~m}, 8 \mathrm{H}), 3.35-3.31(\mathrm{~m}, 4 \mathrm{H}), 2.96-2.81(\mathrm{~m}, 2 \mathrm{H}), 2.25(\mathrm{~s}, 6 \mathrm{H}), 2.08-$ $1.96(\mathrm{~m}, 2 \mathrm{H}), 1.51(\mathrm{~s}, 8 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{DMSO}$ ) $\delta$ 173.21, 170.49, 168.12, 167.98, $167.43,155.34,134.34,130.12,127.56,125.38,125.17,120.79,119.06,118.31,108.47,70.74$, 70.07, 69.84, 69.52, 58.39, 49.30, 47.36, 46.95, 44.05, 35.60, 31.46, 26.42, 26.32, 22.66, 17.14. LC/MS 3.5-3.7 min, $m / z: 975.3739(\mathrm{M}+\mathrm{H})^{+}$.


To a solution of carboxylic compound $110(25 \mathrm{mg}, 0.047 \mathrm{mmol})$ in dry 0.300 mL DMF mL COMU ( $20 \mathrm{mg}, 0.047 \mathrm{mmol}$ ), compound $104(22 \mathrm{mg}, 0.047 \mathrm{mmol})$ in 0.100 mL of dry DMF and DIPEA ( $0.033 \mathrm{~mL}, 0.188 \mathrm{mmol}$ ) were added. The reaction mixture was left to stir for 1 h and monitored by LC-MS (acidic method). When completed, the crude reaction was purified by HPLC with a gradient from $5 \%$ to $90 \% \mathrm{v} / \mathrm{v}$ acetonitrile with $0.01 \% \mathrm{v} / \mathrm{v}$ aqueous solution of formic acid over 15 min to yield the title compound ( $20 \mathrm{mg}, 46 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 8.87(\mathrm{~s}, 1 \mathrm{H}), 7.66(\mathrm{~s}, 2 \mathrm{H}), 7.42(\mathrm{~d}, \mathrm{~J}=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{~d}, \mathrm{~J}=$ $8.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.76(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.60(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.69(\mathrm{~s}, 1 \mathrm{H}), 4.62-4.54(\mathrm{~m}$, $1 \mathrm{H}), 4.54-4.47(\mathrm{~m}, 2 \mathrm{H}), 4.36-4.30(\mathrm{~m}, 1 \mathrm{H}), 4.04(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.93(\mathrm{~s}, 6 \mathrm{H}), 3.90-$ $3.84(\mathrm{~m}, 2 \mathrm{H}), 3.84-3.76(\mathrm{~m}, 2 \mathrm{H}), 3.76-3.66(\mathrm{~m}, 10 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H}), 2.22(\mathrm{~s}$, $1 \mathrm{H}), 2.17-2.04(\mathrm{~m}, 1 \mathrm{H}), 1.04(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 172.93,170.68,170.33$, 166.06, 161.78, 160.08, 159.86, 155.34, 151.49, 148.76, 147.39, 138.90, 132.12, 129.98, 129.06, 128.94, 128.62, 128.09, 127.56, 124.81, 103.46, 98.15, 97.89, 71.80, 71.00, 70.48, $70.43,70.21,69.76,69.67,59.44,56.77,56.71,55.35,55.13,42.33,37.55,35.69,25.61,15.43$, 14.36. LC/MS 3.7-3.9 min, $m / z: 465.2115,929.4130(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of (2S,4R)-1-((S)-2-(2-((6-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy)hexyl)oxy)acetamido)-3,3-dimethylbutanoyl)-4-hydroxy-N-(4-(4-methylthiazol-5-yl)benzyl)pyrrolidine-2-carboxamide (42)


Compound $\mathbf{4 2}$ was obtained as a white solid ( $20.5 \mathrm{mg}, 42 \%$ ) from derivative $111(27 \mathrm{mg}, 0.055$ $\mathrm{mmol})$ and the intermediate $\mathbf{1 0 4}(26 \mathrm{mg}, 0.055 \mathrm{mmol})$ according to the procedure described for 41.
${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 8.81(\mathrm{~s}, 1 \mathrm{H}), 7.64(\mathrm{~s}, 2 \mathrm{H}), 7.45-7.39(\mathrm{~m}, 2 \mathrm{H}), 7.36(\mathrm{~d}, \mathrm{~J}=8.3$ $\mathrm{Hz}, 2 \mathrm{H}), 6.74(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.52(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.73-4.67(\mathrm{~m}, 1 \mathrm{H}), 4.62-4.55(\mathrm{~m}$, $1 \mathrm{H}), 4.53-4.47(\mathrm{~m}, 2 \mathrm{H}), 4.36-4.28(\mathrm{~m}, 1 \mathrm{H}), 3.97(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.90(\mathrm{~d}, \mathrm{~J}=1.4 \mathrm{~Hz}$, $6 \mathrm{H}), 3.80(\mathrm{dd}, \mathrm{J}=7.8,5.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.58(\mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.42(\mathrm{~s}, 3 \mathrm{H}), 2.29(\mathrm{~s}, 6 \mathrm{H}), 2.27-$ $2.20(\mathrm{~m}, 1 \mathrm{H}), 2.15-2.04(\mathrm{~m}, 1 \mathrm{H}), 1.87-1.80(\mathrm{~m}, 2 \mathrm{H}), 1.74-1.67(\mathrm{~m}, 2 \mathrm{H}), 1.62-1.52(\mathrm{~m}$, 6 H ), 1.03 ( $\mathrm{s}, 9 \mathrm{H}$ ). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta$ 172.87, 170.68, 170.61, 170.34, 165.61, $161.45,161.08,161.04,159.45,154.20,151.85,151.34,147.57,138.79,131.99,131.57$, 130.09, 129.10, 128.92, 128.23, 128.09, 127.54, 126.68, 103.96, 99.74, 97.53, 72.01, 71.49, $69.68,69.41,59.71,59.44,56.75,56.69,56.61,55.14,54.93,42.35,37.55,35.81,30.03,29.25$, 25.76, 25.65, 25.57, 15.34, 14.44. LC/MS 4.1-4.3 min, $m / z: 449.2164,897.4281(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of (2S,4R)-1-((S)-2-(2-((4-((4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-
dimethylphenoxy)methyl)benzyl)oxy)acetamido)-3,3-dimethylbutanoyl)-4-hydroxy-N-(4-(4-methylthiazol-5-yl)benzyl)pyrrolidine-2-carboxamide (43)


Compound $\mathbf{4 3}$ was obtained as a white solid ( $15 \mathrm{mg}, 36 \%$ ) from derivative $\mathbf{1 1 2}$ ( $23 \mathrm{mg}, 0.045$ $\mathrm{mmol})$ and the intermediate $\mathbf{1 0 4}$ ( $21 \mathrm{mg}, 0.045 \mathrm{mmol}$ ) according to the procedure described for 41.
${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 8.81(\mathrm{~s}, 1 \mathrm{H}), 7.70(\mathrm{~s}, 2 \mathrm{H}), 7.46(\mathrm{~d}, \mathrm{~J}=3.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.43(\mathrm{~d}, \mathrm{~J}=$ $4.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.81(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.56(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.89$ $(\mathrm{s}, 2 \mathrm{H}), 4.72(\mathrm{~d}, \mathrm{~J}=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.66(\mathrm{~d}, \mathrm{~J}=1.6 \mathrm{~Hz}, 2 \mathrm{H}), 4.58(\mathrm{~d}, \mathrm{~J}=4.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.55-4.48$ $(\mathrm{m}, 2 \mathrm{H}), 4.36-4.29(\mathrm{~m}, 1 \mathrm{H}), 4.12-4.02(\mathrm{~m}, 2 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.91-3.85(\mathrm{~m}$, $1 \mathrm{H}), 3.84-3.78(\mathrm{~m}, 1 \mathrm{H}), 2.43(\mathrm{~s}, 3 \mathrm{H}), 2.31(\mathrm{~s}, 6 \mathrm{H}), 2.27-2.21(\mathrm{~m}, 1 \mathrm{H}), 2.14-2.06(\mathrm{~m}, 1 \mathrm{H})$, $1.03(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 172.93,170.41,170.32,165.58,161.41,151.34$, 147.66, 138.84, 137.29, 137.22, 132.00, 131.76, 130.13, 128.95, 128.17, 128.06, 127.88, $127.54,100.65,97.49,73.70,72.87,69.70,68.67,59.48,56.76,55.09,54.87,42.35,37.55$, 35.81, 25.53, 15.44, 14.41. LC/MS 4.0-4.1 min, $m / z: 459.2432,917.4872(\mathrm{M}+\mathrm{H})^{+}$.

## 5-yl)benzyl)pyrrolidine-2-carboxamide (44)



Compound $\mathbf{4 4}$ was obtained as a white solid ( $15 \mathrm{mg}, 37 \%$ ) from derivative $\mathbf{1 1 3}$ ( $22.2 \mathrm{mg}, 0.042$ $\mathrm{mmol})$ and the intermediate $\mathbf{1 0 4}(19.6 \mathrm{mg}, 0.042 \mathrm{mmol})$ according to the procedure described for 41.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 8.84(\mathrm{~s}, 1 \mathrm{H}), 7.68(\mathrm{~s}, 2 \mathrm{H}), 7.45-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.38(\mathrm{~d}, J=8.3$ $\mathrm{Hz}, 2 \mathrm{H}), 6.78(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.57(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.69(\mathrm{~s}, 1 \mathrm{H}), 4.62-4.54(\mathrm{~m}, 1 \mathrm{H})$, $4.53-4.46(\mathrm{~m}, 2 \mathrm{H}), 4.36-4.30(\mathrm{~m}, 1 \mathrm{H}), 3.96(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.93(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 6 \mathrm{H})$, $3.88(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.85(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.82-3.78(\mathrm{~m}, 1 \mathrm{H}), 3.58(\mathrm{t}, J=5.8 \mathrm{~Hz}, 2 \mathrm{H})$, $3.53-3.47$ (m, 4H), $2.45(\mathrm{~s}, 3 \mathrm{H}), 2.34(\mathrm{~s}, 6 \mathrm{H}), 2.29-2.19(\mathrm{~m}, 1 \mathrm{H}), 2.14-2.07(\mathrm{~m}, 1 \mathrm{H}), 1.92$ $-1.85(\mathrm{~m}, 2 \mathrm{H}), 1.82-1.76(\mathrm{~m}, 2 \mathrm{H}), 1.74-1.68(\mathrm{~m}, 4 \mathrm{H}), 1.03(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 171.44,169.13,168.91,164.32,161.60,160.11,159.46,158.16,153.11,149.92$, $146.13,137.38,130.58,130.24,128.64,127.64,127.49,126.88,126.64,126.10,124.97$, $102.43,98.00,96.17,70.63,69.95,68.79,68.23,67.93,57.99,55.29,55.18,53.74,53.52$, 40.89, 36.09, 34.33, 25.59, 24.72, 24.65, 24.10, 13.90. LC/MS 4.1-4.3 min, $m / z: 471.2740$, $941.5496(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 5-(4-(2-(2-(2-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-
dimethylphenoxy)ethoxy)ethoxy)ethoxy)acetyl)piperazin-1-yl)-2-(2,6-dioxopiperidin-3-yl)isoindoline-

## 1,3-dione (45)



Compound $\mathbf{4 5}$ was obtained as a white solid ( $15 \mathrm{mg}, 37 \%$ ) from derivative $\mathbf{1 1 0}(25 \mathrm{mg}, 0.047$ $\mathrm{mmol})$ and the intermediate $\mathbf{1 0 5}(18 \mathrm{mg}, 0.047 \mathrm{mmol})$ according to the procedure described for 41.
${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.74(\mathrm{~s}, 1 \mathrm{H}), 7.71(\mathrm{~s}, 2 \mathrm{H}), 7.65(\mathrm{~d}, \mathrm{~J}=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.20(\mathrm{~d}, \mathrm{~J}=$ $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.03(\mathrm{dd}, \mathrm{J}=8.5,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.43(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $5.03-4.95(\mathrm{~m}, 1 \mathrm{H}), 4.27(\mathrm{~d}, \mathrm{~J}=2.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H}), 3.86(\mathrm{t}, \mathrm{J}=4.8 \mathrm{~Hz}$, $2 H), 3.83-3.75(\mathrm{~m}, 6 \mathrm{H}), 3.75-3.62(\mathrm{~m}, 8 \mathrm{H}), 3.51-3.40(\mathrm{~m}, 4 \mathrm{H}), 2.96-2.68(\mathrm{~m}, 4 \mathrm{H}), 2.18$ (s, 6H). ${ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 169.20,168.10,166.31,165.85,165.23,163.37,161.38$, $159.65,156.92,153.24,152.12,150.52,132.50,129.74,125.86,123.56,118.45,116.32$, $106.90,103.03,99.42,96.35,69.71,69.44,69.10,68.88,68.75,68.70,54.48,53.85,47.49$, 45.94, 45.44, 43.05, 42.75, 39.56, 29.72, 20.92, 14.38. LC/MS 3.5-3.7 min, $m / z: ~ 841.3401$ $(\mathrm{M}+\mathrm{H})^{+}$.


Compound $\mathbf{4 6}$ was obtained as a white solid ( $20 \mathrm{mg}, 45 \%$ ) from derivative $111(27 \mathrm{mg}, 0.055$ $\mathrm{mmol})$ and the intermediate $\mathbf{1 0 5}(21 \mathrm{mg}, 0.055 \mathrm{mmol})$ according to the procedure described for 41.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.75(\mathrm{~s}, 1 \mathrm{H}), 7.81(\mathrm{~s}, 2 \mathrm{H}), 7.69(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{~d}, J=$ $3.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.08(\mathrm{dd}, J=8.5,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.06-4.93(\mathrm{~m}, 1 \mathrm{H}), 4.27$ - $4.13(\mathrm{~m}, 2 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.87-3.75(\mathrm{~m}, 4 \mathrm{H}), 3.72(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.55(\mathrm{t}$, $J=5.8 \mathrm{~Hz}, 3 \mathrm{H}), 3.50-3.36(\mathrm{~m}, 4 \mathrm{H}), 2.96-2.71(\mathrm{~m}, 4 \mathrm{H}), 2.14(\mathrm{~s}, 6 \mathrm{H}), 1.81-1.72(\mathrm{~m}, 2 \mathrm{H})$, $1.71-1.62(\mathrm{~m}, 2 \mathrm{H}), 1.56-1.40(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 170.99,170.13,168.31$, $167.65,161.55,155.28,134.28,132.00,128.51,125.44,120.64,118.52,108.98,98.79,72.30$, $71.40,71.24,56.39,56.01,49.31,48.29,47.34,44.77,41.29,31.54,30.41,29.50,26.01,25.76$, 22.76, 16.19. LC/MS 4.0-4.2 min, $m / z: 809.3522(\mathrm{M}+\mathrm{H})^{+}$.


Compound $\mathbf{4 7}$ was obtained as a white solid ( $10 \mathrm{mg}, 27 \%$ ) from derivative $\mathbf{1 1 2}(23 \mathrm{mg}, 0.045$ $\mathrm{mmol})$ and the intermediate $\mathbf{1 0 5}(19 \mathrm{mg}, 0.045 \mathrm{mmol})$ according to the procedure described for 41.
${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.07(\mathrm{~s}, 1 \mathrm{H}), 7.74(\mathrm{~s}, 2 \mathrm{H}), 7.65(\mathrm{~d}, \mathrm{~J}=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.43(\mathrm{~d}, \mathrm{~J}=$ $8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.40(\mathrm{~d}, \mathrm{~J}=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.22(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.98(\mathrm{dd}, \mathrm{J}=8.5,2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $6.83(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.02-4.94(\mathrm{~m}, 1 \mathrm{H}), 4.74(\mathrm{~s}, 2 \mathrm{H}), 4.72-4.58$ $(\mathrm{m}, 2 \mathrm{H}), 4.34-4.23(\mathrm{~m}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.77-3.56(\mathrm{~m}, 4 \mathrm{H}), 3.31-3.16(\mathrm{~m}$, $2 \mathrm{H}), 3.14-3.05(\mathrm{~m}, 2 \mathrm{H}), 2.98-2.69(\mathrm{~m}, 4 \mathrm{H}), 2.15(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $170.93,170.21,168.22,167.60,167.01,165.30,161.53,161.43,158.79,155.24,152.48$, 137.43, 137.21, 134.23, 131.77, 129.01, 128.00, 127.71, 125.36, 120.58, 118.17, 109.10, $104.79,101.10,98.30,73.72,73.53,70.81,56.34,55.73,49.31,47.34,47.12,44.42,41.19$, 41.02, 31.59, 22.73, 16.25. LC/MS 3.9-4.0 min, $m / z: 829.3174(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 5-(4-(2-(4-(4-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl

 phenoxy)butoxy)butoxy)acetyl)piperazin-1-yl)-2-(2,6-dioxopiperidin-3-yl)isoindoline-1,3-dione (48)

Compound $\mathbf{4 8}$ was obtained as a white solid ( $11 \mathrm{mg}, 31 \%$ ) from derivative $\mathbf{1 1 3}(23 \mathrm{mg}, 0.042$ $\mathrm{mmol})$ and the intermediate $\mathbf{1 0 5}(16 \mathrm{mg}, 0.042 \mathrm{mmol})$ according to the procedure described for 41.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.61(\mathrm{~s}, 1 \mathrm{H}), 7.74(\mathrm{~s}, 2 \mathrm{H}), 7.70(\mathrm{~d}, \mathrm{~J}=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.28(\mathrm{~d}, \mathrm{~J}=$ $2.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.07(\mathrm{dd}, \mathrm{J}=8.5,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.47(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $5.04-4.96(\mathrm{~m}, 1 \mathrm{H}), 4.21(\mathrm{~d}, \mathrm{~J}=1.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.81(\mathrm{q}, \mathrm{J}=5.2 \mathrm{~Hz}$, 4H), 3.76 (t, J = 6.4 Hz, 2H), $3.58(\mathrm{t}, \mathrm{J}=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.54-3.45(\mathrm{~m}, 6 \mathrm{H}), 2.98-2.72(\mathrm{~m}, 4 \mathrm{H})$, $2.22(\mathrm{~s}, 6 \mathrm{H}), 1.92-1.85(\mathrm{~m}, 2 \mathrm{H}), 1.83-1.76(\mathrm{~m}, 2 \mathrm{H}), 1.73-1.62(\mathrm{~m}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 171.04,169.72,168.32,167.64,167.02,165.45,161.51,161.01,159.49$, $155.02,152.84,134.33,131.90,127.98,125.43,120.37,118.20,108.82,104.55,100.59,98.43$, $72.18,71.31,71.05,70.52,70.41,56.35,55.82,49.28,47.85,47.33,44.64,41.35,41.01,31.52$, 29.68, 29.14, 27.42, 26.57, 26.47, 26.40, 22.75, 16.23, 14.08. LC/MS 3.9-3.1 min, $m / z:$ $853.4184(\mathrm{M}+\mathrm{H})^{+}$.


An aqueous solution ( 3.5 mL ) of $\mathrm{LiOH}(12 \mathrm{mg}, 0.488 \mathrm{mmol})$ was added to a solution of compound $\mathbf{6 0}$ ( $105 \mathrm{mg}, 0.244 \mathrm{mmol}$ ) in 3.5 mL of EtOH and the mixture was stirring for 1 h at room temperature. Then, the mixture was concentrated in vacuo: the aqueous phase was washed with $\mathrm{CHCl}_{3}(3 \times 10 \mathrm{~mL}$ ), acidified with HCl 3 N until pH 3 and extracted with EtOAc (3 x 10 $\mathrm{mL})$. The collected organic phases were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The title compound (49) was obtained as white solid (91 $\mathrm{mg}, 93 \%)$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.64(\mathrm{dd}, \mathrm{J}=8.7,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.52-7.43(\mathrm{~m}, 2 \mathrm{H}), 7.39(\mathrm{~d}, \mathrm{~J}=$ $8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.32(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{~s}, 1 \mathrm{H}), 4.62(\mathrm{~s}, 2 \mathrm{H}), 2.59(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}:$ $402(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(4-(2-aminoethoxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (50)


To a solution of compound $\mathbf{6 5}(50 \mathrm{mg}, 0.111 \mathrm{mmol})$ in 1 mL of EtOH , hydrazine monohydrate $(0.022 \mathrm{~mL}, 0.444 \mathrm{mmol})$ was added and the resulting mixture was stirred for 6 h at $90^{\circ} \mathrm{C}$. After cooling at room temperature, the reaction mixture was concentrated under reduced pressure.

The residual semisolid was taken up with $\mathrm{HCl} 3 \mathrm{~N}(10 \mathrm{~mL})$ : the aqueous phase was washed with $\mathrm{CHCl}_{3}$ ( $3 \times 10 \mathrm{~mL}$ ), basified until pH 10 and extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ). The collected organic phases were washed with brine, dried over sodium sulfate, filtered and concentrate under reduced pressure to give the title compound as white solid ( $35 \mathrm{mg}, 85 \%$ ) ${ }^{1}{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.67(\mathrm{~s}, 2 \mathrm{H}), 6.83(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, 3.98 (s, 3H), 3.93 (s, 3H), $3.87(\mathrm{t}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.13$ (t, $J=5.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.38$ (s, 6H). MS (ESI) $m / z: 370(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(4-((8-aminooctyl)oxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (51a)



Compound 66a ( $42 \mathrm{mg}, 0.087 \mathrm{mmol}$ ) was solubilized in 3 mL of EtOH and $10 \% \mathrm{Pd} / \mathrm{C}$ was added. The mixture was stirred under a hydrogen atmosphere (balloon) for 18 h . Then, the mixture was filtered and the filtrate evaporated to give the title compound as white solid (35 mg, 89\%).
${ }^{1}{ }^{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.67(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~s}, 1 \mathrm{H}), 6.45(\mathrm{~s}, 1 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93$ (s, 3H), $3.83-3.74(\mathrm{~m}, 2 \mathrm{H}), 2.74(\mathrm{t}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}), 1.86-1.76(\mathrm{~m}, 4 \mathrm{H}), 1.56-$ $1.30(\mathrm{~m}, 8 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 454(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-(2-(2-aminoethoxy)ethoxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)one (51b)


Compound 51b was obtained as a white solid ( 55 mg , $98 \%$ ) from derivative $\mathbf{6 6 b}$ ( $60 \mathrm{mg}, 0.136$ mmol ) according to the procedure described for 51a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.66(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $4.03-4.00(\mathrm{~m}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.83(\mathrm{t}, J=4.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.61(\mathrm{t}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H})$, $2.94(\mathrm{t}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.39(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 414(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(4-(2-(2-(2-aminoethoxy)ethoxy)ethoxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin

 -4(3H)-one (51c)

Compound 51c was obtained as a white solid ( $50 \mathrm{mg}, 81 \%$ ) from derivative $\mathbf{6 6 c}(65 \mathrm{mg}, 0.134$ mmol ) according to the procedure described for 51a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.69(\mathrm{~s}, 2 \mathrm{H}), 6.81(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.44(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $4.03-3.99(\mathrm{~m}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.84(\mathrm{t}, J=4.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.76-3.70(\mathrm{~m}, 3 \mathrm{H})$, $3.69-3.63(\mathrm{~m}, 3 \mathrm{H}), 3.54(\mathrm{~d}, \mathrm{~J}=4.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.37(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 458(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-(2-(2-(2-(2-aminoethoxy)ethoxy)ethoxy)ethoxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (51d)


Compound 51d was obtained as a yellow oil ( $57 \mathrm{mg}, 99 \%$ ) from derivative $\mathbf{6 6 d}$ ( $60 \mathrm{mg}, 0.114$ mmol ) according to the procedure described for 51a.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.66(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, J=2.9 \mathrm{~Hz}, 1 \mathrm{H})$, 4.03 (t, $J=4.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.83(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.75-3.61(\mathrm{~m}$, $8 \mathrm{H}), 3.52(\mathrm{t}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.88(\mathrm{t}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.38(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 502(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-((14-amino-3,6,9,12-tetraoxatetradecyl)oxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (51e)


Compound 51e was obtained as a yellow oil ( $81 \mathrm{mg}, 99 \%$ ).) from derivative $66 \mathbf{e}(85 \mathrm{mg}, 0.149$ mmol ) according to the procedure described for 51a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.69(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $4.01(\mathrm{~m}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{t}, \mathrm{J}=4.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.78-3.52(\mathrm{~m}, 16 \mathrm{H}), 2.39(\mathrm{~s}$, 6H). MS (ESI) $m / z: 546(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(4-((17-amino-3,6,9,12,15-pentaoxaheptadecyl)oxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (51f)


Compound $\mathbf{5 1 f}$ was obtained as a yellow oil ( $45 \mathrm{mg}, 94 \%$ ) from derivative $\mathbf{6 6 f}(50 \mathrm{mg}, 0.081$ mmol ) according to the procedure described for 51a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.71(\mathrm{~s}, 2 \mathrm{H}), 6.81(\mathrm{~s}, 1 \mathrm{H}), 6.44(\mathrm{~s}, 1 \mathrm{H}), 4.02-3.99(\mathrm{~m}, 2 \mathrm{H}), 3.96$ $(\mathrm{s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.83(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.75-3.59(\mathrm{~m}, 18 \mathrm{H}), 3.53(\mathrm{t}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H})$, $2.38(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 590(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-(2-(2-(2-(4-aminophenoxy)ethoxy)ethoxy)ethoxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (52a)


Compound 52a was obtained as a yellow oil ( 40 mg , $99 \%$ ) from derivative $\mathbf{6 7 a}(43 \mathrm{mg}, 0.074$ mmol ) according to the procedure described for 51a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.74(\mathrm{~s}, 2 \mathrm{H}), 6.81(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H})$, $6.62(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.42(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{~m}, 2 \mathrm{H}), 3.97(\mathrm{~m}, 2 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H})$, $3.91(\mathrm{~s}, 3 \mathrm{H}), 3.83(\mathrm{~m}, 4 \mathrm{H}), 3.75(\mathrm{~m}, 4 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 550(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-((14-(4-aminophenoxy)-3,6,9,12-tetraoxatetradecyl)oxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (52b)


Compound 52b was obtained as a yellow oil ( $80 \mathrm{mg}, 84 \%$ ) from derivative $\mathbf{6 7 b}$ ( $100 \mathrm{mg}, 0.150$ mmol ) according to the procedure described for 51a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.63(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.75(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, $6.63(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.46(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.07-4.03(\mathrm{~m}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}$, $3 H), 3.86-3.79(\mathrm{~m}, 4 \mathrm{H}), 3.75-3.63(\mathrm{~m}, 14 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 638(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(4-(2-(1-(2-aminoethyl)-1H-1,2,3-triazol-4-yl)ethoxy)-3,5-dimethylphenyl)-5,7-

 dimethoxyquinazolin-4(3H)-one (53a)

Compound 53a was obtained as a yellow oil ( $52 \mathrm{mg}, 83 \%$ ) from derivative 86a ( $80 \mathrm{mg}, 0.134$ mmol ) according to the procedure described for $\mathbf{5 0}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.63(\mathrm{~s}, 2 \mathrm{H}), 7.55(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $6.46(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.41-4.37(\mathrm{~m}, 2 \mathrm{H}), 4.16-4.10(\mathrm{~m}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H})$, $3.32-3.18(\mathrm{~m}, 4 \mathrm{H}), 2.28(\mathrm{~s}, 6 \mathrm{H})$. MS (ESI) $m / z: 465(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-(4-(1-(2-aminoethyl)-1H-1,2,3-triazol-4-yl)butoxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (53b)


Compound 53b was obtained as a yellow oil ( $45 \mathrm{mg}, 81 \%$ ) from derivative $\mathbf{8 6 b}(70 \mathrm{mg}, 0.112$ mmol ) according to the procedure described for $\mathbf{5 0}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.67(\mathrm{~s}, 2 \mathrm{H}), 7.38(\mathrm{~s}, 1 \mathrm{H}), 6.82(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, J=$ $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.38(\mathrm{t}, J=5.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{t}, J=5.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.21$ (t, $J=5.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.83(\mathrm{t}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}), 1.96-1.86(\mathrm{~m}, 4 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}$ : $493(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of (2-amino-5-chlorophenyl)(4-chlorophenyl)methanone (56)



A stirred solution of 4-chlorobenzoyl chloride $\mathbf{5 5}(41.00 \mathrm{~mL}, 319 \mathrm{mmol})$ was heated at $120^{\circ} \mathrm{C}$ and p-chloroaniline 54 ( $5.1 \mathrm{~g}, 39.98 \mathrm{mmol}$ ) was added slowly. Once the p-chloroaniline dissolved, zinc chloride ( $5.45 \mathrm{~g}, 39.98 \mathrm{mmol}$ ) was added, increasing the temperature up to 200$230{ }^{\circ} \mathrm{C}$. The reaction mixture was heated at reflux for 4 hours, after which the reaction mixture was cooled to $120{ }^{\circ} \mathrm{C}$ and washed with water. The residual semisolid was dissolved with a mixture of sulfuric acid $(250 \mathrm{~mL})$, acetic acid $(120 \mathrm{~mL})$ and water $(120 \mathrm{~mL})$ and the solution was heated at reflux for 5 hours. Then, the reaction mixture was poured into ice-water, extracted with dichloromethane ( $3 \times 200 \mathrm{~mL}$ ) and washed with water ( 3 x 100 mL ), $15 \%$ aqueous ammonium hydroxide solution ( $3 \times 100 \mathrm{~mL}$ ) and brine.

The organic phase was dried over sodium sulfate, filtered and evaporated under reduced pressure. Compound 56 was obtained as yellow solid ( $7.45 \mathrm{~g}, 70 \%$ ) after crystallization with EtOH
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.59(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.46(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{~d}, J=$ $2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.26(\mathrm{dd}, J=2.5 \mathrm{~Hz} 1 \mathrm{H}), 6.70(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.06(\mathrm{~s}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z:$ $266(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of ethyl (E)-(((2-amino-5-chlorophenyl)(4-chlorophenyl) methylene) amino) glycinate (57)


A mixture of compound $\mathbf{5 6}(1 \mathrm{~g}, 4.42 \mathrm{mmol})$ and ethyl hydrazinoacetate hydrochloride (889 $\mathrm{mg}, 5.75 \mathrm{mmol}$ ) was refluxed in 10 mL of dry EtOH for 18 h . After cooling, EtOAc and saturated solution of $\mathrm{NaHCO}_{3}$ were added and the mixture was stirred for 10 min . Then, the aqueous phase was extracted with EtOAc ( $3 \times 15 \mathrm{~mL}$ ): the organic phases were collected, washed with brine, dried over sodium sulfate, filtered and concentrated. The product 57 was obtained after purification on silica gel flash-chromatography column (hex-EtOAc 90:10 to $60: 40)$ to yield a yellow oil ( $1.27 \mathrm{~g}, 79 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.50(\mathrm{~d}, J=1.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.47(\mathrm{~d}, 2 \mathrm{H}), 7.46(\mathrm{~s}, 1 \mathrm{H}), 7.45(\mathrm{~d}, J=$ $1.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.19(\mathrm{q}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.81(\mathrm{~s}, 2 \mathrm{H}), 1.25(\mathrm{t}, J=8.0 \mathrm{~Hz}$, 3H). MS (ESI) $m / z: 366(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of ethyl 2-(7-chloro-5-(4-chlorophenyl)-2-oxo-1,2-dihydro-3H-benzo[e][1,2,4]triazepin-3-

 yl)acetate (58)

Compound 57 ( $888 \mathrm{mg}, 2.42 \mathrm{mmol}$ ) was dissolved in 13.30 mL of dry DCM and TEA ( 0.506 $\mathrm{mL}, 3.63 \mathrm{mmol})$ was added. After cooling at $0{ }^{\circ} \mathrm{C},\left(\mathrm{CClO}_{3}\right)_{2} \mathrm{CO}(359 \mathrm{mg}, 1.21 \mathrm{mmol})$ was added and the reaction mixture was stirring for 18 h at room temperature and, then, the solvent was evaporated. The crude was purified on a silica gel flash-chromatography column (hex-EtOAc 90:10 to 70:30) to yield compound $\mathbf{5 8}$ as white solid ( $800 \mathrm{mg}, 85 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.66(\mathrm{~s}, 1 \mathrm{H}), 7.43(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.38(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H})$, $7.03(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.45(\mathrm{~s}, 2 \mathrm{H}), 4.23(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.27$ $(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 392(\mathrm{M}+\mathrm{H})^{+}$ yl)acetate (59)


Lawesson's reagent ( $1.54 \mathrm{~g}, 3.82 \mathrm{mmol}$ ) was added to a solution of compound $\mathbf{5 8}(750 \mathrm{mg}, 1.91$ mmol ) in 50 mL of toluene and the resulting mixture was heated at reflux for 18 h . Then, the solvent was evaporated and the crude was purified with a silica gel flash chromatography column (hex-EtOAc 95:5 to 70:30) to yield the title compound (59) as yellow oil ( 440 mg , 55\%).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.59(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.55(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{~d}, J=$ $8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.40(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.33(\mathrm{dd}, J=7.6,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.62(\mathrm{~s}, 2 \mathrm{H}), 4.19(\mathrm{q}, J=$ $7.9 \mathrm{~Hz}, 2 \mathrm{H}), 1.25(\mathrm{t}, J=8.0 \mathrm{~Hz}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 408(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of ethyl 2-(8-chloro-6-(4-chlorophenyl)-1-methyl-4H-benzo[e][1,2,4]triazolo[3,4-c][1,2,4]

 triazepin-4-yl)acetate (60)

To a solution of compound 59 ( $499 \mathrm{mg}, 1.17 \mathrm{mmol}$ ) in 20 mL of THF/AcOH (4:1,) acethydrazide ( $173 \mathrm{mg}, 2.34 \mathrm{mmol}$ ) and $\mathrm{Hg}(\mathrm{OAc})_{2}(561 \mathrm{mg}, 1.76 \mathrm{mmol})$ were added. The mixture was heated at $95^{\circ} \mathrm{C}$ and stirring for 4 h . The crude reaction was filtered through a pad
of Celite, which was rinsed with MeOH . The filtrate was evaporated and purified with a silica gel flash chromatography column (hex-EtOAc 20:80 to 0:100) to yield a yellow oil (60) (146 $\mathrm{mg}, 30 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.66(\mathrm{~s}, 1 \mathrm{H}), 7.43(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.38(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H})$, $7.03(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 4.45(\mathrm{~s}, 2 \mathrm{H}), 4.23(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.82$ (s, 3H) $1.27(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 430(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of 4-(2-(1,3-dioxoisoindolin-2-yl)ethoxy)-3,5-dimethylbenzaldehyde (61)



To a solution of $\mathbf{6 9}$ ( $540 \mathrm{mg}, 2 \mathrm{mmol}$ ) in 8 mL of dry DMF, 4-hydroxy-3,5-dimethylbenzaldehyde 76a ( $330 \mathrm{mg}, 2.2 \mathrm{mmol}$ ) and $\mathrm{K}_{2} \mathrm{CO}_{3}(304 \mathrm{mg}, 2.2 \mathrm{mmol})$ were added under nitrogen atmosphere. The resulting mixture was stirred ad $80^{\circ} \mathrm{C}$ for 18 h . Then, water ( 20 mL ) was added and the aqueous phase was extracted with EtOAc ( $3 \times 15 \mathrm{~mL}$ ). The collected organic phases were washed with saturated solution of $\mathrm{NaHCO}_{3}(3 \times 15 \mathrm{~mL})$, brine, dried over sodium sulfate, filtered and concentrated under reduced pressure. The title compound was obtained as yellow solid after purification on silica gel flash chromatography column (hex-EtOAc 80:20 to 70:30) ( $330 \mathrm{mg}, 51 \%$ ).
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.85(\mathrm{~s}, 1 \mathrm{H}), 7.89(\mathrm{dd}, J=5.4,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.75(\mathrm{dd}, J=5.4$, $3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{~s}, 2 \mathrm{H}), 4.18-4.13(\mathrm{~m}, 2 \mathrm{H}), 4.12-4.07(\mathrm{~m}, 2 \mathrm{H}), 2.25(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}:$ $324(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-((8-azidooctyl)oxy)-3,5-dimethylbenzaldehyde (62a)



Compound 62a was obtained as a pale yellow solid ( $53 \mathrm{mg}, 73 \%$ ) from derivative 74a ( 80 mg , $0.240 \mathrm{mmol})$ and $76 \mathbf{(} 40 \mathrm{mg}, 0.264 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.87(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{~s}, 2 \mathrm{H}), 3.82(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.31-3.23$ $(\mathrm{m}, 2 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}), 1.88-1.77(\mathrm{~m}, 2 \mathrm{H}), 1.66-1.58(\mathrm{~m}, 2 \mathrm{H}), 1.56-1.48(\mathrm{~m}, 8 \mathrm{H}) . \mathrm{MS}$ (ESI) $m / z: 304(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-(2-(2-azidoethoxy)ethoxy)-3,5-dimethylbenzaldehyde (62b)



Compound 62b was obtained as a pale yellow solid (134 mg, 73\%) from derivative 74b (200 $\mathrm{mg}, 0.700 \mathrm{mmol})$ and $76 \mathbf{a}(116 \mathrm{mg}, 0.770 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.88(\mathrm{~s}, 1 \mathrm{H}), 7.56(\mathrm{~s}, 2 \mathrm{H}), 4.05-4.00(\mathrm{~m}, 2 \mathrm{H}), 3.89-3.84(\mathrm{~m}$, $2 \mathrm{H}), 3.76(\mathrm{t}, J=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.44(\mathrm{t}, J=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 264(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 4-(2-(2-(2-azidoethoxy)ethoxy)ethoxy)-3,5-dimethylbenzaldehyde (62c)


Compound 62c was obtained as a pale yellow solid ( 91 mg , 78\%) from derivative 74c ( 125 mg , 0.379 mmol ) and $76 \mathbf{a}(63 \mathrm{mg}, 0.417 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.88(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{~s}, 2 \mathrm{H}), 4.04-3.99(\mathrm{~m}, 2 \mathrm{H}), 3.87-3.84(\mathrm{~m}$, $2 \mathrm{H}), 3.78-3.74(\mathrm{~m}, 2 \mathrm{H}), 3.73-3.66(\mathrm{~m}, 4 \mathrm{H}), 3.39(\mathrm{t}, J=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H})$.

MS (ESI) $m / z: 308(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-(2-(2-(2-(2-azidoethoxy)ethoxy)ethoxy)ethoxy)-3,5-dimethylbenzaldehyde (62d)



Compound 62d was obtained as a pale yellow solid ( 207 mg , $85 \%$ ) from derivative 74d (260 $\mathrm{mg}, 0.690 \mathrm{mmol})$ and $\mathbf{7 6 a}(114 \mathrm{mg}, 0.760 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$. ${ }^{1}{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.87(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{~s}, 2 \mathrm{H}), 4.01(\mathrm{dd}, \mathrm{J}=5.8,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.84$ $(\mathrm{dd}, \mathrm{J}=5.9,3.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.76-3.70(\mathrm{~m}, 10 \mathrm{H}), 3.42-3.34(\mathrm{~m}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI})$ $m / z: 352(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-((14-azido-3,6,9,12-tetraoxatetradecyl)oxy)-3,5-dimethylbenzaldehyde (62e)



Compound 62e was obtained as a pale yellow solid ( 160 mg , $87 \%$ ) from derivative 74e (192 $\mathrm{mg}, 0.460 \mathrm{mmol})$ and $\mathbf{7 6 a}(76 \mathrm{mg}, 0.506 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.87(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{~s}, 2 \mathrm{H}), 4.01(\mathrm{dd}, \mathrm{J}=5.7,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.84$ (dd, J = 5.7, 3.7 Hz, 2H), 3.72-3.65 (m, 14H), $3.38(\mathrm{t}, \mathrm{J}=5.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}$ (ESI) $m / z: 396(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-((17-azido-3,6,9,12,15-pentaoxaheptadecyl)oxy)-3,5-dimethylbenzaldehyde (62f)



Compound 62 f was obtained as a pale yellow solid ( 150 mg , 63\%) from derivative $\mathbf{7 4 f}$ ( 250 $\mathrm{mg}, 0.541 \mathrm{mmol})$ and $7 \mathbf{7 a}(89 \mathrm{mg}, 0.595 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.87(\mathrm{~s}, 1 \mathrm{H}), 7.54(\mathrm{~s}, 2 \mathrm{H}), 4.02-3.98(\mathrm{~m}, 2 \mathrm{H}), 3.86-3.80(\mathrm{~m}$, 2H), $3.75-3.59(\mathrm{~m}, 20 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H})$. MS (ESI) $m / z: 440(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of 3,5-dimethyl-4-(2-(2-(2-(4-nitrophenoxy)ethoxy)ethoxy)ethoxy)benzaldehyde (63a)



Compound 63a was obtained as a pale yellow solid (174 mg, 87\%) from derivative 75a (212 $\mathrm{mg}, 0.498 \mathrm{mmol})$ and $\mathbf{7 6 a}(82 \mathrm{mg}, 0.548 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.85(\mathrm{~s}, 1 \mathrm{H}), 8.16(\mathrm{~d}, \mathrm{~J}=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.52(\mathrm{~s}, 2 \mathrm{H}), 6.96(\mathrm{~d}, \mathrm{~J}=$ $9.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.26-4.18(\mathrm{~m}, 2 \mathrm{H}), 4.02-3.98(\mathrm{~m}, 2 \mathrm{H}), 3.92-3.88(\mathrm{~m}, 2 \mathrm{H}), 3.86-3.82(\mathrm{~m}$, 2H), $3.76(\mathrm{~s}, 4 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 404(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 3,5-dimethyl-4-((14-(4-nitrophenoxy)-3,6,9,12-tetraoxatetradecyl)oxy)benzaldehyde (63b)



Compound 63b was obtained as a pale yellow solid ( 137 mg , $85 \%$ ) from derivative 75b (169 $\mathrm{mg}, 0.329 \mathrm{mmol}$ ) and $\mathbf{7 6 a}(54 \mathrm{mg}, 0.362 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.87(\mathrm{~s}, 1 \mathrm{H}), 8.18(\mathrm{~d}, \mathrm{~J}=9.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.54(\mathrm{~s}, 2 \mathrm{H}), 6.97(\mathrm{~d}, \mathrm{~J}=$ $9.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.26-4.16(\mathrm{~m}, 2 \mathrm{H}), 4.05-3.95(\mathrm{~m}, 2 \mathrm{H}), 3.93-3.84(\mathrm{~m}, 2 \mathrm{H}), 3.88-3.79(\mathrm{~m}$, 2H), 3.77 - 3.65 (m, 12H), $2.34(\mathrm{~s}, 6 \mathrm{H})$. MS (ESI) $m / z: 492(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-amino-4,6-dimethoxybenzamide (64)



EDC ( $1.05 \mathrm{~g}, 5.47 \mathrm{mmol}$ ), $\mathrm{HOBt}(740 \mathrm{mg}, 5.47 \mathrm{mmol})$ and $\mathrm{NMM}(553 \mathrm{mg}, 5.47 \mathrm{mmol})$ were added to a solution of compound $\mathbf{8 0}(720 \mathrm{mg}, 3.65 \mathrm{mmol})$ in 35 mL of dry THF and the mixture was stirred for 4 h . Then, 0.500 mL of hydroxide ammonium solution ( $33 \% \mathrm{NH}_{3}$ in water) were added and the solution was stirred for 1 h . Water was added and the aqueous layer was extracted with EtOAc. The organic layers were collected, washed with saturated solution of $\mathrm{NaHCO}_{3}$ and brine, dried and concentrated in under reduced pressure. The product was obtained after purification on silica gel flash chromatography column (DCM-MeOH 100:0 to 90:10) yielding the title compound as yellow solid ( $464 \mathrm{mg}, 65 \%$ )
${ }^{1}$ H NMR (400 MHz, DMSO-d6) $\delta 7.46$ (s, 1H), 7.02 (s, 1H), 6.90 (s, 2H), 5.88 (d, J = 2.4 Hz, $1 \mathrm{H}), 5.76(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 3.69(\mathrm{~s}, 3 \mathrm{H})$. MS (ESI) m/z: $197(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy)ethyl)

## isoindoline-1,3-dione (65)



To a stirred solution of $\mathbf{6 1}(180 \mathrm{mg}, 0.556 \mathrm{mmol})$ in 3.5 mL of dry DMAc, 2-amino-4,6dimethoxybenzamide $64(100 \mathrm{mg}, 0.556 \mathrm{mmol}), \mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{5}(139 \mathrm{mg}, 0.667 \mathrm{mmol}), \mathrm{TsOH}(21$ $\mathrm{mg}, 0.111 \mathrm{mmol}$ ) were added. The resulting mixture was stirred at $120^{\circ} \mathrm{C}$ for 18 h . After cooling at room temperature, water ( 30 mL ) was added. The aqueous phase was extracted with EtOAc ( $3 \times 15 \mathrm{~mL}$ ) and the collected organic phases were washed with saturated solution of
$\mathrm{NaHCO}_{3}(3 \times 15 \mathrm{~mL})$, brine, dried over sodium sulfate and filtered. After evaporation, the crude was purified with a silica gel flash chromatography column (DCM-MeOH 95:5 to 90:10) to give a pale yellow solid ( $158 \mathrm{mg}, 63 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.04(\mathrm{~s}, 1 \mathrm{H}), 7.90(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.76(\mathrm{dd}, J=5.5$, $3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.61(\mathrm{~s}, 2 \mathrm{H}), 6.81(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.16(\mathrm{t}, J=5.7 \mathrm{~Hz}$, $2 \mathrm{H}), 4.08(\mathrm{t}, J=5.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 2.28(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 500(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(4-((8-azidooctyl)oxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (66a)



Compound 66a was obtained as a pale yellow solid ( $42 \mathrm{mg}, 53 \%$ ) from derivative $\mathbf{6 2 a}(50 \mathrm{mg}$, $0.165 \mathrm{mmol})$ and $\mathbf{6 4}(32 \mathrm{mg}, 0.165 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.34(\mathrm{~s}, 1 \mathrm{H}), 7.67(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, J=$ $2.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.80(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.27(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.36$ $(\mathrm{s}, 6 \mathrm{H}), 1.87-1.78(\mathrm{~m}, 2 \mathrm{H}), 1.55-1.48(\mathrm{~m}, 2 \mathrm{H}), 1.39(\mathrm{~d}, J=4.1 \mathrm{~Hz}, 8 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 480$ $(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-(2-(2-azidoethoxy)ethoxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (66b)


Compound 66b was obtained as a pale yellow solid (130 mg, 58\%) from derivative 71b (134 $\mathrm{mg}, 0.510 \mathrm{mmol})$ and $\mathbf{6 4}(100 \mathrm{mg}, 0.510 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.28(\mathrm{~s}, 1 \mathrm{H}), 7.67(\mathrm{~s}, 2 \mathrm{H}), 6.83(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, J=$ $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.01(\mathrm{t}, J=4.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.87(\mathrm{t}, J=4.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.77$ $(\mathrm{t}, J=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.45(\mathrm{t}, J=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.39(\mathrm{~s}, 6 \mathrm{H}) \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 440(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(4-(2-(2-(2-azidoethoxy)ethoxy)ethoxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-

 4(3H)-one (66c)

Compound 66c was obtained as a pale yellow solid ( $65 \mathrm{mg}, 46 \%$ ) from derivative $\mathbf{6 2 c}(91 \mathrm{mg}$, 0.296 mmol ) and $\mathbf{6 4}(58 \mathrm{mg}, 0.296 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.66(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H})$, $4.03-3.99(\mathrm{~m}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.88-3.85(\mathrm{~m}, 2 \mathrm{H}), 3.78-3.75(\mathrm{~m}, 2 \mathrm{H}), 3.74$ $-3.69(\mathrm{~m}, 4 \mathrm{H}), 3.40(\mathrm{t}, J=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.38(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 484(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-(2-(2-(2-(2-azidoethoxy)ethoxy)ethoxy)ethoxy)-3,5-dimethylphenyl)-5,7-dimethoxy quinazolin-4(3H)-one (66d)


Compound 66d was obtained as a pale yellow solid ( 200 mg , 70\%) from derivative 62d (190 $\mathrm{mg}, 0.541 \mathrm{mmol})$ and $\mathbf{6 4}(106 \mathrm{mg}, 0.541 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.39(\mathrm{~s}, 1 \mathrm{H}), 7.68(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, J=$ $2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.03-3.99(\mathrm{~m}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{t}, J=4.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.76-$ $3.65(\mathrm{~m}, 10 \mathrm{H}), 3.39(\mathrm{t}, J=5.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.38(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 528(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-((14-azido-3,6,9,12-tetraoxatetradecyl)oxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (66e)


Compound 66e was obtained as a pale yellow solid ( 160 mg , 40\%) from derivative 62e (160 $\mathrm{mg}, 0.404 \mathrm{mmol})$ and $\mathbf{6 4}(79 \mathrm{mg}, 0.404 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.66(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $3.98(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.84(\mathrm{t}, J=4.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.78-3.62(\mathrm{~m}, 16 \mathrm{H}), 3.39(\mathrm{t}, J=5.1 \mathrm{~Hz}$, 2H), $2.38(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 571(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-((17-azido-3,6,9,12,15-pentaoxaheptadecyl)oxy)-3,5-dimethylphenyl)-5,7-dimethoxy quinazolin-4(3H)-one (66f)


Compound $\mathbf{6 6 f}$ was obtained as a pale yellow solid ( 130 mg , $62 \%$ ) from derivative $\mathbf{6 2 f}$ ( 150 $\mathrm{mg}, 0.341 \mathrm{mmol})$ and $64(67 \mathrm{mg}, 0.341 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.14(\mathrm{~s}, 1 \mathrm{H}), 7.66(\mathrm{~s}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, J=$ $2.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.03-3.99(\mathrm{~m}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.84(\mathrm{t}, J=4.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.74-$ $3.64(\mathrm{~m}, 18 \mathrm{H}), 3.42-3.34(\mathrm{~m}, 2 \mathrm{H}), 2.38(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 616(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(3,5-dimethyl-4-(2-(2-(2-(4-nitrophenoxy)ethoxy)ethoxy)ethoxy)phenyl)-5,7-dimethoxy quinazolin-4(3H)-one (67a)


Compound 67a was obtained as a pale yellow solid (43 mg, $64 \%$ ) from derivative $\mathbf{6 3 a}$ ( 47 mg , $0.116 \mathrm{mmol})$ and $\mathbf{6 4}(23 \mathrm{mg}, 0.116 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$.
${ }^{1}{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.14(\mathrm{~d}, \mathrm{~J}=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.78(\mathrm{~s}, 2 \mathrm{H}), 6.93(\mathrm{~d}, \mathrm{~J}=9.2 \mathrm{~Hz}, 2 \mathrm{H})$, $6.80(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.42(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.23-4.16(\mathrm{~m}, 2 \mathrm{H}), 3.97(\mathrm{dd}, \mathrm{J}=5.8,3.6$ $\mathrm{Hz}, 2 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.88(\mathrm{dd}, \mathrm{J}=3.7,2.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.84-3.80(\mathrm{~m}, 2 \mathrm{H}), 3.76(\mathrm{~s}$, 4H), $2.35(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 580(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(3,5-dimethyl-4-((14-(4-nitrophenoxy)-3,6,9,12-tetraoxatetradecyl)oxy)phenyl)-5,7-dimethoxyquinazolin-4(3H)-one (67b)


Compound 67b was obtained as a pale yellow solid ( 503 mg , $80 \%$ ) from derivative 63b (464 $\mathrm{mg}, 0.944 \mathrm{mmol})$ and $\mathbf{6 4}(185 \mathrm{mg}, 0.944 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.16(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.67(\mathrm{~s}, 2 \mathrm{H}), 6.94(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H})$, $6.81(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.26-4.16(\mathrm{~m}, 2 \mathrm{H}), 4.03-3.98(\mathrm{~m}, 2 \mathrm{H})$, $3.98(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.91-3.87(\mathrm{~m}, 2 \mathrm{H}), 3.83(\mathrm{~m}, 2 \mathrm{H}), 3.75-3.64(\mathrm{~m}, 12 \mathrm{H}), 2.37(\mathrm{~s}$, $6 \mathrm{H})$. MS (ESI) $m / z: 668(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(1,3-dioxoisoindolin-2-yl)ethyl methanesulfonate (69)



Compound $68(300 \mathrm{mg}, 1.57 \mathrm{mmol})$ was dissolved in 15 mL of dry DCM and TEA $(0.544 \mathrm{~mL}$, $3.45 \mathrm{mmol})$ was added. The solution was cooling at $0^{\circ} \mathrm{C}$ and mesyl chloride $(0.236 \mathrm{~mL}, 3.45$ mmol ) was added. The reaction mixture was stirring for 2 h at room temperature under nitrogen atmosphere. Then, the mixture was diluted with DCM (10mL) and the organic phase was washed with brine ( $3 \times 20 \mathrm{~mL}$ ), dried over sodium sulfate, filtered and concentrated under reduced pressure. The product ( 69 ) was obtained as white solid ( $470 \mathrm{mg}, 99 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.88(\mathrm{dd}, \mathrm{J}=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.75(\mathrm{dd}, \mathrm{J}=5.5,3.1 \mathrm{~Hz}, 2 \mathrm{H})$, $4.49(\mathrm{t}, \mathrm{J}=5.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.05(\mathrm{t}, \mathrm{J}=5.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.02(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 270(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of 8-hydroxyoctyl 4-methylbenzenesulfonate (71a)



To an ice-cold solution of 1,8 -octanediol 70a ( $500 \mathrm{mg}, 3.42 \mathrm{mmol}$ ) and $\mathrm{Ag}_{2} \mathrm{O}(1.19 \mathrm{~g}, 5.13$ $\mathrm{mmol})$ in 31 mL of toluene, $\mathrm{TsCl}(717 \mathrm{mg}, 3.76 \mathrm{mmol})$ and $\mathrm{KI}(57 \mathrm{mg}, 0.342 \mathrm{mmol})$ were added. The resulting mixture was stirred at room temperature for 18 h . The precipitated silver salt was removed by filtration on silica gel, which was rinsed with EtOAc. The filtered was concentrated under reduced pressure. Then, the crude was taken up with EtOAc ( 40 mL ). The organic phase was washed with water ( $3 \times 20 \mathrm{~mL}$ ), brine ( 40 mL ), dried over sodium sulfate,
filtered and concentrated. The residue was purified by silica gel chromatography (hex-EtOAc 20:80 to $100 \%$ ) to give the title compound 71a as a pale yellow oil ( $600 \mathrm{mg}, 58 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.77(\mathrm{~d}, \mathrm{~J}=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.33(\mathrm{~d}, \mathrm{~J}=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.03(\mathrm{t}, \mathrm{J}=6.3$ $\mathrm{Hz}, 2 \mathrm{H}), 3.65(\mathrm{t}, \mathrm{J}=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}), 1.63-1.50(\mathrm{~m}, 4 \mathrm{H}) 1.25-1.27(\mathrm{~m}, 8 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI})$ $m / z: 301(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(2-hydroxyethoxy)ethyl 4-methylbenzenesulfonate (71b)


Compound 71b was obtained as a colorless oil ( $1.85 \mathrm{~g}, 75 \%$ ) from derivative 70b ( $1 \mathrm{~g}, 9.42$ mmol ) according to the procedure described for 71a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.80(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.20(\mathrm{t}, \mathrm{J}=4.6$ Hz, 2H), $3.73-3.65(\mathrm{~m}, 4 \mathrm{H}), 3.56-3.51(\mathrm{~m}, 2 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 261(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(2-(2-hydroxyethoxy)ethoxy)ethyl 4-methylbenzenesulfonate (71c)


Compound 71c was obtained as a colorless oil ( $2.43 \mathrm{~g}, 67 \%$ ) from derivative 70c ( $1.8 \mathrm{~g}, 11.98$ $\mathrm{mmol})$ according to the procedure described for 71a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.80(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.17(\mathrm{t}, J=$ $4.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.71(\mathrm{~d}, J=4.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.64-3.53(\mathrm{~m}, 8 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 305$ $(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(2-(2-(2-hydroxyethoxy)ethoxy)ethoxy)ethyl 4-methylbenzenesulfonate (71d)


Compound 71d was obtained as a pale yellow oil ( $1.47 \mathrm{~g}, 82 \%$ ) from derivative $70 \mathrm{~d}(1 \mathrm{~g}, 5.15$ mmol ) according to the procedure described for 71a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.82-7.78(\mathrm{~d}, \mathrm{~J}=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, \mathrm{~J}=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 4.19-$ $4.11(\mathrm{~m}, 2 \mathrm{H}), 3.72-3.55(\mathrm{~m}, 14 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 349(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 14-hydroxy-3,6,9,12-tetraoxatetradecyl 4-methylbenzenesulfonate (71e)


Compound 71e was obtained as a pale yellow oil ( $2.17 \mathrm{~g}, 66 \%$ ) from derivative 70e ( $2 \mathrm{~g}, 8.4$ mmol ) according to the procedure described for 71a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.80(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.19-4.12$ (m, 2H), $3.73-3.55(\mathrm{~m}, 18 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 393(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 17-hydroxy-3,6,9,12,15-pentaoxaheptadecyl 4-methylbenzenesulfonate (71f)


Compound 71f was obtained as a pale yellow oil ( $410 \mathrm{mg}, 53 \%$ ) from derivative $\mathbf{7 0 f}$ ( 500 mg , 1.77 mmol ) according to the procedure described for 71a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.79(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.16-4.11$ $(\mathrm{m}, 2 \mathrm{H}), 3.67-3.55(\mathrm{~m}, 22 \mathrm{H}), 2.43(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 437(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 8-azidooctan-1-ol (72a)



To a solution of 71a ( $600 \mathrm{mg}, 2 \mathrm{mmol}$ ) in 7 mL of DMF, sodium azide ( $390 \mathrm{mg}, 6 \mathrm{mmol}$ ) was added and the mixture was stirred for 18 h . Then, water ( 30 mL ) was added: the aqueous phase was extracted with EtOAc ( $3 \times 15 \mathrm{~mL}$ ) and the collected organic phases were washed with a saturated solution of $\mathrm{NaHCO}_{3}(3 \times 15 \mathrm{~mL})$, brine and dried over sodium sulfate. After filtration and evaporation, the product was obtained as pale yellow oil ( $240 \mathrm{mg}, 70 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.63(\mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.25(\mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.61-1.52(\mathrm{~m}$, 4H), 1.39-1.32 (m, 8H). MS (ESI) m/z: $172(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(2-azidoethoxy)ethan-1-ol (72b)



Compound 72b was obtained as a colorless oil ( 250 mg , $50 \%$ ) from derivative $\mathbf{7 1 b}(1 \mathrm{~g}, 3.84$ mmol ) according to the procedure described for 72a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.76(\mathrm{t}, J=4.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.70(\mathrm{t}, J=4.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.64-3.59(\mathrm{~m}$, $2 \mathrm{H}), 3.41(\mathrm{t}, \mathrm{J}=4.9 \mathrm{~Hz}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 132(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(2-(2-azidoethoxy)ethoxy)ethan-1-ol (72c)



Compound 72c was obtained as a colorless oil ( $122 \mathrm{mg}, 53 \%$ ) from derivative 71c ( 400 mg , 1.31 mmol ) according to the procedure described for 72a
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.74(\mathrm{t}, J=4.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.71-3.66(\mathrm{~m}, 6 \mathrm{H}), 3.64-3.60(\mathrm{~m}$, 2H), $3.40(\mathrm{t}, J=5.0 \mathrm{~Hz}, 2 \mathrm{H})$. MS (ESI) $m / z: 176(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(2-(2-(2-azidoethoxy)ethoxy)ethoxy)ethan-1-ol (72d)



Compound 72d was obtained as a pale yellow solid ( $573 \mathrm{mg}, 62 \%$ ) from derivative 71d (1.47 $\mathrm{g}, 4.22 \mathrm{mmol})$ according to the procedure described for 72a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.73(\mathrm{t}, J=4.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.69-3.66(\mathrm{~m}, 10 \mathrm{H}), 3.62(\mathrm{~d}, J=4.6$ $\mathrm{Hz}, 2 \mathrm{H}), 3.42-3.36(\mathrm{~m}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 220(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 14-azido-3,6,9,12-tetraoxatetradecan-1-ol (72e)


Compound 72e was obtained as a pale yellow oil ( $411 \mathrm{mg}, 62 \%$ ) from derivative $71 \mathrm{e}(1 \mathrm{~g}, 2.55$ mmol ) according to the procedure described for 72a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.73-3.70(\mathrm{~m}, 2 \mathrm{H}), 3.69-3.63(\mathrm{~m}, 12 \mathrm{H}), 3.62-3.58(\mathrm{~m}, 4 \mathrm{H})$, $3.41-3.36(\mathrm{~m}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 264(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 17-azido-3,6,9,12,15-pentaoxaheptadecan-1-ol (72f)



Compound $\mathbf{7 2 f}$ was obtained as a colorless oil ( $122 \mathrm{mg}, 53 \%$ ) from derivative $\mathbf{7 1 f}(400 \mathrm{mg}$, 1.31 mmol ) according to the procedure described for 72a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.73(\mathrm{t}, J=4.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.70-3.65(\mathrm{~m}, 18 \mathrm{H}), 3.58(\mathrm{t}, J=4.4$ $\mathrm{Hz}, 2 \mathrm{H}), 3.38(\mathrm{t}, J=5.1 \mathrm{~Hz}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 308(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(2-(2-(4-nitrophenoxy)ethoxy)ethoxy)ethan-1-ol (73a)


4-nitrophenol ( $640 \mathrm{mg}, 4.6 \mathrm{mmol}$ ) and $\mathrm{K}_{2} \mathrm{CO}_{3}(635 \mathrm{mg}, 4.6 \mathrm{mmol})$ were added to a solution of 71c ( $700 \mathrm{mg}, 2.30 \mathrm{mmol}$ ) in 2.5 mL of dry DMF and the mixture was stirring at $85^{\circ} \mathrm{C}$ for 18 h. Then, $\mathrm{H}_{2} \mathrm{O}(30 \mathrm{~mL})$ was added: the aqueous phase was extracted by EtOAc ( 3 x 15 mL ). The collected organic phases were washed with saturated solution of $\mathrm{K}_{2} \mathrm{CO}_{3}(3 \times 15 \mathrm{~mL})$, brine and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. Filtration and evaporation gave compound 73a as yellow oil ( $656 \mathrm{mg}, 2.30$ mmol, $99 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.19(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.98(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.24-4.21$ $(\mathrm{m}, 2 \mathrm{H}), 3.92-3.88(\mathrm{~m}, 2 \mathrm{H}), 3.75-3.68(\mathrm{~m}, 6 \mathrm{H}), 3.63-3.60(\mathrm{~m}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 272$ $(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 14-(4-nitrophenoxy)-3,6,9,12-tetraoxatetradecan-1-ol (73b)



Compound 73b was obtained as a pale yellow solid ( $883 \mathrm{mg}, 96 \%$ ) from derivative 71e ( 857 $\mathrm{mg}, 2.18 \mathrm{mmol}$ ) according to the procedure described for 73a.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.19(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.98(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.26-4.18$ $(\mathrm{m}, 2 \mathrm{H}), 3.92-3.84(\mathrm{~m}, 2 \mathrm{H}), 3.77-3.68(\mathrm{~m}, 2 \mathrm{H}), 3.71-3.62(\mathrm{~m}, 12 \mathrm{H}), 3.64-3.57(\mathrm{~m}, 2 \mathrm{H})$. MS (ESI) $m / z: 360(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 8-azidooctyl 4-methylbenzenesulfonate (74a)



To a solution of 72a ( $400 \mathrm{mg}, 2.34 \mathrm{mmol}$ ) in 26 mL of dry DCM, TEA ( $0.391 \mathrm{~mL}, 2.81 \mathrm{mmol}$ ) and $\mathrm{TsCl}(536 \mathrm{mg}, 2.81 \mathrm{mmol})$ were added. The resulting mixture was stirred for 18 h at room temperature under nitrogen atmosphere. Then, DCM was added $(40 \mathrm{~mL})$ : the organic phase was washed with $\mathrm{HCl} 1 \mathrm{~N}(3 \times 20 \mathrm{~mL})$, saturated solution of $\mathrm{NaHCO}_{3}(3 \times 20 \mathrm{~mL})$, brine and dried over sodium sulfate. After filtration and evaporation, the crude was purified with a silica gel flash chromatography column (hex-EtOAc 80:20 to 50:50) to give a yellow oil ( $500 \mathrm{mg}, 66 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.81(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.37(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.04(\mathrm{t}, \mathrm{J}=6.5$ $\mathrm{Hz}, 2 \mathrm{H}), 3.27(\mathrm{t}, \mathrm{J}=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.48(\mathrm{~s}, 3 \mathrm{H}), 1.68-1.61(\mathrm{~m}, 2 \mathrm{H}), 1.38-1.26(\mathrm{~m}, 10 \mathrm{H}) . \mathrm{MS}$ (ESI) $m / z: 326(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(2-azidoethoxy)ethyl 4-methylbenzenesulfonate (74b)


Compound 74b was obtained as a pale yellow oil ( 220 mg , $51 \%$ ) from derivative 72b ( 200 mg , $1.52 \mathrm{mmol})$ according to the procedure described for $\mathbf{7 4 a}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.81(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 4.19-4.16$ $(\mathrm{m}, 2 \mathrm{H}), 3.73-3.68(\mathrm{~m}, 2 \mathrm{H}), 3.61(\mathrm{t}, J=5.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.32(\mathrm{t}, J=5.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H})$. MS (ESI) $m / z: 286(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(2-(2-azidoethoxy)ethoxy)ethyl 4-methylbenzenesulfonate (74c)



Compound 74c was obtained as a pale yellow oil ( $125 \mathrm{mg}, 54 \%$ ) from derivative 72c ( 123 mg , 0.700 mmol ) according to the procedure described for 74a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.80(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.19-4.14$ $(\mathrm{m}, 3 \mathrm{H}), 3.73-3.69(\mathrm{~m}, 3 \mathrm{H}), 3.64(\mathrm{~m}, 3 \mathrm{H}), 3.36(\mathrm{~m}, 3 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 330$ $(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(2-(2-(2-azidoethoxy)ethoxy)ethoxy)ethyl 4-methylbenzenesulfonate (74d)


Compound 74d was obtained as a pale yellow oil ( 420 mg , $54 \%$ ) from derivative $\mathbf{7 2 d}$ ( 460 mg , 2.10 mmol ) according to the procedure described for 74a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.80(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.16(\mathrm{t}, J=$ $4.8,2 \mathrm{H}), 3.72-3.57(\mathrm{~m}, 12 \mathrm{H}), 3.38(\mathrm{t}, J=5.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 374$ $(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 14-azido-3,6,9,12-tetraoxatetradecyl 4-methylbenzenesulfonate (74e)


Compound 74e was obtained as a pale yellow oil ( $300 \mathrm{mg}, 49 \%$ ) from derivative $\mathbf{7 2 e}$ ( 391 mg , 1.48 mmol ) according to the procedure described for $\mathbf{7 4} \mathbf{a}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.80(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.19-4.14$ $(\mathrm{m}, 3 \mathrm{H}), 3.72-3.61(\mathrm{~m}, 14 \mathrm{H}), 3.38(\mathrm{~m}, 3 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 418(\mathrm{M}+\mathrm{H})^{+}$.


Compound $\mathbf{7 4 f}$ was obtained as a pale yellow oil ( $450 \mathrm{mg}, 60 \%$ ) from derivative $\mathbf{7 2 f}$ ( 500 mg , 1.63 mmol ) according to the procedure described for $74 \boldsymbol{a}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.79(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.19-4.12$ $(\mathrm{m}, 2 \mathrm{H}), 3.71-3.58(\mathrm{~m}, 18 \mathrm{H}), 3.51-3.34(\mathrm{~m}, 4 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 462(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-(2-(2-(4-nitrophenoxy)ethoxy)ethoxy)ethyl 4-methylbenzenesulfonate (75a)



To a solution of 73a ( $656 \mathrm{mg}, 2.30 \mathrm{mmol}$ ) in 25 mL of dry DCM, TEA ( $730 \mathrm{mg}, 2,76 \mathrm{mmol}$ ) and $\mathrm{TsCl}(526 \mathrm{mg}, 2.76 \mathrm{mmol})$ were added. The resulting solution was stirred for 18 h at room temperature. Then, DCM was added ( 20 mL ): the organic phase was washed with $\mathrm{HCl} 1 \mathrm{~N}(3 \mathrm{x}$ 15 mL ), saturated solution of $\mathrm{NaHCO}_{3}(3 \times 15 \mathrm{~mL})$, brine and dried over sodium sulfate. After filtration and evaporation, the crude was purified with a silica gel flash chromatography column (hex-EtOAc 80:20 to 70:30) to give a yellow oil ( $400 \mathrm{mg}, 41 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.19(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.79(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.33(\mathrm{~d}, J=$ $8.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.98(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.25-4.18(\mathrm{~m}, 2 \mathrm{H}), 4.19-4.12(\mathrm{~m}, 2 \mathrm{H}), 3.91-3.84(\mathrm{~m}$, 2H), $3.74-3.64(\mathrm{~m}, 4 \mathrm{H}), 3.67-3.59(\mathrm{~m}, 2 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 426(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 1-(4-nitrophenoxy)-14-tosyl-3,6,9,12-tetraoxatetradecane (75b)



Compound 75b was obtained as a pale yellow solid ( 550 mg , $50 \%$ ) from derivative 73b ( 865 $\mathrm{mg}, 2.18 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{7 5} \mathbf{a}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.18(\mathrm{~d}, \mathrm{~J}=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.79(\mathrm{~d}, \mathrm{~J}=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.33(\mathrm{~d}, \mathrm{~J}=$ $8.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.98(\mathrm{~d}, \mathrm{~J}=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.25-4.19(\mathrm{~m}, 2 \mathrm{H}), 4.17-4.11(\mathrm{~m}, 2 \mathrm{H}), 3.92-3.84(\mathrm{~m}$, 2H), $3.73-3.61(\mathrm{~m}, 14 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H})$. MS (ESI) $\mathrm{m} / \mathrm{z}: 514(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 3,5-dimethoxyaniline hydrochloride (78)



3,5-dimethoxyaniline $77(5 \mathrm{~g}, 32.64 \mathrm{mmol})$ was dissolved in $150 \mathrm{~mL}^{2} \mathrm{Et}_{2} \mathrm{O}$ and the reaction mixture was cooled at $0^{\circ} \mathrm{C} . \mathrm{HCl}$ (gas) was bubbled into the mixture for 1 h .

The corresponding hydrochloride salt (78) was obtained after filtration like white solid (6 g, $31.64 \mathrm{mmol}, 97 \%)$. Product was used for the next step without further purification.

## Synthesis of 4,6-dimethoxyindoline-2,3-dione (79)



The hydrochloride salt of 3,5-dimethoxyaniline $78(6 \mathrm{~g}, 31.64 \mathrm{mmol})$ was dissolved in 10.71 mL of oxalyl chloride at $0^{\circ} \mathrm{C}$ and the reaction mixture was heated at $170^{\circ} \mathrm{C}$ for 90 min . The solvent was removed under reduced pressure. The semisolid residue was dissolved in MeOH at $0{ }^{\circ} \mathrm{C}$ and then heated to reflux for 30 min . The reaction mixture was hot filtered and the precipitate was washed with MeOH to give a green solid 79 ( $4.13 \mathrm{~g}, 61 \%$ ). Product was used for the next step without further purification.

## Synthesis of 2-amino-4,6-dimethoxybenzoic acid (80)



Compound 79 ( $4.13 \mathrm{~g}, 19.94 \mathrm{mmol}$ ) was dissolved in NaOH ( $33 \%$ in water, 24.53 mL ) and the resulting solution was heated to $70{ }^{\circ} \mathrm{C} .6 .70 \mathrm{~mL}$ of $\mathrm{H}_{2} \mathrm{O}_{2}$ were added to solution dropwise and the mixture was stirred for 3 h increasing up the temperature to $100^{\circ} \mathrm{C}$. The reaction mixture was adjusted to pH 8 with HCl and then to pH 5 with AcOH . The precipitate formed was filtered, washed with water and dried to give the compound $\mathbf{8 0}$ ( $2.31 \mathrm{~g}, 59 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 6.71$ (brs, 2H), $5.93(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.78(\mathrm{~d}, J=2.4 \mathrm{~Hz}$, 1H), 3.76 (s, 3H), 3.70 (s, 3H). MS (ESI) m/z: 197 (M+H) ${ }^{+}$.

## Synthesis of 2-(2-azidoethyl)isoindoline-1,3-dione (81)



To a solution of compound $\mathbf{6 9}(300 \mathrm{mg}, 1.11 \mathrm{mmol})$ in 11 mL of DMF, sodium azide ( 362 mg , 5.57 mmol ) was added. The resulting solution was stirred at room temperature for 18 h . Then water ( 20 mL ) was added: the aqueous phase was extracted with EtOAc ( 3 x 15 mL ), the collected organic phases were washed with saturated solution of $\mathrm{NaHCO}_{3}(3 \times 15 \mathrm{~mL})$, brine and dried over sodium sulfate. After filtration and evaporation, the title compound was obtained as pale yellow solid ( $180 \mathrm{mg}, 80 \%$ )
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.88(\mathrm{dd}, J=5.4,2.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.74(\mathrm{dd}, J=5.4,2.9 \mathrm{~Hz}, 2 \mathrm{H})$, $3.90(\mathrm{~m}, 2 \mathrm{H}), 3.59(\mathrm{~m}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 217(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-(but-3-yn-1-yloxy)-3,5-dimethylbenzaldehyde (84a)



Compound 84a was obtained as a pale yellow oil ( 950 mg , $66 \%$ ) from derivative $\mathbf{8 3 a}$ ( 1.05 g , $7.13 \mathrm{mmol})$ and intermediate $76 \mathbf{a}(1.18 \mathrm{~g}, 7.84 \mathrm{mmol})$ according to the procedure described for 61.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.88(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{~s}, 2 \mathrm{H}), 3.95(\mathrm{t}, J=6.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.72(\mathrm{t}, J=$ 6.8, $2.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.08-2.06(\mathrm{~m}, 1 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 203(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-(hex-5-yn-1-yloxy)-3,5-dimethylbenzaldehyde (84b)



Compound 84b was obtained as a pale yellow solid ( 800 mg , 68\%) from derivative 83b (987 $\mathrm{mg}, 5.09 \mathrm{mmol})$ and intermediate $76(764 \mathrm{mg}, 5.09 \mathrm{mmol})$ according to the procedure described for 70.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.88(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{~s}, 2 \mathrm{H}), 3.85(\mathrm{t}, J=6.3 \mathrm{~Hz}, 3 \mathrm{H}), 2.34(\mathrm{~s}, 6 \mathrm{H})$, $2.32-2.29(\mathrm{~m}, 1 \mathrm{H}), 2.00-1.91(\mathrm{~m}, 3 \mathrm{H}), 1.83-1.75(\mathrm{~m}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 231(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-(but-3-yn-1-yloxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (85a)


Compound 85a was obtained as a pale yellow solid ( 270 mg , 72\%) from derivative 84a (200 $\mathrm{mg}, 0.990 \mathrm{mmol})$ and intermediate $\mathbf{6 4}(194 \mathrm{mg}, 0.990 \mathrm{mmol})$ according to the procedure described for 65.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta 11.83(\mathrm{~s}, 1 \mathrm{H}), 7.90(\mathrm{~s}, 2 \mathrm{H}), 6.74(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.51(\mathrm{~d}$, $J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.91-3.87(\mathrm{~m}, 5 \mathrm{H}), 3.84(\mathrm{~s}, 3 \mathrm{H}), 2.94-2.92(\mathrm{~m}, 1 \mathrm{H}), 2.70-2.62(\mathrm{~m}, 2 \mathrm{H})$, $2.32(\mathrm{~s}, 6 \mathrm{H})$. MS (ESI) $m / z: 379(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(4-(hex-5-yn-1-yloxy)-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (85b)


Compound $\mathbf{8 5 b}$ was obtained as a pale yellow solid ( 618 mg , 70\%) from derivative 84b ( 500 $\mathrm{mg}, 2.17 \mathrm{mmol}$ ) according to the procedure described for $\mathbf{6 5}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.14(\mathrm{~s}, 1 \mathrm{H}), 7.66(\mathrm{~s}, 2 \mathrm{H}), 6.83(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, J=$ $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.84(\mathrm{t}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H}), 2.34-2.29(\mathrm{~m}$, 1H), $2.02-1.93(\mathrm{~m}, 3 \mathrm{H}), 1.86-1.75(\mathrm{~m}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 407(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(2-(4-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy) ethyl)-1H-1,2,3-triazol-1-yl)ethyl)isoindoline-1,3-dione (86a)


A solution of compound $\mathbf{8 1}(60 \mathrm{mg}, 0.277 \mathrm{mmol})$ in 0.500 mL of dry DCM was added to a solution of $\mathrm{CuI}(1 \mathrm{mg}, 0.005 \mathrm{mmol})$, DIPEA ( $0.0057 \mathrm{~mL}, 0.010 \mathrm{mmol}$ ) and $\mathrm{AcOH}(0.0017$ $\mathrm{mL}, 0.010 \mathrm{mmol}$ ) in 0.300 mL of dry DCM. After $5 \mathrm{~min}, \mathbf{8 5 a}(100 \mathrm{mg}, 0.264 \mathrm{mmol})$ was added
and the resulting mixture was stirred at room temperature for 18 h . Then, the solvent was evaporated and the crude was purified by silica gel chromatography (DCM/EtOAc 70:30 to 20:80) to give the title compound 86a as white solid ( $144 \mathrm{mg}, 87 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 11.82(\mathrm{~s}, 1 \mathrm{H}), 8.06(\mathrm{~s}, 1 \mathrm{H}), 7.87(\mathrm{~s}, 2 \mathrm{H}), 7.83-7.81(\mathrm{~m}$, $4 \mathrm{H}), 6.73(\mathrm{~s}, 1 \mathrm{H}), 6.51(\mathrm{~s}, 1 \mathrm{H}), 4.63(\mathrm{t}, J=5.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.07-3.94(\mathrm{~m}, 4 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H})$, $3.84(\mathrm{~s}, 3 \mathrm{H}), 3.15-2.98(\mathrm{~m}, 2 \mathrm{H}), 2.19(\mathrm{~s}, 6 \mathrm{H})$. MS (ESI) m/z: $595(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis 2-(2-(4-(4-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy)butyl)

## -1H-1,2,3-triazol-1-yl)ethyl)isoindoline-1,3-dione (86b)



Compound 86b was obtained as a pale yellow solid (193 mg, 67\%) from derivative 85b (179 $\mathrm{mg}, 0.440 \mathrm{mmol})$ and intermediate $81(100 \mathrm{mg}, 0.462 \mathrm{mmol})$ according to the procedure described for 86a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.23(\mathrm{~s}, 1 \mathrm{H}), 7.82(\mathrm{dd}, J=5.8,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.74-7.69(\mathrm{~m}, 2 \mathrm{H})$, $7.66(\mathrm{~s}, 2 \mathrm{H}), 7.37(\mathrm{~s}, 1 \mathrm{H}), 6.82(\mathrm{~s}, 1 \mathrm{H}), 6.46(\mathrm{~s}, 1 \mathrm{H}), 4.67(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.15(\mathrm{t}, J=6.0$ $\mathrm{Hz}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.84-3.80(\mathrm{~m}, 2 \mathrm{H}), 2.80(\mathrm{t}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H})$, $1.94-1.82(\mathrm{~m}, 4 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 623(\mathrm{M}+\mathrm{H})^{+}$.

## [1,2,4]triazolo[3,4-c][1,2,4]triazepine-4-carboxamide (87)



To a solution of $\mathbf{4 9}(30 \mathrm{mg}, 0.074 \mathrm{mmol})$ in 2 mL of dry DMF, compound $90(17 \mathrm{mg}, 0.081$ mmol ) was added. Then, EDC ( $24 \mathrm{mg}, 0.126 \mathrm{mmol}$ ), HOBt ( $20 \mathrm{mg}, 0.126 \mathrm{mmol}$ ) and NMM ( $0.036 \mathrm{~mL}, 0.325 \mathrm{mmol}$ ) were added. The resulting mixture was stirred at room temperature for 18 h . Then, water was added ( 10 mL ): the aqueous organic phase was extracted with EtOAc (3 x 15 mL ). The collected organic phases were washed with saturated solution of $\mathrm{NaHCO}_{3}(3 \mathrm{x}$ 15 mL ), brine, dried over sodium sulfate, filtered and concentrated under reduced pressure. After purification on a silica gel flash-chromatography column (DCM-MeOH 97:3 to 90:1), the title compound was obtained as yellow solid ( $31 \mathrm{mg}, 75 \%$ )
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.64(\mathrm{dd}, J=8.6,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.39(\mathrm{~d}$, $J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.32(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.28(\mathrm{~s}, 1 \mathrm{H}), 6.80-6.68(\mathrm{~m}, 1 \mathrm{H}), 3.64-3.58(\mathrm{~m}$, $2 \mathrm{H}), 3.55-3.43(\mathrm{~m}, 8 \mathrm{H}), 3.34(\mathrm{t}, J=5.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.60(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 544(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of (ethane-1,2-diylbis(oxy))bis(ethane-2,1-diyl) bis(4-methylbenzenesulfonate) (88)


To a cooled solution of compound $70 \mathrm{c}(3 \mathrm{~g}, 19.98 \mathrm{mmol})$ in 40 mL of $\mathrm{DCM}, \mathrm{TsCl}(7.62 \mathrm{~g}$, $39.95 \mathrm{mmol})$ and KOH powder $(8.97 \mathrm{~g}, 159.84 \mathrm{mmol})$ were added in three portions at $0^{\circ} \mathrm{C}$. The resulting mixture was warm up at room temperature and stirred for 18 h . Then, water ( 80 mL ) was added: the aqueous phase was extracted with DCM ( $3 \times 40 \mathrm{~mL}$ ). The collected organic
phases were washed with $\mathrm{NaHCO}_{3}(3 \times 40 \mathrm{~mL})$, brine and dried over sodium sulfate. After filtration and evaporation under reduced pressure, the title compound was obtained as white solid ( $8.5 \mathrm{~g}, 93 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.79(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 4 \mathrm{H}), 7.34(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 4 \mathrm{H}), 4.18-4.11$ (m, 4H), $3.68-3.62(\mathrm{~m}, 4 \mathrm{H}), 3.57-3.52(\mathrm{~m}, 4 \mathrm{H}), 2.45(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 459(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 1,2-bis(2-azidoethoxy)ethane (89)



To a solution of compound $\mathbf{8 8}(3 \mathrm{~g}, 6.54 \mathrm{mmol})$ in 22 mL of DMF, sodium azide ( 850 mg , 13.08 mmol ) and tetrabutylammonium iodide ( $121 \mathrm{mg}, 0.327 \mathrm{mmol}$ ) were added. The resulting mixture was stirred at $80^{\circ} \mathrm{C}$ for 18 h . Then the mixture was concentrated under vacuo and the insoluble salts were filtered and the filtrated concentrated yielding the title compounds as colorless liquid ( $1.3 \mathrm{~g}, 99 \%$ )
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.71-3.62(\mathrm{~m}, 8 \mathrm{H}), 3.37(\mathrm{q}, \mathrm{J}=4.7 \mathrm{~Hz}, 4 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}$ : $201(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 2-(2-(2-azidoethoxy)ethoxy)ethan-1-amine hydrochloride (90)


To a solution of compound $\mathbf{8 9}(1.31 \mathrm{~g}, 6.54 \mathrm{mmol})$ in 40 mL of $\mathrm{EtOAc}, \mathrm{Ph}_{3} \mathrm{P}(1.71 \mathrm{~g}, 6.54$ mmol ) and 8 mL of HCl 1 M were added. The resulting mixture was stirred at room temperature for 18 h . Then the mixture was diluted with acidified water $(40 \mathrm{~mL})$ : the aqueous phase was washed with $\operatorname{EtOAc}(3 \times 20 \mathrm{~mL}$ ) and then concentrated under reduced pressure to give the title compound as yellow oil ( $1.08 \mathrm{~g}, 78 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.83$ (t, $J=5.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), $3.74-3.67$ (m, 6H), 3.46 (d, $J=5.1$ $\mathrm{Hz}, 2 \mathrm{H}), 3.25(\mathrm{~d}, \mathrm{~J}=5.2 \mathrm{~Hz}, 2 \mathrm{H})$. phenol (91)


Compound 91 was obtained as a yellow oil ( $26 \mathrm{mg}, 98 \%$ ) from derivative $93(30 \mathrm{mg}, 0.058$ mmol ) according to the procedure described for 51a.
${ }^{1}{ }^{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.16(\mathrm{~s}, 2 \mathrm{H}), 6.93(\mathrm{~s}, 1 \mathrm{H}), 6.44(\mathrm{~s}, 1 \mathrm{H}), 4.83(\mathrm{t}, \mathrm{J}=5.1 \mathrm{~Hz}, 2 \mathrm{H})$, $4.06-3.99(\mathrm{~m}, 2 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 3.86-3.81(\mathrm{~m}, 2 \mathrm{H}), 3.71-3.66(\mathrm{~m}, 2 \mathrm{H}), 3.55$ $-3.51(\mathrm{~m}, 2 \mathrm{H}), 2.90-2.82(\mathrm{~m}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 458(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(4-hydroxy-3,5-dimethylphenyl)-5,7-dimethoxyquinazolin-4(3H)-one (92)


Compound 92 was obtained as a yellow solid ( $573 \mathrm{mg}, 94 \%$ ) from derivative $64(370 \mathrm{mg}, 1.88$ mmol ) and intermediate $\mathbf{7 6 a}(282 \mathrm{mg}, 1.88 \mathrm{mmol})$ according to the procedure described for $\mathbf{6 5}$. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta 11.66(\mathrm{~s}, 1 \mathrm{H}), 8.93(\mathrm{~s}, 1 \mathrm{H}), 7.85(\mathrm{~s}, 2 \mathrm{H}), 6.70(\mathrm{~d}, J=1.7 \mathrm{~Hz}$, $1 \mathrm{H}), 6.48(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H}), 2.23(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 327$ $(\mathrm{M}+\mathrm{H})^{+}$. phenol (93)


Compound $\mathbf{9 3}$ was obtained as a yellow oil ( $80 \mathrm{mg}, 37 \%$ ) from derivative $\mathbf{9 2}$ ( $150 \mathrm{mg}, 0.455$ $\mathrm{mmol})$ and intermediate $\mathbf{7 4 c}(165 \mathrm{mg}, 0.500 \mathrm{mmol})$ according to the procedure described for 61.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.16(\mathrm{~s}, 2 \mathrm{H}), 6.92(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.43(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H})$, $4.94(\mathrm{~s}, 1 \mathrm{H}), 4.85-4.82(\mathrm{~m}, 2 \mathrm{H}), 4.06-4.01(\mathrm{~m}, 2 \mathrm{H}), 3.94(\mathrm{~s}, 3 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.89-3.85$ $(\mathrm{m}, 2 \mathrm{H}), 3.74-3.64(\mathrm{~m}, 4 \mathrm{H}), 3.39-3.34(\mathrm{~m}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 484(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of ethyl 2-((2-nitrophenyl)amino)-2-oxoacetate (95)



To a solution of 2-nitroaniline $\mathbf{9 4}(2.00 \mathrm{~g}, 14.48 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{ml})$, ethyl chlorooxoacetate $(1.78 \mathrm{ml}, 15.93 \mathrm{mmol})$ was added in 4 portions ant the resulting yellow suspension was stirred at room temperature for 18 h . Then the solvent was evaporated: the crude residue was taken up with EtOAc ( 100 mL ), washed with saturated $\mathrm{NaHCO}_{3}(3 \times 30 \mathrm{~mL})$, brine ( 30 mL ), dried over sodium sulfate and evaporated to dryness, giving the title compound as a yellow solid ( 3.38 g , 98\%).
${ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO- $d_{6}$ ) d $11.38(\mathrm{~s}, 1 \mathrm{H}), 8.17-8.05(\mathrm{~m}, 2 \mathrm{H}), 7.85-7.76(\mathrm{~m}, 1 \mathrm{H}), 7.52-$ $7.41(\mathrm{~m}, 1 \mathrm{H}), 4.34(\mathrm{q}, \mathrm{J}=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.33(\mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 239(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-amino-3-nitrobenzenesulfonyl chloride (96)



A solution of ethyl 2-(2-nitrophenylamino)-2-oxoacetate 95 ( $3 \mathrm{~g}, 12.55 \mathrm{mmol}$ ) in 6.7 ml of chlorosulfonic acid was heated at $80^{\circ} \mathrm{C}$ for 3 h . The red mixture was poured slowly into ice water ( 200 ml ) and stirred for 30 min . The product was extracted from the aqueous solution using $\mathrm{Et}_{2} \mathrm{O}(3 \times 40 \mathrm{~mL})$. The combined organic phases were washed with brine $(10 \mathrm{~mL})$, dried over sodium sulfate, filtered, and concentrated in vacuo to give the title compound as a brown solid which was immediately used for the next reaction without purification.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) d $8.16(\mathrm{~d}, \mathrm{~J}=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{dd}, \mathrm{J}=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.96$ (d, J = $8.8 \mathrm{~Hz}, 1 \mathrm{H}$ ).

## Synthesis of 4-amino-3-nitro-N-(pyridin-2-yl)benzenesulfonamide (98a)



To a stirred solution at $0^{\circ} \mathrm{C}$ of crude $96(1.41 \mathrm{~g}, 5.98 \mathrm{mmol})$ in dry pyridine ( 6 mL ) was added dropwise, under nitrogen atmosphere, 2-aminopyridine $\mathbf{9 7 a}$ ( $506 \mathrm{mg}, 5.38 \mathrm{mmol}$ ). The reaction was kept at $0^{\circ} \mathrm{C}$ for 3 hours until starting material disappearance. Then, water was added (10 mL ): the resulting solid was filtered and washed several time with water affording the title compound as orange solid ( $700 \mathrm{mg}, 40 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.47(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.04-8.00(\mathrm{~m}, 1 \mathrm{H}), 7.94(\mathrm{~s}, 2 \mathrm{H})$, $7.80-7.68$ (m, 2H), 7.09 (dd, $J=16.2,8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.87(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}:$ $295(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-amino-3-nitro-N-(pyridin-2-ylmethyl)benzenesulfonamide (98b)



To a stirred solution at $0{ }^{\circ} \mathrm{C}$ of crude 96 ( $765 \mathrm{mg}, 3.24 \mathrm{mmol}$ ) in dry THF ( 30 mL ), 2picolylamine 97b ( $1.33 \mathrm{~mL}, 12.93 \mathrm{mmol}$ ) was added dropwise, under nitrogen atmosphere, and the reaction was stirred for 18 h at room temperature. Then, the reaction mixture was concentrated under reduced pressure and the residue was taken up with 50 ml of water. The aqueous phase was extracted with EtOAc ( $3 \times 20 \mathrm{~mL}$ ), the combined organic phases were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated under reduced pressure. The title compound was obtained as yellow solid after crystallization with EtOH ( $590 \mathrm{mg}, 60 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.40-8.38(\mathrm{~m}, 1 \mathrm{H}), 8.26(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.95(\mathrm{~s}, 2 \mathrm{H})$, $7.72-7.61(\mathrm{~m}, 2 \mathrm{H}), 7.33(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.23-7.18(\mathrm{~m}, 1 \mathrm{H}), 7.06(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H})$, $4.09(\mathrm{~s}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 309(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-amino-3-nitro-N-(pyridin-3-ylmethyl)benzenesulfonamide (98c)



Compound 98c was obtained as a yellow solid ( $560 \mathrm{mg}, 62 \%$ ) from derivative 96 ( $700 \mathrm{mg}, 2.96$ $\mathrm{mmol})$ and the 3-picolylamine $\mathbf{9 7 c}(1.20 \mathrm{~mL}, 11.83 \mathrm{mmol})$ according to the procedure described for 98b.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.39(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.28(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.16(\mathrm{~s}$, $1 \mathrm{H}), 7.97(\mathrm{~s}, 1 \mathrm{H}), 7.66-7.61(\mathrm{~m}, 1 \mathrm{H}), 7.30-7.24(\mathrm{~m}, 1 \mathrm{H}), 7.07(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.03(\mathrm{~s}$, 2H). MS (ESI) $m / z: 309(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 3-amino-N-(tert-butyl)-4-nitrobenzenesulfonamide (98d)


Compound 98d was obtained as orange solid ( $250 \mathrm{mg}, 73 \%$ ) from derivative $96(420 \mathrm{mg}, 1.77$ mmol ) and the tert-butylamine $97 \mathrm{~d}(0.746 \mathrm{~mL}, 7.09 \mathrm{mmol})$ according to the procedure described for 98b.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) d $8.38(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.95(\mathrm{~s}, 2 \mathrm{H}), 7.70(\mathrm{dd}, \mathrm{J}=9.0,2.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.44(\mathrm{~s}, 1 \mathrm{H}), 7.11(\mathrm{~d}, \mathrm{~J}=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.10(\mathrm{~s}, 9 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 274(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-amino-N-methyl-3-nitrobenzenesulfonamide (98e)



Compound 98e was obtained as yellow solid ( $260 \mathrm{mg}, 70 \%$ ) from derivative 96 ( $380 \mathrm{mg}, 1.61$ $\mathrm{mmol})$ and 2 M THF solution of methylamine $97 \mathrm{e}(3.22 \mathrm{~mL}, 6.44 \mathrm{mmol})$ according to the procedure described for $\mathbf{9 8 b}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.36-8.29(\mathrm{~m}, 1 \mathrm{H}), 7.99(\mathrm{~s}, 2 \mathrm{H}), 7.66(\mathrm{dd}, J=9.0,2.2 \mathrm{~Hz}$, 1H), $7.38-7.34(\mathrm{~m}, 1 \mathrm{H}), 7.14(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.39(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 232$ $(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of 4-amino-N-cyclopropyl-3-nitrobenzenesulfonamide (98f)



Compound $98 f$ was obtained as yellow solid ( $450 \mathrm{mg}, 85 \%$ ) from derivative 96 ( $488 \mathrm{mg}, 2.06$ $\mathrm{mmol})$ and cyclopropylamine $\mathbf{9 7 f}(0.571 \mathrm{~mL}, 8.25 \mathrm{mmol})$ according to the procedure described for 98b.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta 8.37(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.00(\mathrm{~s}, 2 \mathrm{H}), 7.80(\mathrm{~d}, \mathrm{~J}=2.9 \mathrm{~Hz}$, $1 \mathrm{H}), 7.68(\mathrm{dd}, \mathrm{J}=9.0,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{~d}, \mathrm{~J}=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.16-2.08(\mathrm{~m}, 1 \mathrm{H}), 0.53-0.46$ (m, 2H), $0.39-0.34(\mathrm{~m}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 258(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-amino-N-cyclopentyl-3-nitrobenzenesulfonamide (98g)



Compound 98g was obtained as orange solid ( $387 \mathrm{mg}, 70 \%$ ) from derivative $96(468 \mathrm{mg}, 1.97$ $\mathrm{mmol})$ and cyclopentylamine $\mathbf{9 7 g}(0.780 \mathrm{~mL}, 7.9 \mathrm{mmol})$ according to the procedure described for 98b.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.36(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.97(\mathrm{~s}, 2 \mathrm{H}), 7.68(\mathrm{dd}, J=9.0,2.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.55(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.13(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.38-3.35(\mathrm{~m}, 1 \mathrm{H}), 1.66-1.48$ (m, 4H), $1.42-1.25(\mathrm{~m}, 4 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 286(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 4-amino-N-cyclohexyl-3-nitrobenzenesulfonamide (98h)



Compound $\mathbf{9 8 h}$ was obtained as yellow solid ( $950 \mathrm{mg}, 63 \%$ ) from derivative $96(1.2 \mathrm{~g}, 5.07$ $\mathrm{mmol})$ and cyclohexylamine $\mathbf{9 7 h}(2.32 \mathrm{~mL}, 20.28 \mathrm{mmol})$ according to the procedure described for 98b.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.36(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.96(\mathrm{~s}, 2 \mathrm{H}), 7.70(\mathrm{dd}, J=9.0,2.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.56(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.12(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.94-2.85(\mathrm{~m}, 1 \mathrm{H}), 1.62-1.52$ $(\mathrm{m}, 5 \mathrm{H}), 1.19-1.09(\mathrm{~m}, 5 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 300(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 3,4-diamino-N-(pyridin-2-yl)benzenesulfonamide (99a)



Compound 98a ( $440 \mathrm{mg}, 1.5 \mathrm{mmol}$ ) was dissolved in 7.20 mL of AcOH and Zn dust ( 977 mg , 1.5 mmol ) was added portionwise. The reaction mixture was stirred for 1.5 h at room temperature until starting material disappearance. Then, the insoluble salts are filtered and the filtrate, concentrated under reduced pressure. The residual semisolid was taken up with water: the aqueous phase ( 20 mL ) was extracted with EtOAc ( $3 \times 15 \mathrm{~mL}$ ) and the collected organic phases were washed with saturated solution of $\mathrm{NaHCO}_{3}(3 \times 15 \mathrm{~mL})$ brine, dried over sodium sulfate, filtered and evaporated to dryness. The title compound was obtained as light brown solid ( $257 \mathrm{mg}, 64 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 10.86(\mathrm{~s}, 1 \mathrm{H}), 8.09(\mathrm{~d}, J=5.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.66-7.60(\mathrm{~m}, 1 \mathrm{H})$, $7.08(\mathrm{~s}, 1 \mathrm{H}), 7.00(\mathrm{~s}, 1 \mathrm{H}), 6.97-6.87(\mathrm{~m}, 2 \mathrm{H}), 6.48(\mathrm{dd}, J=8.3,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.23(\mathrm{~s}, 2 \mathrm{H}), 4.82$ (s, 2H). MS (ESI) m/z: $265(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 3,4-diamino-N-(pyridin-2-ylmethyl)benzenesulfonamide (99b)


To a solution of $\mathbf{9 8 b}(300 \mathrm{mg}, 0.973 \mathrm{mmol})$ in 10 mL of $\mathrm{EtOAc}, 10 \% \mathrm{Pd} / \mathrm{C}$ was added. The mixture was stirred under a hydrogen atmosphere (balloon) for 18 h . Then, the solvent was evaporated and the title compound was obtained as light brown solid ( $250 \mathrm{mg}, 92 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.44(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.75(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.60(\mathrm{t}, J$ $=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{dd}, J=7.6,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.96(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, 6.86 (dd, $J=8.1,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.22$ (brs, 2H), 4.84 (brs, 2H), 3.96 (d, $J=6.2 \mathrm{~Hz}, 2 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 279(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 3,4-diamino-N-(pyridin-3-ylmethyl)benzenesulfonamide (99c)



Compound 99c was obtained as a yellow solid ( $85 \mathrm{mg}, 94 \%$ ) from derivative $\mathbf{9 8 c}$ ( 100 mg , 0.324 mmol ) according to the procedure described for $\mathbf{9 9 b}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 8.42(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.69-7.57(\mathrm{~m}, 2 \mathrm{H}), 7.31(\mathrm{dd}, J=$ $7.8,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{dd}, J=8.1,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $1 \mathrm{H}), 5.23$ (brs, 2H), 4.84 (brs, 2H), $3.90(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H})$. MS (ESI) $m / z: 279(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 3-amino-N-(tert-butyl)-4-nitrobenzenesulfonamide (99d)



Compound 99d was obtained as orange solid ( $240 \mathrm{mg}, 89 \%$ ) from derivative $98 \mathbf{~ ( ~} 283 \mathrm{mg}, 1.03$ $\mathrm{mmol})$ according to the procedure described for 99b.
${ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO- $\left.d_{6}\right) \delta 6.96(\mathrm{~d}, \mathrm{~J}=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.90-6.84(\mathrm{~m}, 2 \mathrm{H}), 6.53(\mathrm{~d}, \mathrm{~J}=8.2$ Hz, 1H), 5.13 (brs, 2H), 4.79 (brs, 2H), 1.08 (s, 9H). MS (ESI) m/z: 244 (M+H) ${ }^{+}$

## Synthesis of 4-amino-N-methyl-3-nitrobenzenesulfonamide (99e)



Compound 99e was obtained as yellow solid ( $80 \mathrm{mg}, 92 \%$ ) from derivative $\mathbf{9 8 e}$ ( $100 \mathrm{mg}, 0.432$ mmol ) according to the procedure described for $\mathbf{9 9 b}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 6.90(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.83-6.78(\mathrm{~m}, 2 \mathrm{H}), 6.55(\mathrm{~d}, J=$ $8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.19(\mathrm{~s}, 2 \mathrm{H}), 4.83(\mathrm{~s}, 2 \mathrm{H}), 2.32(\mathrm{~d}, \mathrm{~J}=5.2 \mathrm{~Hz}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 202(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 3,4-diamino-N-cyclopropylbenzenesulfonamide (99f)



Compound $\mathbf{9 9 f}$ was obtained as brown oil ( 340 mg , $97 \%$ ) from derivative $\mathbf{9 8 f}(400 \mathrm{mg}, 1.55$ mmol ) according to the procedure described for $99 \mathbf{b}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 7.32(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.85$ (dd, $J=8.2,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.20(\mathrm{~s}, 2 \mathrm{H}), 4.82(\mathrm{~s}, 2 \mathrm{H}), 2.04-1.99(\mathrm{~m}, 1 \mathrm{H})$, $0.46-0.32(\mathrm{~m}, 4 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 228(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 3,4-diamino-N-cyclopentylbenzenesulfonamide (99g)



Compound $\mathbf{9 9} \mathrm{g}$ was obtained as orange solid $(0.320 \mathrm{~g}, 92 \%)$ from derivative $\mathbf{9 8 g}(0.387 \mathrm{~g}, 1.36$ mmol ) according to the procedure described for $99 \mathbf{b}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 6.99(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.83(\mathrm{dd}$, $J=8.1,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.53(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.16(\mathrm{brs}, 2 \mathrm{H}), 4.83(\mathrm{brs}, 2 \mathrm{H}), 3.29-3.21(\mathrm{~m}$, $1 \mathrm{H}), 1.63-1.46(\mathrm{~m}, 4 \mathrm{H}), 1.40-1.24(\mathrm{~m}, 4 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 256(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 3,4-diamino-N-cyclohexylbenzenesulfonamide (99h)



Compound 99h was obtained as light brown solid ( $420 \mathrm{mg}, 94 \%$ ) from derivative 98h ( 496 mg , 1.66 mmol ) according to the procedure described for $\mathbf{9 9 b}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 6.98(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.83(\mathrm{dd}$, $J=8.1,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.52(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.14($ brs, 2 H$), 4.80($ brs, $2 \mathrm{H}), 2.79(\mathrm{~s}, 1 \mathrm{H}), 1.62$ $-1.39(\mathrm{~m}, 5 \mathrm{H}), 1.14-0.94(\mathrm{~m}, 5 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 270(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of N-(tert-butyl)-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-sulfonamide

 (100)

Compound 99d ( $300 \mathrm{mg}, 1.24 \mathrm{mmol}$ ) was solubilized in 12 mL of dry DMF and 4-hydroxy-3,5-dimethylbenzaldehyde $76 \mathbf{~}(186 \mathrm{mg}, 1.24 \mathrm{mmol})$ and $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{5}(335 \mathrm{mg}, 1.61 \mathrm{mmol})$ were added. The resulting mixture was heated at $80^{\circ} \mathrm{C}$ and stirred at this temperature for 18 h . After cooling at room temperature, water was added. The brown precipitate formed was recovered by filtration and was washed several times with water. After recrystallization from EtOH, compound $\mathbf{1 0 0}$ was obtained as light yellow solid ( $295 \mathrm{mg}, 64 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta 7.96(\mathrm{~s}, 1 \mathrm{H}), 7.79(\mathrm{~s}, 2 \mathrm{H}), 7.63(\mathrm{~s}, 2 \mathrm{H}), 7.42(\mathrm{~s}, 1 \mathrm{H}), 2.27(\mathrm{~s}$, $6 \mathrm{H}), 1.08(\mathrm{~s}, 9 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}: 374(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 2-bromo-4-hydroxy-3,5-dimethylbenzaldehyde (101)



To a solution of $76 \mathbf{a}(1 \mathrm{~g}, 6.66 \mathrm{mmol})$ in 3.33 mL of $\mathrm{H}_{2} \mathrm{SO}_{4}$, N -bromosuccinimide ( $3.55 \mathrm{~g}, 19.98$ mmol ) was added portionwise with continuous stirring. The resulting mixture was heated at 60 ${ }^{\circ} \mathrm{C}$ and stirred for 6 hours. Then water was added and the resulting precipitate was filtered and washed several time with water affording the title compound as red solid ( $290 \mathrm{mg}, 20 \%$ ).
${ }^{1}{ }^{1} \mathrm{~N}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 10.29(\mathrm{~s}, 1 \mathrm{H}), 7.64(\mathrm{~s}, 1 \mathrm{H}), 2.41(\mathrm{~s}, 3 \mathrm{H}), 2.26(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}$ (ESI) $m / z: 228(\mathrm{M}+\mathrm{H})^{+}, 230(\mathrm{M}+\mathrm{H}+2)^{+}$

## Synthesis of 5-hydroxy-4,6-dimethyl-[1,1'-biphenyl]-2-carbaldehyde (103a)



In a 20 mL CEM pressure vessel equipped with a stirrer bar, compound $101(100 \mathrm{mg}, 0.436$ mmol ) was dissolved in 8.8 mL of EtOH. Then, phenylboronic acid $\mathbf{1 0 2 a}(64 \mathrm{mg}, 0.524 \mathrm{mmol})$, Tetrakis(triphenylphosphine)palladium ( $23 \mathrm{mg}, 0.022 \mathrm{mmol}$ ) and $\mathrm{K}_{2} \mathrm{CO}_{3}(157 \mathrm{mg}, 1.13 \mathrm{mmol}$ ) were added. The microwave vial was sealed and heated in a CEM Discover microwave synthesizer to $80^{\circ} \mathrm{C}$ for 30 min . After cooling to room temperature, the reaction 0mixture was filtered and the filtrate concentrated under vacuo. The residual semisolid was taken up with EtOAc ( 30 mL ) and the organic phase was washed with water ( $3 \times 15 \mathrm{~mL}$ ), brine, dried over
sodium sulfate and filtered. The solvent was removed under reduced pressure and the resulting crude material was purified by silica gel chromatography (hex/EtOAc $95: 5$ to $50: 50$ ) to give compound 103a a light brown solid ( $70 \mathrm{mg}, 70 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.51(\mathrm{~s}, 1 \mathrm{H}), 7.73(\mathrm{~s}, 1 \mathrm{H}), 7.48-7.38(\mathrm{~m}, 3 \mathrm{H}), 7.23-7.18(\mathrm{~m}$, 2H), $2.35(\mathrm{~s}, 3 \mathrm{H}), 2.00(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) m / z: 227(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 5-hydroxy-3'-methoxy-4,6-dimethyl-[1,1'-biphenyl]-2-carbaldehyde (103b)



Compound 103b was obtained as orange solid ( 75 mg , $67 \%$ ) from derivative 101 ( 100 mg , $0.436 \mathrm{mmol})$ and $\mathbf{1 0 2 b}(80 \mathrm{mg}, 0.524 \mathrm{mmol})$ according to the procedure described for 103a.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ) $\delta 9.36(\mathrm{~s}, 1 \mathrm{H}), 7.54(\mathrm{~s}, 1 \mathrm{H}), 7.39(\mathrm{t}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.08-$ $6.95(\mathrm{~m}, 1 \mathrm{H}), 6.81-6.73(\mathrm{~m}, 1 \mathrm{H}), 6.38-6.27(\mathrm{~m}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 3.68(\mathrm{~s}, 2 \mathrm{H}), 2.27(\mathrm{~s}, 2 \mathrm{H})$. MS (ESI) $m / z: 257(\mathrm{M}+\mathrm{H})^{+}$.

Synthesis of 3'-(dimethylamino)-5-hydroxy-4,6-dimethyl-[1,1'-biphenyl]-2-carbaldehyde (103c)


Compound 103c was obtained as orange solid ( $55 \mathrm{mg}, 67 \%$ ) from derivative $101(70 \mathrm{mg}, 0.305$ $\mathrm{mmol})$ and $\mathbf{1 0 2 c}(61 \mathrm{mg}, 0.367 \mathrm{mmol})$ according to the procedure described for $\mathbf{1 0 3 a}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.56(\mathrm{~s}, 1 \mathrm{H}), 7.72(\mathrm{~s}, 1 \mathrm{H}), 7.31-7.28(\mathrm{~m}, 1 \mathrm{H}), 6.76(\mathrm{dd}, J=8.6$, $2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.58-6.53(\mathrm{~m}, 1 \mathrm{H}), 6.51(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.96(\mathrm{~s}, 6 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}), 2.04(\mathrm{~s}$, 3H). MS (ESI) m/z: $270(\mathrm{M}+\mathrm{H})^{+}$.

## Synthesis of 15-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole)-6-sulfonamido)

 pyridin-3-yl)-3,6,9,12-tetraoxapentadec-14-ynoic acid (106)

Compound $\mathbf{1 1 9}$ ( $9 \mathrm{mg}, 0.013 \mathrm{mmol}$ ) was dissolved in 0.200 mL of a solution DCM/TFA (1:1) and the mixture was stirred for 30 min and monitored by LC-MS. Then, solvent was evaporated, and the resulting solid was used for the next step without further purification ( $8 \mathrm{mg}, 96 \%$ ). LC/MS 1.2-1.25 min, $m / z: 639.2(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-((6-((3-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole)-6-sulfonamido) pyridin-3-yl)prop-2-yn-1-yl)oxy)hexyl)oxy)acetic acid (107)


Compound $\mathbf{1 0 7}$ was obtained as brown solid ( $8 \mathrm{mg}, 96 \%$ ) from compound $\mathbf{1 2 0}(7 \mathrm{mg}, 0.011$ $\mathrm{mmol})$ according to the procedure described for $\mathbf{1 0 6}$.

LC/MS 1.3-1.4 min, $m / z: 607.2(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-((4-(()-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole)-6-sulfonamido) pyridin-3-yl)prop-2-yn-1-yl)oxy)methyl)benzyl)oxy)acetic acid (108)


Compound $\mathbf{1 0 8}$ was obtained as brown solid ( $6 \mathrm{mg}, 96 \%$ ) from compound $\mathbf{1 2 1}(7 \mathrm{mg}, 0.010$ mmol ) according to the procedure described for $\mathbf{1 0 6}$.

LC/MS 1.35-1.45 min, $m / z: 627.2(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(4-(4-((3-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole)-6-sulfon amido)pyridin-3-yl)prop-2-yn-1-yl)oxy)butoxy)butoxy)acetic acid (109)


Compound $\mathbf{1 0 9}$ was obtained as brown solid ( $13 \mathrm{mg}, 95 \%$ ) from compound $\mathbf{1 2 2}(16 \mathrm{mg}, 0.021$ $\mathrm{mmol})$ according to the procedure described for 106.

LC/MS 1.3-1.4 min, $m / z: 651.2(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(2-(2-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy) ethoxy)ethoxy)ethoxy)acetic acid (110)


Compound $\mathbf{1 1 0}$ was obtained as yellow solid ( 47 mg , $98 \%$ ) from compound $\mathbf{1 4 1}(53 \mathrm{mg}, 0.093$ $\mathrm{mmol})$ according to the procedure described for 106.

LC/MS 1.3-1.4, m/z: $517.1(\mathrm{M}+\mathrm{H})^{+}$ hexyl)oxy)acetic acid (111)


Compound $\mathbf{1 1 1}$ was obtained as yellow solid ( $59 \mathrm{mg}, 99 \%$ ) from compound $142(60 \mathrm{mg}, 0.111$ $\mathrm{mmol})$ according to the procedure described for $\mathbf{1 0 6}$.

LC/MS 1.5-1.6 min, $m / z: 485.1(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-((4-((4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy) methyl)benzyl)oxy)acetic acid (112)


Compound $\mathbf{1 1 2}$ was obtained as yellow solid ( $45 \mathrm{mg}, 99 \%$ ) from compound $\mathbf{1 4 3}$ ( $50 \mathrm{mg}, 0.09$ $\mathrm{mmol})$ according to the procedure described for $\mathbf{1 0 6}$.

LC/MS 1.7-1.8 min, $m / z: 505.2(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of 2-(4-(4-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethylphenoxy)
butoxy)butoxy)acetic acid (113)


Compound $\mathbf{1 1 3}$ was obtained as brown solid ( $43 \mathrm{mg}, 97 \%$ ) from compound $\mathbf{1 4 4}(49 \mathrm{mg}, 0.084$ mmol ) according to the procedure described for $\mathbf{1 0 6}$.

LC/MS 1.5-1.6 min, $m / z: 529.2(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of N-(5-bromopyridin-2-yl)-2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole-6-

 sulfonamide (114)

Compound $\mathbf{1 1 4}$ was obtained as brown solid ( $380 \mathrm{mg}, 56 \%$ ) from derivative $124(560 \mathrm{mg}, 1.5$ $\mathrm{mmol})$ and the intermediate $\mathbf{7 6 a}(225 \mathrm{mg}, 1.5 \mathrm{mmol})$ according to the procedure described for 100.
${ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO-d6) $\delta 11.16(\mathrm{~s}, 1 \mathrm{H}), 8.86-8.84(\mathrm{~m}, 1 \mathrm{H}), 8.26(\mathrm{~d}, \mathrm{~J}=2.5 \mathrm{~Hz}, 1 \mathrm{H})$, $8.10(\mathrm{~s}, 1 \mathrm{H}), 7.88(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.77(\mathrm{~s}, 2 \mathrm{H}), 7.73-7.55(\mathrm{~m}, 1 \mathrm{H}), 7.10(\mathrm{dd}, \mathrm{J}=8.6,5.1$ $\mathrm{Hz}, 1 \mathrm{H}), 2.26(\mathrm{~s}, 6 \mathrm{H})$. LC/MS: $1.3 \mathrm{~min} \mathrm{~m} / \mathrm{z} 473(\mathrm{M}+\mathrm{H})^{+}, 475(\mathrm{M}+\mathrm{H}+2)^{+}$

## Synthesis of tert-butyl 3,6,9,12-tetraoxapentadec-14-ynoate (115)



To a cooled solution of $\mathbf{1 2 5}(700 \mathrm{mg}, 3.72 \mathrm{mmol})$ in 8 mL of dry THF, $\mathrm{NaH} 60 \%$ in mixture $(178 \mathrm{mg}, 4.46 \mathrm{mmol})$ was added and the resulting mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min . Then, tert-butyl 2-bromoacetate $(0.660 \mathrm{~mL}, 4.46 \mathrm{mmol})$ was added and the final suspension was continuously stirred for 14 h at room temperature. Then, the solvent was removed under reduced pressure and the residual was taken up with saturated solution of $\mathrm{NH}_{4} \mathrm{Cl}(30 \mathrm{~mL})$ : the aqueous phase was extracted with EtOAc ( $3 \times 20 \mathrm{~mL}$ ). The collected organic phases were washed with brine, dried over sodium sulfate, filtered and concentrated. The product $\mathbf{1 1 5}$ was obtained after
purification filtration on a silica gel flash-chromatography column (heptane-EtOAc 70:30 to $0: 100$ ) to yield a colorless oil ( $570 \mathrm{mg}, 50 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.20(\mathrm{~s}, 2 \mathrm{H}), 4.02(\mathrm{~s}, 2 \mathrm{H}), 3.75-3.64(\mathrm{~m}, 12 \mathrm{H}), 2.42(\mathrm{~s}, 1 \mathrm{H})$, $1.48(\mathrm{~s}, 9 \mathrm{H})$.

## Synthesis of tert-butyl 2-((6-(prop-2-yn-1-yloxy)hexyl)oxy)acetate (116)



Compound 127 ( $100 \mathrm{mg}, 0.640 \mathrm{mmol}$ ) and rhodium acetate ( $15 \mathrm{mg}, 0.032 \mathrm{mmol}$ ) were dissolved in 0.810 mL of dry DCM under nitrogen atmosphere and the resulting green suspension was cooled at $0^{\circ} \mathrm{C}$. Then, tert-Butyl diazoacetate $(0.100 \mathrm{~mL}, 0.768 \mathrm{mmol})$ was added dropwise under 1 hour. Then, the reaction was allowed to reach room temperature and was stirred for 18 h . The crude was purified on a silica gel flash-chromatography column (heptane- EtOAc 90:10 to 40:60) to yield the title compound as colorless oil ( $115 \mathrm{mg}, 66 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.03(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.85(\mathrm{~s}, 2 \mathrm{H}), 3.42(\mathrm{t}, J=6.6 \mathrm{~Hz}, 4 \mathrm{H})$, $2.35(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.59-1.44(\mathrm{~m}, 4 \mathrm{H}), 1.39(\mathrm{~s}, 9 \mathrm{H}), 1.31(\mathrm{q}, J=3.9 \mathrm{~Hz}, 4 \mathrm{H})$.

## Synthesis of tert-butyl 2-((4-((prop-2-yn-1-yloxy)methyl)benzyl)oxy)acetate (117)



A solution of propargyl alcohol $(0.040 \mathrm{~mL}, 0.698 \mathrm{mmol})$ in 10 mL of dry DMF was cooled at $0^{\circ} \mathrm{C}$ and $\mathrm{NaH} 60 \%$ in mixture ( $28 \mathrm{mg}, 0.698 \mathrm{mmol}$ ) was added. The resulting mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min under nitrogen atmosphere. Then, compound $\mathbf{1 3 0}$ ( $200 \mathrm{mg}, 0.634$
mmol ) was added and the final suspension was continuously stirred for 14 h at room temperature. The crude was concentrated under reduced pressure and the residual was taken up with saturated solution of $\mathrm{NH}_{4} \mathrm{Cl}(30 \mathrm{~mL})$ : the aqueous phase was extracted with EtOAc (3 x 20 mL ). The collected organic phases were washed with brine, dried over sodium sulfate, filtered and concentrated. The product $\mathbf{1 1 7}$ was obtained after purification filtration on a silica gel flash-chromatography column (heptane-EtOAc 90:10 to 50:50) to yield a colorless oil (110 $\mathrm{mg}, 60 \%)$.
${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40-7.32(\mathrm{~m}, 4 \mathrm{H}), 4.61(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 2 \mathrm{H}), 4.17(\mathrm{~d}, J=2.4$ $\mathrm{Hz}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 2 \mathrm{H}), 2.46(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H})$.

Synthesis of tert-butyl 2-(4-(4-(prop-2-yn-1-yloxy)butoxy)butoxy)acetate (118)


Compound 118 was obtained as colorless oil ( $15 \mathrm{mg}, 30 \%$ ) from compound $\mathbf{1 3 6}(62 \mathrm{mg}, 0.160$ $\mathrm{mmol})$ and propargyl alcohol $(0.011 \mathrm{~mL}, 0.176 \mathrm{mmol})$ according to the procedure described for 117.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.12(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.93(\mathrm{~s}, 2 \mathrm{H}), 3.52(\mathrm{td}, J=6.3,1.6 \mathrm{~Hz}$, $4 \mathrm{H}), 3.46-3.38(\mathrm{~m}, 4 \mathrm{H}), 2.40(\mathrm{t}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.70-1.61(\mathrm{~m}, 8 \mathrm{H}), 1.47(\mathrm{~s}, 9 \mathrm{H})$.

## Synthesis of tert-butyl 15-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole)-6-sulfon

amido) pyridin-3-yl)-3,6,9,12-tetraoxapentadec-14-ynoate (119)


A solution of compound $\mathbf{1 1 4}(50 \mathrm{mg}, 0.106 \mathrm{mmol})$ in 1 mL of dry DMF was added to $\mathrm{K}_{2} \mathrm{CO}_{3}$ $(44 \mathrm{mg}, 0.318 \mathrm{mmol})$ and $\mathrm{CuI}(6 \mathrm{mg}, 0.027 \mathrm{mmol})$ under argon atmosphere. Then, compound

115 ( $48 \mathrm{mg}, 0.158 \mathrm{mmol}$ ), tri-tert-butylphosphine ( $0.011 \mathrm{~mL}, 0.042 \mathrm{mmol}$ ) bis(tri-tertbutylphosphine)palladium( 0 ) ( $40 \mathrm{mg}, 0.078 \mathrm{mmol}$ ) were added and the resulting black suspension was stirred at $100{ }^{\circ} \mathrm{C}$ for 18 h . Then, the solvent was removed under reduced pressure and the solid residual was taken up with a mixture of $\mathrm{DCM} / \mathrm{MeOH}$ (9:1) and filtered on a path of celite. The filtrate was evaporated under reduced pressure and the crude was purified on silica gel flash-chromatography column (DCM-MeOH 97:3 to 80:20) to give the title compound as brown solid ( $9 \mathrm{mg}, 13 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 8.44(\mathrm{~s}, 1 \mathrm{H}), 8.18(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.81(\mathrm{dd}, J=8.6,1.8 \mathrm{~Hz}$, $1 \mathrm{H}), 7.72(\mathrm{~s}, 2 \mathrm{H}), 7.67(\mathrm{dd}, J=8.6,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.21-7.16(\mathrm{~m}, 1 \mathrm{H})$, 4.37 (s, 2H), $4.00(\mathrm{~s}, 2 \mathrm{H}), 3.77-3.55(\mathrm{~m}, 12 \mathrm{H}), 2.31(\mathrm{~s}, 6 \mathrm{H}), 1.43(\mathrm{~s}, 9 \mathrm{H})$.

LC/MS: $1.47 \mathrm{~min} m / z 695.3(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of tert-butyl 2-((6-((3-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole)-6-

 sulfonamido)pyridin-3-yl)prop-2-yn-1-yl)oxy)hexyl)oxy)acetate (120)

Compound $\mathbf{1 2 0}$ was obtained as brown solid ( $8 \mathrm{mg}, 11 \%$ ) from compound $\mathbf{1 1 4}(50 \mathrm{mg}, 0.106$ $\mathrm{mmol})$ and intermediate $\mathbf{1 1 6}(43 \mathrm{mg}, 0.159 \mathrm{mmol})$ according to the procedure described for 119.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 8.21-8.14(\mathrm{~m}, 3 \mathrm{H}), 7.81(\mathrm{dd}, J=8.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.71(\mathrm{~s}$, $2 \mathrm{H}), 7.67(\mathrm{dd}, J=8.7,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.19(\mathrm{dd}, J=8.7,0.9 \mathrm{~Hz}, 1 \mathrm{H})$, $4.30(\mathrm{~s}, 2 \mathrm{H}), 3.92(\mathrm{~s}, 2 \mathrm{H}), 3.53(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.46(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.31(\mathrm{~s}, 6 \mathrm{H}), 1.57-$ $1.54(\mathrm{~m}, 4 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H}), 1.38-1.35(\mathrm{~m}, 4 \mathrm{H})$.

LC/MS: $1.64 \mathrm{~min} m / z 663.3(\mathrm{M}+\mathrm{H})^{+}$ sulfonamido)pyridin-3-yl)prop-2-yn-1-yl)oxy)methyl)benzyl)oxy)acetate (121)


Compound $\mathbf{1 2 1}$ was obtained as brown solid ( $10 \mathrm{mg}, 14 \%$ ) from compound 114 ( $50 \mathrm{mg}, 0.106$ $\mathrm{mmol})$ and intermediate $117(46 \mathrm{mg}, 0.159 \mathrm{mmol})$ according to the procedure described for 119.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 8.20-8.15(\mathrm{~m}, 2 \mathrm{H}), 7.81(\mathrm{dd}, J=8.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.71(\mathrm{~s}$, $2 \mathrm{H}), 7.67(\mathrm{dd}, J=8.5,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.34(\mathrm{~s}, 4 \mathrm{H}), 7.19(\mathrm{dd}, J=8.7,0.9$ Hz, 1H), 4.60 (s, 2H), 4.55 ( $\mathrm{s}, 2 \mathrm{H}$ ), 4.35 ( $\mathrm{s}, 2 \mathrm{H}$ ), 3.99 ( $\mathrm{s}, 2 \mathrm{H}), 2.31$ (s, 6H), 1.47 (s, 9H).

LC/MS: $1.72 \mathrm{~min} m / z 683.2(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of tert-butyl 2-(4-(4-((3-(6-((2-(4-hydroxy-3,5-dimethylphenyl)-1H-benzo[d]imidazole)-6-

 sulfonamido)pyridin-3-yl)prop-2-yn-1-yl)oxy)butoxy)butoxy)acetate (122)

Compound $\mathbf{1 2 2}$ was obtained as brown solid ( $15 \mathrm{mg}, 20 \%$ ) from compound $\mathbf{1 1 4}$ ( $50 \mathrm{mg}, 0.106$ $\mathrm{mmol})$ and intermediate $118(46 \mathrm{mg}, 0.159 \mathrm{mmol})$ according to the procedure described for 119.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 8.16(\mathrm{~s}, 2 \mathrm{H}), 7.79(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.71(\mathrm{~s}, 2 \mathrm{H}), 7.61(\mathrm{~d}, J$ $=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.20-7.02(\mathrm{~m}, 2 \mathrm{H}), 4.29(\mathrm{~s}, 2 \mathrm{H}), 3.92(\mathrm{~s}, 2 \mathrm{H}), 3.57-3.51(\mathrm{~m}, 2 \mathrm{H}), 3.46(\mathrm{~d}, \mathrm{~J}=$ $5.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.43-3.36(\mathrm{~m}, 4 \mathrm{H}), 2.31(\mathrm{~s}, 6 \mathrm{H}), 1.68-1.54(\mathrm{~m}, 8 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H})$.

LC/MS: $1.72 \mathrm{~min} m / z 707.3(\mathrm{M}+\mathrm{H})^{+}$


Compound $\mathbf{1 2 3}$ was obtained as yellow solid ( $1 \mathrm{~g}, 23 \%$ ) from derivative $96(2.85 \mathrm{~g}, 12.05$ $\mathrm{mmol})$ and 2-amino-5-bromopyridine $\mathbf{9 7 i}(1.87 \mathrm{~g}, 10.85 \mathrm{mmol})$ according to the procedure described for 98a.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO-d6) $\delta 11.18(\mathrm{~s}, 1 \mathrm{H}), 8.51(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.28(\mathrm{~d}, \mathrm{~J}=2.5 \mathrm{~Hz}$, $1 \mathrm{H}), 8.00(\mathrm{~s}, 2 \mathrm{H}), 7.90(\mathrm{dd}, \mathrm{J}=8.8,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.79(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.09(\mathrm{~d}, \mathrm{~J}=9.0 \mathrm{~Hz}$, $1 \mathrm{H}), 6.99(\mathrm{~d}, \mathrm{~J}=8.8 \mathrm{~Hz}, 1 \mathrm{H})$. LC/MS: 1.5-1.6 min, $m / z 372.8(\mathrm{M}+\mathrm{H})^{+}, 374.8(\mathrm{M}+\mathrm{H}+2)^{+}$

## Synthesis of 3,4-diamino-N-(5-bromopyridin-2-yl)benzenesulfonamide (124)



Compound $\mathbf{1 2 4}$ was obtained as brown solid ( $514 \mathrm{mg}, 98 \%$ ) from derivative $\mathbf{1 2 3}$ ( $560 \mathrm{mg}, 1.5$ mmol ) according to the procedure described for $99 \mathbf{a}$.
${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO-d6) $\delta 10.75(\mathrm{~s}, 1 \mathrm{H}), 8.26(\mathrm{~d}, \mathrm{~J}=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.28-7.22(\mathrm{~m}, 1 \mathrm{H})$, $7.17(\mathrm{~d}, \mathrm{~J}=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.07-6.98(\mathrm{~m}, 2 \mathrm{H}), 6.97-6.91(\mathrm{~m}, 1 \mathrm{H}), 5.29(\mathrm{brs}, 2 \mathrm{H}), 4.84$ (brs, 2H). LC/MS: $1.2 \mathrm{~min}, m / z 343(\mathrm{M}+\mathrm{H})^{+}, 345(\mathrm{M}+\mathrm{H}+2)^{+}$

Synthesis of 2-(2-(2-(prop-2-yn-1-yloxy)ethoxy)ethoxy)ethan-1-ol (125)


To a suspension of potassium tert-butoxide ( $1.12 \mathrm{~g}, 9.99 \mathrm{mmol}$ ) in 50 mL of anhydrous THF, 70c ( $3 \mathrm{~g}, 19.97 \mathrm{mmol}$ ) was added and the resulting white suspension was stirred for 30 min . Then, propargyl bromide was added ( $1.11 \mathrm{~mL}, 9.99 \mathrm{mmol}$ ) and the resulting mixture was stirred
at room temperature for 18 h . Then, the insoluble salts were removed by filtration and the filtrated was concentrated in vacuo and purified on a silica gel flash-chromatography column (heptane-EtOAc 40:60 to 0:100) affording the title compound ( $1.5 \mathrm{~g}, 80 \%$ ).
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.21(\mathrm{~s}, 2 \mathrm{H}), 3.76-3.67(\mathrm{~m}, 10 \mathrm{H}), 3.65-3.57(\mathrm{~m}, 2 \mathrm{H}), 2.43(\mathrm{q}$, $J=2.1 \mathrm{~Hz}, 1 \mathrm{H})$.

## Synthesis of 6-(prop-2-yn-1-yloxy)hexan-1-ol (127)



To a cooled solution of $\mathbf{1 2 6}(3 \mathrm{~g}, 25.39 \mathrm{mmol})$ in 20 mL of dry DMF, NaH $60 \%$ in mixture ( 406 $\mathrm{mg}, 10.15 \mathrm{mmol}$ ) was added under nitrogen atmosphere and the resulting mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min . Then, propargyl bromide $(0.754 \mathrm{~mL}, 8.46 \mathrm{mmol})$ was added and the final suspension was continuously stirred for 14 h at room temperature. The crude was concentrated and the residual was taken up with saturated solution of $\mathrm{NH}_{4} \mathrm{Cl}(30 \mathrm{~mL})$ : the aqueous phase was extracted with EtOAc (3 x 20 mL ). The collected organic phases were washed with brine, dried over sodium sulfate, filtered and concentrated. The product $\mathbf{1 2 7}$ was obtained after purification filtration on a silica gel flash-chromatography column (heptane-EtOAc 70:30 to 0:100) to yield a colorless oil ( $1 \mathrm{~g}, 76 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.13(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.68-3.62(\mathrm{~m}, 2 \mathrm{H}), 3.52(\mathrm{t}, J=6.6$ $\mathrm{Hz}, 2 \mathrm{H}), 2.41(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.64-1.54(\mathrm{~m}, 4 \mathrm{H}), 1.44-1.35(\mathrm{~m}, 4 \mathrm{H})$.

## Synthesis of tert-butyl 2-((6-hydroxyhexyl)oxy)acetate (128)


$\mathrm{NaH} 60 \%$ in mixture ( $203 \mathrm{mg}, 5.08 \mathrm{mmol}$ ) was added under nitrogen atmosphere to a solution of $\mathbf{1 2 6}(1.5 \mathrm{~g}, 12.7 \mathrm{mmol})$ in 12 mL of dry DMF at $0^{\circ} \mathrm{C}$. The resulting mixture was stirred at 0 ${ }^{\circ} \mathrm{C}$ for 30 min . Then, tert-butyl 2-bromoacetate ( $0.624 \mathrm{~mL}, 4.23 \mathrm{mmol}$ ) was added and the final
suspension was continuously stirred for 14 h at room temperature. The crude was concentrated and the residual was taken up with saturated solution of $\mathrm{NH}_{4} \mathrm{Cl}(30 \mathrm{~mL})$ : the aqueous phase was extracted with EtOAc ( $3 \times 20 \mathrm{~mL}$ ). The collected organic phases were washed with brine, dried over sodium sulfate, filtered and concentrated. The product $\mathbf{1 2 8}$ was obtained after purification filtration on a silica gel flash-chromatography column (heptane-EtOAc 70:30 to 0:100) to yield a colorless oil (394 mg, 40\%).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.94(\mathrm{~s}, 2 \mathrm{H}), 3.68-3.61(\mathrm{~m}, 2 \mathrm{H}), 3.51(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.69$ $-1.54(\mathrm{~m}, 4 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H}), 1.43-1.38(\mathrm{~m}, 4 \mathrm{H})$.

## Synthesis of tert-butyl 2-((4-(bromomethyl)benzyl)oxy)acetate (130)



Compound 130 was obtained as colorless oil ( 420 mg , $67 \%$ ) from derivative (4(bromomethyl)phenyl)methanol $\mathbf{1 2 9}(400 \mathrm{mg}, 2 \mathrm{mmol})$ according to the procedure described for 116.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.46-7.27(\mathrm{~m}, 4 \mathrm{H}), 4.61(\mathrm{~s}, 2 \mathrm{H}), 4.49(\mathrm{~s}, 2 \mathrm{H}), 3.99(\mathrm{~s}, 2 \mathrm{H}), 1.49$ ( $\mathrm{s}, 9 \mathrm{H}$ ).

## Synthesis of 4-(4-(benzyloxy)butoxy)butan-1-ol (133)



To a cooled solution of 1,4-butanediol $132(1.04 \mathrm{~mL}, 12.34 \mathrm{mmol})$ in 6 mL of dry DMF NaH $60 \%$ in mixture ( $197 \mathrm{mg}, 4.93 \mathrm{mmol}$ ) was added. The resulting mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min under nitrogen atmosphere. Then, 4-benzyloxy-1-butanol 131 ( 0.781 mL g, 4.11 mmol )
was added and the final suspension was continuously stirred for 14 h at room temperature. The crude was concentrated under reduced pressure and the residual was taken up with saturated solution of $\mathrm{NH}_{4} \mathrm{Cl}(30 \mathrm{~mL})$ : the aqueous phase was extracted with EtOAc ( $3 \times 20 \mathrm{~mL}$ ). The collected organic phases were washed with brine, dried over sodium sulfate, filtered and concentrated. The product $\mathbf{1 3 3}$ was obtained after purification filtration on a silica gel flashchromatography column (heptane-EtOAc 90:10 to 50:50) to yield a yellow oil ( $870 \mathrm{mg}, 84 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.35-7.27(\mathrm{~m}, 5 \mathrm{H}), 4.50(\mathrm{~s}, 2 \mathrm{H}), 3.64(\mathrm{~d}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.55$ $-3.40(\mathrm{~m}, 6 \mathrm{H}), 1.72-1.64(\mathrm{~m}, 8 \mathrm{H})$.

Synthesis of tert-butyl 2-(4-(4-(benzyloxy)butoxy)butoxy)acetate (134)


Compound 134 was obtained as colorless oil (297 mg, 75\%) from 4-(4-(benzyloxy)butoxy)butan-1-ol $\mathbf{1 3 3}$ ( $250 \mathrm{mg}, 0.990 \mathrm{mmol}$ ) according to the procedure described for 116.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.36-7.28(\mathrm{~m}, 5 \mathrm{H}), 4.50(\mathrm{~s}, 2 \mathrm{H}), 3.94(\mathrm{~s}, 2 \mathrm{H}), 3.58-3.47(\mathrm{~m}$, $4 \mathrm{H}), 3.42-3.39(\mathrm{~m}, 4 \mathrm{H}), 1.72-1.62(\mathrm{~m}, 8 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H})$.

## Synthesis of tert-butyl 2-(4-(4-hydroxybutoxy)butoxy)acetate (135)



Compound 135 ( $430 \mathrm{mg}, 1.17 \mathrm{mmol}$ ) was solubilized in 3 mL of MeOH and $10 \% \mathrm{Pd} / \mathrm{C}$ was added. The mixture was stirred under a hydrogen atmosphere (balloon) for 18 h . Then, the mixture was filtered and the filtrate evaporated to give the title compound as colorless oil (310 $\mathrm{mg}, 96 \%)$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.94(\mathrm{~s}, 2 \mathrm{H}), 3.67-3.62(\mathrm{~m}, 2 \mathrm{H}), 3.55-3.51(\mathrm{~m}, 2 \mathrm{H}), 3.49-$ $3.43(\mathrm{~m}, 4 \mathrm{H}), 1.72-1.66(\mathrm{~m}, 8 \mathrm{H}), 1.47(\mathrm{~s}, 9 \mathrm{H})$.

Synthesis of tert-butyl 2-(4-(4-iodobutoxy)butoxy)acetate (136)


Iodine ( $465 \mathrm{mg}, 1.83 \mathrm{mmol}$ ) was added to triphenylphosiphine ( $480 \mathrm{mg}, 1.83 \mathrm{mmol}$ ) and imidazole ( $124 \mathrm{mg}, 1.83 \mathrm{mmol}$ ) in 14 mL of dry DCM at $0^{\circ} \mathrm{C}$. The resulting mixture was stirred at room temperature for 5 min , then was cooled to $0^{\circ} \mathrm{C}$. A solution of compound $\mathbf{1 3 5}(390 \mathrm{mg}$, 1.41 mmol ) in 9 mL of dichloromethane was added to the reaction mixture at $0^{\circ} \mathrm{C}$ and the resulting mixture was stirred at room temperature for 3 h . The reaction was quenched with saturated $\mathrm{Na}_{2} \mathrm{SO}_{3}$ solution ( 5 mL ): the aqueous phase was extracted with EtOAc (3 x 5 mL ). The combined organic phases were washed with brine, dried over sodium sulfate, filtered and evaporated to dryness. The crude material was purified by column chromatography (heptaneEtOAc 70:30 to 20:80) to afford the title compound as colorless oil ( $305 \mathrm{mg}, 56 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.94(\mathrm{~s}, 2 \mathrm{H}), 3.55-3.51(\mathrm{~m}, 2 \mathrm{H}), 3.42-3.38(\mathrm{~m}, 4 \mathrm{H}), 3.21(\mathrm{t}$, $J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.96-1.86(\mathrm{~m}, 2 \mathrm{H}), 1.70-1.63(\mathrm{~m}, 6 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H})$.

Synthesis of tert-butyl 2-(2-(2-(2-(4-formyl-2,6-dimethylphenoxy)ethoxy)ethoxy)ethoxy)acetate (137)


Compound $\mathbf{1 3 7}$ was obtained as a pale yellow solid ( $450 \mathrm{mg}, 80 \%$ ) from derivative 4-hydroxy-3,5-dimethylbenzaldehyde $\mathbf{7 6 a}(237 \mathrm{mg}, 1.58 \mathrm{mmol})$ and intermediate $146(600 \mathrm{mg}, 1.43$ $\mathrm{mmol})$ according to the procedure described for $\mathbf{6 1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.87(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{~s}, 2 \mathrm{H}), 4.02(\mathrm{~s}, 2 \mathrm{H}), 3.84(\mathrm{t}, J=5.7,3.8 \mathrm{~Hz}$, $2 \mathrm{H}), 3.77-3.66(\mathrm{~m}, 10 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}), 1.47(\mathrm{~s}, 9 \mathrm{H}) . \mathrm{LC} / \mathrm{MS}: 1.68 \mathrm{~min} \mathrm{~m} / \mathrm{z} 397.2(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of tert-butyl 2-((6-(4-formyl-2,6-dimethylphenoxy)hexyl)oxy)acetate (138)



Compound 138 was obtained as a pale yellow solid ( $260 \mathrm{mg}, 70 \%$ ) from derivative 4-hydroxy-3,5-dimethylbenzaldehyde 76a ( $170 \mathrm{mg}, 1.13 \mathrm{mmol}$ ) and intermediate $147(400 \mathrm{mg}, 1.03$ mmol ) according to the procedure described for $\mathbf{6 1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.87(\mathrm{~s}, 1 \mathrm{H}), 7.55(\mathrm{~s}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 2 \mathrm{H}), 3.81(\mathrm{t}, \mathrm{J}=6.6 \mathrm{~Hz}, 2 \mathrm{H})$, $3.53(\mathrm{t}, \mathrm{J}=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.33(\mathrm{~s}, 6 \mathrm{H}), 1.88-1.80(\mathrm{~m}, 2 \mathrm{H}), 1.72-1.64(\mathrm{~m}, 2 \mathrm{H}), 1.60-1.51(\mathrm{~m}$, $2 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H}), 1.35-1.21(\mathrm{~m}, 2 \mathrm{H})$. LC/MS: $1.5 \mathrm{~min} \mathrm{~m} / \mathrm{z} 365.2(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of tert-butyl 2-(4-((4-formyl-2,6-dimethylphenoxy)methyl)phenyl)acetate (139)



Compound 139 was obtained as a pale yellow solid ( 235 mg , $96 \%$ ) from derivative 4-hydroxy-3,5-dimethylbenzaldehyde 76a ( $105 \mathrm{mg}, 0.698 \mathrm{mmol}$ ) and intermediate $\mathbf{1 3 0}$ ( $200 \mathrm{mg}, 0.634$ mmol ) according to the procedure described for $\mathbf{6 1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.90(\mathrm{~s}, 1 \mathrm{H}), 7.58(\mathrm{~s}, 2 \mathrm{H}), 7.48-7.39(\mathrm{~m}, 4 \mathrm{H}), 4.87(\mathrm{~s}, 2 \mathrm{H}), 4.65$ (s, 2H), $4.01(\mathrm{~s}, 2 \mathrm{H}), 2.35(\mathrm{~s}, 6 \mathrm{H}), 1.50(\mathrm{~s}, 9 \mathrm{H}) . \operatorname{LC} / \mathrm{MS}: 1.4 \mathrm{~min} \mathrm{~m} / \mathrm{z} 385(\mathrm{M}+\mathrm{H})^{+}$


Compound $\mathbf{1 4 0}$ was obtained as a pale yellow solid ( $155 \mathrm{mg}, 96 \%$ ) from derivative 4-hydroxy-3,5-dimethylbenzaldehyde 76a ( $59 \mathrm{mg}, 0.394 \mathrm{mmol}$ ) and intermediate $148(154 \mathrm{mg}, 0.358$ mmol ) according to the procedure described for $\mathbf{6 1}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.87(\mathrm{~s}, 1 \mathrm{H}), 7.54(\mathrm{~s}, 2 \mathrm{H}), 3.93(\mathrm{~s}, 2 \mathrm{H}), 3.84(\mathrm{t}, \mathrm{J}=6.4 \mathrm{~Hz}$, $2 \mathrm{H}), 3.55-3.51(\mathrm{~m}, 2 \mathrm{H}), 3.50-3.44(\mathrm{~m}, 2 \mathrm{H}), 2.32(\mathrm{~s}, 6 \mathrm{H}), 1.93-1.85(\mathrm{~m}, 2 \mathrm{H}), 1.83-1.75$ (m, 2H), 1.69 - $1.63(\mathrm{~m}, 4 \mathrm{H}), 1.47(\mathrm{~s}, 9 \mathrm{H})$. LC/MS: $1.5 \mathrm{~min} \mathrm{~m} / \mathrm{z} 409(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of tert-butyl 2-(2-(2-(2-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy)ethoxy)ethoxy)ethoxy)acetate (141)



To a solution of compound $\mathbf{1 3 7}$ ( $240 \mathrm{mg}, 0.605 \mathrm{mmol}$ ) in 4 ml of dry DMF, derivative $\mathbf{6 4}$ (119 $\mathrm{mg}, 0.605 \mathrm{mmol})$ and $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{5}(152 \mathrm{mg}, 0.726 \mathrm{mmol})$ were added. The resulting mixture was heated at $80^{\circ} \mathrm{C}$ and stirred at this temperature for 18 h . After cooling at room temperature, water was added ( 15 mL ). The aqueous phase was extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ). The collected organic phases were washed with saturated solution of $\mathrm{NaHCO}_{3}(3 \times 10 \mathrm{~mL})$, brine, dried over sodium sulfate and filtered. After purification on silica gel flash-chromatography column (DCM-MeOH 97:3 to 80:20), the title compound was obtained as pale yellow solid ( $430 \mathrm{mg}, 68 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.46(\mathrm{~s}, 1 \mathrm{H}), 7.69(\mathrm{~s}, 2 \mathrm{H}), 6.83(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, \mathrm{~J}=$ $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.03(\mathrm{~s}, 3 \mathrm{H}), 4.02-3.99(\mathrm{~m}, 2 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 2 \mathrm{H}), 3.88-3.83(\mathrm{~m}, 2 \mathrm{H})$, $3.78-3.68(\mathrm{~m}, 8 \mathrm{H}), 2.38(\mathrm{~s}, 6 \mathrm{H}), 1.47(\mathrm{~s}, 9 \mathrm{H}) . \mathrm{LC} / \mathrm{MS}: 1.4 \mathrm{~min} \mathrm{~m} / \mathrm{z} 573(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of tert-butyl 2-((6-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl

 phenoxy)hexyl)oxy)acetate (142)

Compound $\mathbf{1 4 2}$ was obtained as a pale yellow solid ( 331 mg , 86\%) from derivative $\mathbf{1 3 8}$ (260 $\mathrm{mg}, 0.713 \mathrm{mmol})$ and intermediate $64(140 \mathrm{mg}, 0.713 \mathrm{mmol})$ according to the procedure described for 141.
${ }^{1} \mathrm{H}_{\mathrm{NMR}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.75(\mathrm{~s}, 1 \mathrm{H}), 7.72(\mathrm{~s}, 2 \mathrm{H}), 6.83(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{~d}, \mathrm{~J}=$ $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.97(\mathrm{~s}, 2 \mathrm{H}), 3.95(\mathrm{~s}, 2 \mathrm{H}), 3.93(\mathrm{~s}, 2 \mathrm{H}), 3.81(\mathrm{t}, \mathrm{J}=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.54(\mathrm{t}, \mathrm{J}=6.6$ $\mathrm{Hz}, 2 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H}), 1.90-1.80(\mathrm{~m}, 2 \mathrm{H}), 1.74-1.63(\mathrm{~m}, 2 \mathrm{H}), 1.63-1.51(\mathrm{~m}, 4 \mathrm{H}), 1.48(\mathrm{~s}$, 9H). LC/MS: $1.5 \mathrm{~min} m / z 541(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of tert-butyl 2-(4-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl phenoxy)methyl)phenyl)acetate (143)


Compound $\mathbf{1 4 3}$ was obtained as a pale yellow solid ( 290 mg , 85\%) from derivative 143 (229 $\mathrm{mg}, 0.647 \mathrm{mmol})$ and intermediate $\mathbf{6 4}(127 \mathrm{mg}, 0.647 \mathrm{mmol})$ according to the procedure described for 141.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.91(\mathrm{~s}, 1 \mathrm{H}), 7.76(\mathrm{~s}, 2 \mathrm{H}), 7.50-7.37(\mathrm{~m}, 5 \mathrm{H}), 6.84(\mathrm{~d}, \mathrm{~J}=2.3$ $\mathrm{Hz}, 2 \mathrm{H}), 6.46(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.86(\mathrm{~s}, 2 \mathrm{H}), 4.65(\mathrm{~s}, 2 \mathrm{H}), 4.01(\mathrm{~s}, 2 \mathrm{H}), 3.97(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}$, 3H), 2.37 (s, 6H), 1.49 ( $\mathrm{s}, 9 \mathrm{H})$. LC/MS: $1.6 \mathrm{~min} m / z 561(\mathrm{M}+\mathrm{H})^{+}$

## Synthesis of tert-butyl 2-(4-(4-(4-(5,7-dimethoxy-4-oxo-3,4-dihydroquinazolin-2-yl)-2,6-dimethyl

 phenoxy)butoxy)butoxy)acetate (144)

Compound $\mathbf{1 4 4}$ was obtained as a pale yellow solid (190 mg, 86\%) from derivative 140 (155 $\mathrm{mg}, 0.380 \mathrm{mmol})$ and intermediate $64(75 \mathrm{mg}, 0.380 \mathrm{mmol})$ according to the procedure described for 141.
${ }^{1} \mathrm{H}$ NMR (400 MHz, CDCl ${ }_{3}$ ) $\delta 9.34(\mathrm{~s}, 1 \mathrm{H}), 7.68(\mathrm{~s}, 2 \mathrm{H}), 6.83(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.46(\mathrm{~d}, \mathrm{~J}=$ $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.98(\mathrm{~s}, 3 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 3.93(\mathrm{~s}, 2 \mathrm{H}), 3.84(\mathrm{t}, \mathrm{J}=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.56-3.44(\mathrm{~m}$, $4 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H}), 1.94-1.87(\mathrm{~m}, 2 \mathrm{H}), 1.84-1.78(\mathrm{~m}, 2 \mathrm{H}), 1.68(\mathrm{qt}, \mathrm{J}=4.8,2.3 \mathrm{~Hz}, 6 \mathrm{H}), 1.48$ (s, 9H). LC/MS: $1.6 \mathrm{~min} m / z 585(\mathrm{M}+\mathrm{H})^{+}$

Synthesis of tert-butyl 2-(2-(2-(2-hydroxyethoxy)ethoxy)ethoxy)acetate (145)


Compound $\mathbf{1 4 5}$ was obtained as colorless oil ( $1.74 \mathrm{~g}, 60 \%$ ) from compound 70c ( $5 \mathrm{~g}, 33.3$ $\mathrm{mmol})$ according to the procedure described for $\mathbf{1 2 8}$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.02(\mathrm{~s}, 2 \mathrm{H}), 3.76-3.65(\mathrm{~m}, 10 \mathrm{H}), 3.61(\mathrm{dd}, J=5.3,3.9 \mathrm{~Hz}$, $2 \mathrm{H}), 1.47$ (s, 9H).

## Synthesis of tert-butyl 2-(2-(2-(2-(tosyloxy)ethoxy)ethoxy)ethoxy)acetate (146)



Compound 146 was obtained as colorless oil ( 785 mg , 64\%) from compound 145 ( $800 \mathrm{mg}, 3.02$ mmol ) according to the procedure described for 74a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.80(\mathrm{~d}, J=8.2,2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.16(\mathrm{t}, J=4.9$ $\mathrm{Hz}, 2 \mathrm{H}), 4.00(\mathrm{~s}, 2 \mathrm{H}), 3.73-3.64(\mathrm{~m}, 10 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 9 \mathrm{H})$.

## Synthesis of tert-butyl 2-((6-(tosyloxy)hexyl)oxy)acetate (147)



Compound $\mathbf{1 4 7}$ was obtained as colorless oil ( $206 \mathrm{mg}, 62 \%$ ) from compound $\mathbf{1 2 8}$ ( 200 mg , 0.861 mmol ) according to the procedure described for 74a.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.79(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.03(\mathrm{t}, J=$ $6.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.92(\mathrm{~s}, 2 \mathrm{H}), 3.47(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}), 1.70-1.61(\mathrm{~m}, 2 \mathrm{H}), 1.60-$ $1.50(\mathrm{~m}, 2 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H}), 1.39-1.28(\mathrm{~m}, 4 \mathrm{H})$.

## Synthesis of tert-butyl 2-(4-(4-(tosyloxy)butoxy)butoxy)acetate (148)



Compound 148 was obtained as colorless oil ( $160 \mathrm{mg}, 60 \%$ ) from compound $\mathbf{1 3 5}$ ( 182 mg , 0.658 mmol ) according to the procedure described for 74a.
${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.79(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.06(\mathrm{t}, J=$ $6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.94(\mathrm{~s}, 2 \mathrm{H}), 3.51(\mathrm{t}, J=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.43-3.27(\mathrm{~m}, 4 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}), 1.80-$ $1.52(\mathrm{~m}, 8 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H})$.

### 7.3 Biological protocols

### 7.3.1 FRET assay

Compounds were tested on BD1 and BD2 domains (6xHis-tagged) of each BET protein in a dose-response format measuring binding competition between the compounds and an AlexaFluor647 derivative of JQ1. Compounds were diluted in assay buffer ( 150 mM HEPES, $150 \mathrm{mM} \mathrm{NaCl}, 5 \%$ glycerol, 1 mM DTT , and 1 mM CHAPS, pH 7.4 ), starting from a stock solution of $10 \mathrm{mM}(100 \% \mathrm{DMSO})$. The highest concentration tested was $100 \mu \mathrm{M}$ and from this concentration 12 threefold dilutions were prepared. $5 \mu \mathrm{~L}$ of each concentration were transferred into a low volume black 384-well plate and, thanking advance of a Thermo Scientific Multidrop Combi, $2 \mu \mathrm{~L}$ of protein $(10 \mathrm{nM}), 2 \mu \mathrm{~L}$ of AlexaFluor647 derivative of JQ1 ( 50 nM ) and $1 \mu \mathrm{~L}$ of europium chelate-labeled anti-6His antibody ( 1 nM ) were transferred in each well. After an equilibration of 30 min in the dark at rt , the binding of the protein to the fluorescent ligand was detected on BMG Labtech Pherastar luminescence plate reader (excitation $=337 \mathrm{~nm}$; emission $1=615 \mathrm{~nm}$; emission $2=665 \mathrm{~nm}$; dual wavelength bias dichroic $=400 \mathrm{~nm}, 630 \mathrm{~nm}$ ). TR-FRET ratio was calculated using the following equation:
ratio $=(($ acceptor fluorescence at 665 nm$) /($ donor fluorescence at 615 nm$)) \times 1000$
Each compound was tested in triplicate and data were analyzed using Graphpad Prism software.

### 7.3.2 Cell culture, testing compounds and immunoblotting

HeLa cells were cultured ( $37^{\circ} \mathrm{C}, 5 \% \mathrm{CO}_{2}$ ) in DMEM medium (Gibco) supplemented with $10 \%$ fetal bovine serum (FBS) (Gibco), L-glutamine (Gibco), penicillin, streptomycin. The cells
were seeded at $1 \times 10^{6}$ per well on a standard 6 -well plate. After 24 h , cells were treated with compounds at different concentrations (DMSO concentration of $0.1 \%$ ). After the proper incubation time ( 6 h and 16 h ) cells were washed with PBS twice and lysed with RIPA buffer (Sigma), supplemented with protease inhibitor cocktail (Roche), Benzonase (Merck), and 0.5 mM MgCl 2 . Lysate was briefly sonicated and centrifuged at 20000 g for 10 min at $4^{\circ} \mathrm{C}$. From each sample, supernatant was collected and protein concentration measured by BCA assay. Regarding the classic Western-blot procedure, proteins were resolved by SDS-PAGE on NuPage 4-12\% Bis-Tris Midi Gel (Invitrogen) and transferred to Amersham Protran 0.45 NC nitrocellulose membrane (GE Healthcare) using wet transfer. The membrane was blocked with $5 \% \mathrm{w} / \mathrm{v}$ milk in Tris-buffered saline (TBS) with $0.1 \% \mathrm{w} / \mathrm{v}$ Tween-20. Blots were probed with anti-Brd4 (AbCam, ab128874), anti-Brd3 (AbCam, ab50818), anti-Brd2 (AbCam, ab139690), and hFAB Rhodamine anti-tubulin IgG (BioRAD \#12004166). Blots were developed with IRDye secondary antibody (Licor) anti-mouse or anti-rabbit and bands visualized using ChemiDoc imaging system (Bio-Rad). ImageLab software was used for densitometric analysis. Regarding the use of the instrument Simple Western ${ }^{\text {TM }}$ Jess, samples were prepared and analyzed following the standard protocol reported.

### 7.3.2 Cell viability assay

MV4-11 cells were grown in RPMI (Gibco) supplemented with $10 \%$ fetal bovine serum (FBS) and L-glutamine. MV4-11 cells were seeded into a clear-bottomed 384-well plate (Sigma Greiner CELLSTAR) at $\sim 0.6 \times 10^{5}$ cells $/ \mathrm{mL}$ in $25 \mu \mathrm{~L}$ of media. Under sterile conditions, each test compound (stock solution $20 \mathrm{mM}, 100 \%$ DMSO) was diluted in RPMI media to reach $0.1 \%$ DMSO. Each compound was tested at 12 concentrations (from $20 \mu \mathrm{M}$ to 2 nM ), and each condition was performed in triplicate. For the no-cell control $25 \mu \mathrm{~L}$ of plain RPMI was dispensed into $\sim 2$ rows of the plate. The assay plate was incubated at $37^{\circ} \mathrm{C} 5 \% \mathrm{CO}_{2}$ for 48 h . After the treatment, cell viability was measured with Promega CellTiter-Glo luminescent cell
viability assay kit according to the manufacturer instructions. Signal was recorded on a BMG Labtech Pherastar luminescence plate reader with recommended settings. Data were analyzed with Graphpad Prism software.

### 7.4 NMR Data







































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